

# INTERNATIONAL STANDARD

Fuel cell technologies –  
Part 7-2: Test methods – Single cell and stack performance tests for solid oxide  
fuel cells (SOFCs)

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**Fuel cell technologies –  
Part 7-2: Test methods – Single cell and stack performance tests for solid oxide  
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INTERNATIONAL  
ELECTROTECHNICAL  
COMMISSION

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## INTERNATIONAL ELECTROTECHNICAL COMMISSION

## FUEL CELL TECHNOLOGIES –

**Part 7-2: Test methods – Single cell and stack performance tests  
for solid oxide fuel cells (SOFCs)**

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IEC 62282-7-2 has been prepared by IEC technical committee 105: Fuel cell technologies. It is an International Standard.

This second edition cancels and replaces the first edition published in 2021. This edition constitutes a technical revision.

This edition includes the following significant technical changes with respect to the previous edition:

- a) Table 1 has been revised to specify the units missing for some terms;
- b) bibliographical entries (ISO/TR 15916, SOCTESQA test modules and ISO/IEC Guide 98-6:2021) have been added to provide further information.

The text of this International Standard is based on the following documents:

Draft	Report on voting
105/1093/FDIS	105/1099/RVD

Full information on the voting for its approval can be found in the report on voting indicated in the above table.

The language used for the development of this International Standard is English.

This document was drafted in accordance with ISO/IEC Directives, Part 2, and developed in accordance with ISO/IEC Directives, Part 1 and ISO/IEC Directives, IEC Supplement, available at [www.iec.ch/members\\_experts/refdocs](http://www.iec.ch/members_experts/refdocs). The main document types developed by IEC are described in greater detail at [www.iec.ch/publications](http://www.iec.ch/publications).

A list of all parts in the IEC 62282 series, published under the general title *Fuel cell technologies*, can be found on the IEC website.

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- reconfirmed,
- withdrawn, or
- revised.

## INTRODUCTION

~~This part of IEC 62282 specifies test methods for a single cell and stack (denoted as "cell/stack" hereafter) that is required in power generation systems using~~ Solid oxide fuel cells (SOFCs) have a broad range of geometry and size. As such, in general, peripherals like current collectors and gas manifolds are unique to each cell or stack and are often incorporated into a cell or stack to form one integrated unit. In addition, they tend to have a significant effect on the power generation characteristics of the cell or stack. This document therefore introduces as its subject "cell/stack assembly units", which are defined as those units containing not only a cell or stack but also peripherals.

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## FUEL CELL TECHNOLOGIES –

### Part 7-2: Test methods – Single cell and stack performance tests for solid oxide fuel cells (SOFCs)

#### 1 Scope

This part of IEC 62282 applies to SOFC cell/stack assembly units, testing systems, instruments and measuring methods, and specifies test methods to test the performance of SOFC cells and stacks.

This document is not applicable to small button cells that are designed for SOFC material testing and provide no practical means of fuel utilization measurement.

This document is used based on the recommendation of the entity that provides the cell performance specification or for acquiring data on a cell or stack in order to estimate the performance of a system based on it. Users of this document can selectively execute test items suitable for their purposes from those described in this document.

Users can substitute selected test methods of this document with equivalent test methods of IEC 62282-8-101 for solid oxide cell (SOC) operation for energy storage purposes, operated in reverse or reversible mode.

#### 2 Normative references

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

IEC 60050-485, *International Electrotechnical Vocabulary (IEV) – Part 485: Fuel cell technologies*, available at <https://www.electropedia.org>

IEC 60584-1, *Thermocouples – Part 1: EMF specifications and tolerances*

IEC 60584-3, *Thermocouples – Part 3: Extension and compensating cables – Tolerances and identification system*

IEC 61515, *Mineral insulated metal-sheathed thermocouple cables and thermocouples*

ISO 5168, *Measurement of fluid flow – Procedures for the evaluation of uncertainties*

~~ISO 6141, Gas analysis – Contents of certificates for calibration gas mixtures~~

~~ISO 6142-1, Gas analysis – Preparation of calibration gas mixtures – Gravimetric method for Class I mixtures~~

~~ISO 6143, Gas analysis – Comparison methods for determining and checking the composition of calibration gas mixtures~~

~~ISO 6145-7, Gas analysis – Preparation of calibration gas mixtures using dynamic methods – Part 7: Thermal mass-flow controllers~~

ISO 6974 (all parts), *Natural gas – Determination of composition with defined uncertainty by gas chromatography*

ISO 7066-2, *Assessment of uncertainty in the calibration and use of flow measurement devices – Part 2: Non-linear calibration relationships*

ISO 8573-1, *Compressed air – Part 1: Contaminants and purity classes*

ISO 8756, *Air quality – Handling of temperature, pressure and humidity data*

ISO 12185, *Crude petroleum, petroleum products and related products – Determination of density – Laboratory density meter with an oscillating U-tube ~~method~~ sensor*

### 3 Terms, definitions and symbols

#### 3.1 Terms and definitions

For the purposes of this document, the terms and definitions given in IEC 60050-485 and the following apply.

ISO and IEC maintain terminology databases for use in standardization at the following addresses:

- IEC Electropedia: available at <https://www.electropedia.org/>
- ISO Online browsing platform: available at <https://www.iso.org/obp>

##### 3.1.1

##### **cell/stack assembly unit**

unit including a single cell or stack, as well as gas supply parts, current collector parts, and any other peripherals ~~as required for~~ used in power generation tests

##### 3.1.2

##### **active electrode area**

geometric electrode area upon which an electrochemical reaction occurs

Note 1 to entry: Usually the active electrode area is the smaller of the anode and cathode areas.

##### 3.1.3

##### **current density**

current divided by the active electrode area

##### 3.1.4

##### **average repeating unit voltage**

cell/stack assembly unit voltage divided by the number of the cells in a series connection in the unit

##### ~~3.1.5~~

##### ~~**standard temperature and pressure**~~

##### ~~**STP**~~

~~temperature of 0 °C and an absolute pressure of 101,325 kPa, respectively~~

### 3.1.5 anode gas

gas that is supplied to the inlet of the anode of a single cell/stack assembly unit

Note 1 to entry: Such a gas belongs to one of the following categories:

- a) pure hydrogen or mixture that contains hydrogen as a principal component with water vapour or nitrogen;
- b) reformed gas of raw fuel of SOFC such as methane or kerosene premixed with water vapour or air as oxidant;
- c) simulated gas of reformat that contains hydrogen, water vapour, carbon monoxide, carbon dioxide, methane, nitrogen, etc., as main components;
- d) methane, alcohols and other raw fuels directly supplied in pure form or mixed with water vapour or air, or both.
- e) condensable gas operating in gas phase such as anhydrous ammonia ( $\text{NH}_3$ ) as raw input fuel or in cracked form.

### 3.1.6 cathode gas

gas that is supplied to the inlet of the cathode of a single cell/stack assembly unit

Note 1 to entry: Oxygen and nitrogen are its main components.

### 3.1.7 current collector

conductive material in a ~~fuel~~ cell/stack assembly unit that collects electrons from the anode side or conducts electrons to the cathode side

### 3.1.8 stable state

condition of a cell/stack assembly unit at which the unit is stable enough for any controlling parameter and the output voltage or output current of the unit to remain within its tolerance range of variation

### 3.1.9 theoretical current

current when the supplied anode gas or cathode gas is completely consumed in electrochemical reactions divided by the number of cells in a series connection

### 3.1.10 effective fuel utilization

ratio of the actual output current of the cell/stack assembly unit to the theoretical current that is calculated for the supplied fuel

Note 1 to entry: The effective utilization is the utilization of reactants in the electrochemical reaction at the anode due to the actual current. This ~~may~~ can be less than the actual or total utilization if there are gas inlet and cross leaks.

Note 2 to entry: Causes of less-than-optimal currents include losses due to electronic conduction within the cell/stack assembly, gas leaks ~~and anode gas pass through~~.

Note 3 to entry: A calculation method of effective fuel utilization is given in Annex B.

### 3.1.11 effective oxygen utilization

ratio of the actual output current of the cell/stack assembly unit to the theoretical current that is calculated for the supplied oxygen

Note 1 to entry: The effective utilization is the utilization of reactants in the electrochemical reaction at the cathode due to the actual current. This ~~may~~ can be less than the actual or total utilization if there are gas inlet and cross leaks.

Note 2 to entry: A calculation method of effective oxygen utilization is given in Annex C.

**3.1.12****maximum effective fuel utilization**

highest effective fuel utilization that the cell/stack assembly unit can operate at, without causing unacceptable degradation

Note 1 to entry: The acceptable degradation rate is usually obtained from the developer.

**3.1.13****minimum cell/stack assembly unit voltage**

lowest cell/stack assembly unit voltage specified by the manufacturer

**3.1.14****open circuit voltage****OCV**

voltage across the terminals of a fuel cell/stack assembly unit with cathode and anode gases present and in the absence of external current flow

Note 1 to entry: Also known as "no-load voltage".

**3.1.16****power density**

~~ratio of the power to the active electrode area of a cell/stack assembly unit~~

~~Note 1 to entry: Power density is calculated from the voltage multiplied by the current density ( $P_d = V \times J$ , where  $J$  is current density).~~

**3.1.15****total impedance**

frequency-dependent losses due to ohmic, activation, diffusion, concentration effects, stray (parasitic) capacitance and inductances

**3.1.16****total resistance**

real part of the low-frequency limit of total impedance

**3.1.17****stoichiometric ratio**

ratio between the number of moles of reactant gas flowing per unit time to that needed used by the electrochemical reaction

Note 1 to entry: The terms, "stoichiometric ratio" and "reactant gas utilization," are related. The reciprocal of the fraction of the gas utilized is the stoichiometric ratio.

**3.2 Symbols**

Table 1 lists the symbols and units that are used in this document.

**Table 1 – Symbols**

Symbol	Term	Unit
$a$	Error limit specified from specification of instrument	a
$I$	Current	A
$J$	Current density	A/cm <sup>2</sup>
$A$	Active electrode area	cm <sup>2</sup>
$Z$	Total impedance	$\Omega$ cm <sup>2</sup>
$n$	Number of transferred electrons	
$N$	Number of cells in a series connection in the cell/stack assembly unit	
$p_a$	Absolute pressure of anode gas	kPa
$p_c$	Absolute pressure of cathode gas	kPa
$P$	Output power	W
$P_d$	Output power density	W/cm <sup>2</sup>
$q_a$	Flow rate of anode gas	l/min (STP <sup>b</sup> )
$q_c$	Flow rate of cathode gas	l/min (STP)
$q_j$	Flow rate of fuel component $j$ in anode gas	l/min (STP)
$t$	Time	s, min, h
$T_{op}$	Cell/stack assembly unit operating temperature	°C or K
$\#_i u_c$	Combined standard uncertainty for instruments	a
$u_{1,i}$	Standard uncertainty for instrument $i$	a
$U_f$	Effective fuel utilization	%
$U_{O_2}$	Effective oxygen utilization	%
$U_1$	<del>Extended instrument</del> Instrument expanded uncertainty	a
$V$	Voltage	V
$x_i$	Molar fraction of component $i$ or mole percent of component $i$	mol/mol or mol % <sup>bc</sup>
$c_i$	Concentration of component $i$	mol/m <sup>3</sup>
$\xi_j$	Hydrocarbon conversion rate for hydrocarbon component $j$	%
<sup>a</sup> Denotes where the unit varies depending on the specification. <sup>b</sup> Abbreviation for standard temperature and pressure <sup>bc</sup> Mole percent expressed as one hundred times mole fraction.		

#### 4 General safety conditions

An operating fuel cell uses oxidizing and combustible gases. Typically, these gases are stored in high-pressure containers. In some cases, the fuel can be a toxic condensable gas (such as ammonia). The fuel cell itself ~~may~~ can be operated at pressures greater than atmospheric pressure. Leaks or outlet flows from cell/stack assembly unit can contain toxic elements (e.g. when using ammonia as a fuel). Those who carry out cell/stack assembly unit testing shall be trained and experienced in the operation of test systems and specifically in safety procedures involving electrical equipment and reactive, compressed gases, and toxic compounds if applicable (e.g. when using ammonia as a fuel).

~~The test personnel are responsible for obtaining and following all applicable safety codes and generally accepted engineering practices related to their test system, facility, fuels (with particular attention to compressed gases), and exhaust products.~~

Materials which are compatible with the use and storage of the reactant gases shall be used during testing. ~~Local safety codes and standards for working with hydrogen, hydrocarbons and carbon monoxide should be followed.~~

In summary, safely operating a test station requires appropriate technical training and experience as well as safety facilities and equipment, all of which are outside the scope of this document.

## 5 Cell/stack assembly unit

A cell/stack assembly unit includes a cell or stack, gas supply, current leads, and such other peripherals as required for power generation ~~tests~~. It shall be provided with single or multiple measuring points for temperature and voltage, and one set of current lead points, all to be specified by the manufacturer.

As shown in Annex A, the boundary of a cell/~~stack~~ assembly unit goes through the anode gas supply port, cathode gas supply port, temperature, ~~pressure~~ measuring point, current lead points, voltage measuring points and mechanical load application points.

Some cell/stack assembly units ~~may~~ can have no exhaust port for the anode gas or cathode gas because of the configuration of the cells. In such cases, the gas flow field pattern and its material shall be determined by the method recommended by the manufacturer. The load application method shall be also based on the recommendation of the manufacturer. The maximum operating temperature ~~from~~ recommended by the manufacturer shall not be exceeded.

If the components of a cell/stack assembly unit other than a cell/stack are not specified by the manufacturer, the following shall be described in the test report, as a minimum:

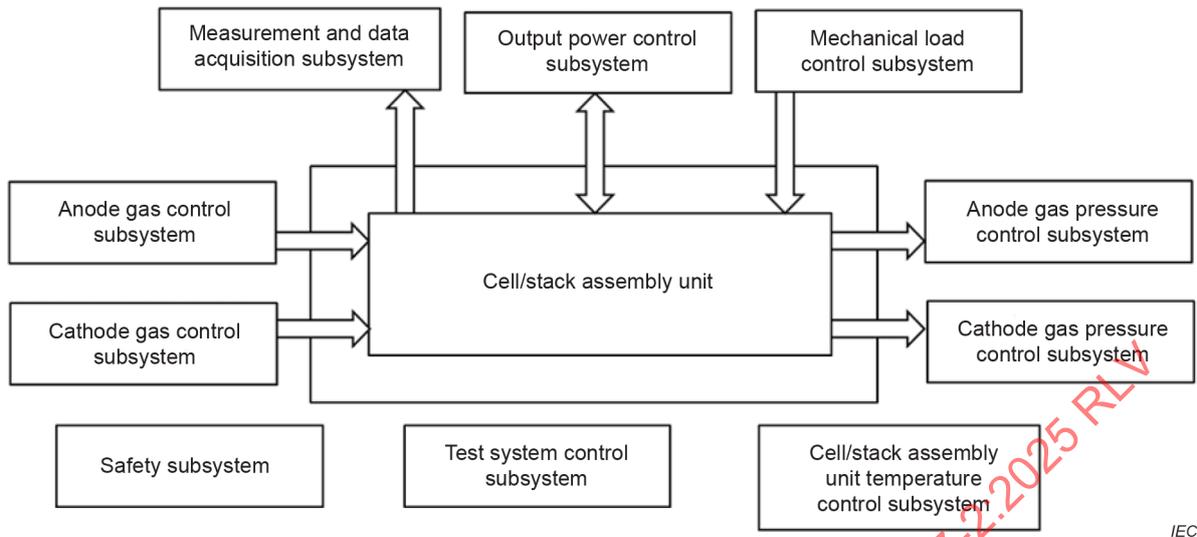
- a) materials and geometry of the peripheral components to be used for testing;
- b) flow patterns and directions of anode and cathode gases;
- c) locations of temperature measurement, mechanical load application, voltage measurement and current leads;
- d) magnitude of the mechanical load;
- e) configuration of assembly unit and its assembling method.

## 6 Testing system

### 6.1 Subsystems in testing system

#### 6.1.1 General

As shown in Figure 1, a testing system consists of an anode gas control subsystem, cathode gas control subsystem, cell/stack assembly unit temperature control subsystem, output power control subsystem, measurement and data acquisition subsystem and safety subsystem. It ~~may~~ can also include a mechanical load control subsystem, anode gas and cathode gas pressure control subsystem or a test system control subsystem that controls the whole testing system, or both, if necessary.



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Figure 1 – Testing system

**6.1.2 Anode gas control subsystem**

The anode gas control subsystem controls the flow rate, composition and temperature of the anode gas supplied to the cell/stack assembly unit. If the gas composition is to be maintained throughout the piping, then ~~attention shall be paid to~~ the materials, temperature, inner diameter and length of the piping shall be selected such as to ensure that any changes the gas composition can have within the piping are insignificant. Where necessary, the piping shall be heated or thermally insulated, or both in order to prevent condensation of water vapour.

Care should be taken to avoid other phenomena, such as carbon deposits, and the evaporation and transport of undesired materials in the gas streams, such as chromium species.

**6.1.3 Cathode gas control subsystem**

The cathode gas control subsystem controls the flow rate, composition and temperature of the cathode gas supplied to the cell/stack assembly unit.

**6.1.4 Cell/stack assembly unit temperature control subsystem**

The cell/stack assembly unit temperature control subsystem controls, at least, the electric furnace or the unit temperature. It maintains the operating temperature. The electric furnace shall be selected to maintain the temperature distribution within the specified tolerance level. Efforts should be made to minimize the electrical noise that the electric furnace generates while providing heat. It is assumed that all the test systems will use an electrical furnace for simplicity and safety reasons.

**6.1.5 Output power control subsystem**

The output power control subsystem controls the output current or output voltage of the cell/stack assembly unit.

### 6.1.6 Measurement and data acquisition subsystem

The measurement and data acquisition subsystem acquires and records the cell/stack assembly unit temperature, current, voltage, anode gas flow rate, cathode gas flow rate, and optionally, environmental conditions (ambient temperature, relative humidity, and atmospheric pressure) in accordance with the specified method. If necessary, it also acquires and records the mechanical load applied to the cell; the temperature, composition and pressure of the cathode gas and the anode gas; the flow rate, composition, temperature and pressure of anode and cathode exhaust gases; and cell/stack assembly unit impedance data, etc., in accordance with the specified method.

### 6.1.7 Safety subsystem

The safety subsystem functions as a detector and alarm system for malfunctioning of the test system based on predefined parameters and criteria. If it detects a serious fault, then it shall automatically establish a safe state in the test system. The anode should be purged with an inert gas, such as nitrogen, which ~~could~~ can also contain hydrogen at concentrations below the lower flammability limit.

### 6.1.8 Mechanical load control subsystem

The optional mechanical load control subsystem regulates the mechanical load that is applied to increase the contact among components in the cell/stack assembly unit. The subsystem should be strong enough to apply the required mechanical load under the test conditions and to maintain the load for long term operation.

### 6.1.9 Gas pressure control subsystem for anode and cathode

The optional gas pressure control subsystem for anode and cathode gases regulates the pressure of these gases by the use of a back pressure control valve, etc.

### 6.1.10 Test system control subsystem

The test system control subsystem provides the integrated control for each control subsystem and data acquisition subsystem.

## 6.2 Maximum variation in control items of testing system

The tolerable variation of each control item in the testing system shall fall within the following ranges:

in the case of current control:	current: $\pm 1$ % relative to rated value point;
in the case of voltage control:	voltage: $\pm 1$ % relative to set point;
temperature:	$\pm 1,0$ % relative to set point;

NOTE 1 Temperature variation at the set point of less than  $\pm 5$  K will increase reproducibility.

anode and cathode gas flow rates:	$\pm 1$ % relative to rated;
anode gas composition:	$\pm 2,0$ mol % for $H_2$ , $N_2$ ;
	$\pm 2,0$ mol % for $CO$ , $CO_2$ , $CH_4$ , $NH_3$ ;
	$\pm 5,0$ mol % for $H_2O$ (water vapour concentration);

in case of bubbler or sparger humidification: dew point temperature:  $\pm 1$  °C;

NOTE 2 At water vapour concentrations greater than 10 mol %, a bubbler system (sparger) can cause higher uncertainty

cathode gas composition:  $\pm 1,0$  mol % of the target O<sub>2</sub> concentration;

where pressures of anode and cathode gases are to be controlled, pressures of anode and cathode gases:  $\pm 1$  % of rated condition, when pressure of rated condition is equal to or larger than 0,3 MPa; and 3 kPa, when pressure of rated condition is smaller than 0,3 MPa.

## 7 Instruments and measurement methods

### 7.1 General

Measuring instruments shall meet the requirement of 7.2. As a minimum, the flow rate and composition of the anode and cathode gases as well as the temperature, voltage, and current of the cell/stack assembly unit shall be measured. Additional measurements shall be taken based on the test parameters or test conditions, or both. It is possible that some of the following items specified in 7.3 or 7.4 may will not be measurable in the case of a cell/stack assembly unit having no anode or cathode gas exhaust port.

### 7.2 Instrument uncertainty

The expanded uncertainty of each measuring instrument (coverage factor  $k = 2$ ) at the time of calibration or that estimated from the class of instrument shall meet the following requirements:

NOTE Coverage factor is defined in ISO/IEC Guide 98-3.

current:	$\pm 1$ % relative to rated;
voltage:	$\pm 0,5$ % relative to the open circuit voltage (OCV);
temperature:	$\pm 1,0$ % of reading;
flow rates of anode and cathode gases:	$\pm 2$ % of rated;
pressures of anode and cathode gases:	$\pm 1$ % of reading; average
anode gas composition:	$\pm 2$ mol % for H <sub>2</sub> , H <sub>2</sub> O, and N <sub>2</sub> ; $\pm 1$ mol % for CO, CO <sub>2</sub> , and CH <sub>4</sub> , NH <sub>3</sub> ;
cathode gas composition:	$\pm 0,3$ mol % for O <sub>2</sub> (balance N <sub>2</sub> ).

A method for determining instrument expanded uncertainty is given in Annex G.

### 7.3 Anode gas

#### 7.3.1 Anode gas flow rate

The anode gas flow rate shall be measured using mass flow meters, volumetric flow meters or turbine-type flow meters. The flow meter shall be selected by taking into consideration the species in the supplied gas, the range of flow rates, and the allowable uncertainty of the flow meter. When measurements are made on a volumetric basis, they shall be converted to mass flow rate by measuring the gas temperature and pressure or gas density in the vicinity of the flow meters. Measurement uncertainty for dry gases should be evaluated in accordance with ISO 5168 or ISO 7066-2.

#### 7.3.2 Anode gas composition

The anode gas composition should be measured when the performance of the cell/stack assembly unit is measured. If this is not possible, however, the anode gas composition shall be measured during the preparation of the performance test under the same conditions as those of the cell performance test. See 10.6.2.2.

When anode gas is supplied in one of the following conditions a) to d) below, and if the gas supply line has no reactors, such as a reformer, and is confirmed to insignificantly change the gas composition, composition ~~may~~ can be calculated based on the composition table published by the ~~gas~~ fuel supplier and values obtained from each flow meter, in accordance with ISO 6145-7:

- a) a single-composition gas such as hydrogen is supplied;
- b) a mixed gas of known composition is supplied;
- c) anode gas is supplied by mixing component gases in a controlled manner using multiple flow meters;
- d) gases under b) and c) above are supplied in combination.

The anode gas shall be sampled near the anode gas supply port of the cell/stack assembly unit and analysed using an infrared ~~spectroscopy~~ spectrometer, mass spectrometer, gas chromatograph or similar device. The gas sample shall be transported from its origin to the point of analysis in a manner which minimizes changes in composition. Thus, the material, temperature, diameter and the length of the tubing shall be carefully chosen to minimize the compositional change in the sampling tubing. When necessary, it shall be heated to avoid the condensation of the water vapour.

If water vapour is likely to affect measurement, remove water from the gas sample or dilute the gas sample with argon or a similar inert gas.

The result of such analysis for gas component  $i$ , expressed as  $c_i$  (mol/m<sup>3</sup>) shall be normalized to obtain a normalized concentration,  $x_i$  (mol/mol), using the following equation:

$$x_i = \frac{c_i}{\sum_i c_i} \quad (1)$$

where  $\sum_i c_i$  represents the sum of concentrations of all component gases in the analysis.

The gas analyser shall be calibrated using a standard gas of known mass ratio.

The measurement uncertainty shall be evaluated in accordance with the ISO 6974 series, ~~ISO 6141, ISO 6142-1, or ISO 6143.~~

### 7.3.3 Anode gas temperature

The gas temperature shall be measured near the anode gas supply port of the cell/stack assembly unit by using a thermocouple ~~or sheathed thermocouple and an extension leadwire of a type and class in accordance with IEC 60584-1, IEC 60584-3 or IEC 61515~~ of a type and class in accordance with IEC 60584-1 or sheathed thermocouple of a type and class in accordance with IEC 61515 and an extension lead wire of a type and class in accordance with IEC 60584-3. When there is a reactor such as a reformer, the gas temperature at the exit of the reactor should also be measured.

NOTE There can be significant differences between the temperature of the tube wall and the temperature of the bulk gas.

If it is difficult to measure the gas temperature during the cell performance test, the anode gas temperature shall be measured during the preparation of the performance test under the same conditions as those of the performance test.

### 7.3.4 Anode gas pressure

The anode gas pressure shall be measured upstream of the anode gas supply port of the cell/stack assembly unit by using a calibrated pressure sensor, manometer, Bourdon tube or similar instrument. The measuring instrument shall be located in such a manner that the uncertainty is minimized in consideration of any pressure loss within the piping, piping temperature and other factors. Condensation of water vapour during measurement shall be prevented. One way ~~may~~ can be to measure the pressure by injecting a very small amount of dry nitrogen gas or similar into the pipe, close to the measuring instrument.

### 7.3.5 Anode exhaust gas flow rate

The anode exhaust gas flow rate shall be measured using mass flow meters, volumetric flow meters or turbine-type flow meters after implementing a means to prevent water condensation from affecting the stability of anode gas flow or after removing water from the gas flow. When measurements are made on a volumetric basis, they shall be converted to mass flow rate by measuring the gas temperature and pressure or gas density in the vicinity of the flow meters. Alternatively, the anode exhaust gas flow rate can be calculated from the component concentrations of the anode exhaust gas, tracer concentration and tracer flow rate by precisely adding a minute amount of a gas that is not contained in the anode exhaust gas as the tracer. The gas analyser shall be calibrated using a standard gas of known mass ratio. Measurement uncertainty shall be evaluated in accordance with the ISO 6974 series, ~~ISO 6141, ISO 6142-1, or ISO 6143.~~

The exhaust gas shall be handled with caution for reasons of safety and the environment, since it ~~may~~ can still contain hydrogen, carbon monoxide and hydrocarbons.

### 7.3.6 Anode exhaust gas component

The anode exhaust gas shall be sampled near the anode gas exhaust port of the cell/stack assembly unit. See 10.6.2.2. The sample shall be analysed using an infrared spectrophotometer, mass spectrometer, gas chromatograph or similar device. If water vapour is likely to affect the measurement, remove water from the gas sample or dilute the sample with argon gas or similar. When measuring, ~~attention shall be paid to~~ the materials, temperature, inner diameter and length of piping shall be selected such as to ensure that any changes the gas composition ~~may~~ can have within the piping are insignificant. In particular, the piping shall be heated where necessary to prevent water vapour from condensing in the piping. The gas analyser shall be calibrated using a standard gas of known mass ratio.

### 7.3.7 Anode exhaust gas temperature

The gas temperature shall be measured near the anode gas exhaust port of the cell/stack assembly unit by selecting a thermocouple ~~or sheathed thermocouple, and an extension lead wire of the type and class that are appropriate and in accordance with IEC 61515, IEC 60584-1 or IEC 60584-3~~ of a type and class in accordance with IEC 60584-1 or sheathed thermocouple of a type and class in accordance with IEC 61515, and an extension lead wire of the type and class IEC 60584-3. If it is difficult to measure the gas temperature during the cell performance test, the anode exhaust gas temperature shall be measured during the preparation of the performance test under the same conditions as those of the performance test.

NOTE There can be significant differences between the temperature of the tube wall and the temperature of the bulk gas.

### 7.3.8 Anode exhaust gas pressure

The anode exhaust gas pressure shall be measured downstream of the anode gas exhaust port of the cell/stack assembly unit by using a pressure sensor, manometer, Bourdon tube or similar device. The measuring instrument should be located in such a manner that the uncertainty is minimized in consideration of any pressure loss within the piping, piping (gas) temperature and other factors. Condensation of water vapour during measurement shall be prevented. One way

~~may~~ can be to measure the pressure by injecting a very small amount of dry nitrogen gas or similar into the pipe close to the measuring instrument.

## 7.4 Cathode gas

### 7.4.1 Cathode gas flow rate

The cathode gas flow rate shall be measured by using mass flow meters, volumetric flow meters or turbine-type flow meters. When measurements are made on a volumetric basis, they shall be converted to mass flow rate by measuring the gas temperature and pressure or gas density in the vicinity of the flow meters. The flow meter shall be selected in consideration of the expected range of flow rates and the allowable uncertainty of the flow meter. Usually, uncertainty shall be evaluated in accordance with ISO 5168 ~~or~~, and if there is non-linearity, evaluated in accordance with ISO 7066-2.

### 7.4.2 Cathode gas component

For cathode gas composition, the oxygen concentration shall be measured using a gas chromatograph or an oxygen concentration meter. The cathode gas should consist of clean (oil-free), compressed air or bottled gas. If a bottled gas mixture is used, the values described on its composition certificate published by the ~~gas~~ fuel supplier ~~may~~ can be used. The uncertainty of the instrument shall be evaluated in accordance with the ISO 6974 series, ~~ISO 6141, ISO 6142-1, ISO 6143 or ISO 6145-7.~~

When it is necessary to measure humidity, a dew point meter, water content meter or gas chromatograph shall be used while controlling the gas temperature to prevent condensation of water vapour.

### 7.4.3 Cathode gas temperature

The gas temperature shall be measured near the cathode gas supply port of the cell/stack assembly unit by selecting a thermocouple ~~or sheathed thermocouple, and an extension leadwire of the type and class that are appropriate in accordance with IEC 60584-1, IEC 60584-3 or IEC 61515~~ of a type and class in accordance with IEC 60584-1 or sheathed thermocouple of a type and class in accordance with IEC 61515, and an extension lead wire of the type and class IEC 60584-3.

NOTE There can be significant differences between the temperature of the tube wall and the temperature of the bulk gas.

If it is difficult to measure the gas temperature during the cell performance test, the cathode gas temperature shall be measured during the preparation of the performance test under the same conditions as those of the performance test.

### 7.4.4 Cathode gas pressure

The cathode gas pressure shall be measured upstream of the cathode gas supply port of the cell/stack assembly unit by using a pressure sensor, manometer, Bourdon tube, or similar device. The measuring instrument should be located in such a manner that the uncertainty is minimized in consideration of any pressure loss within the piping, piping temperature, and other factors.

### 7.4.5 Cathode exhaust gas flow rate

The cathode exhaust gas flow rate shall be measured using a mass flow meter, volumetric flow meter or turbine-type flow meter after cooling the gas. When measurements are made on a volumetric basis, they shall be converted to mass flow rate by measuring the gas temperature and pressure or gas density in the vicinity of the flow meter. The flow meter shall be selected in consideration of the expected range of flow rates and the allowable uncertainty of the instrument. The uncertainty of the instrument shall be evaluated in accordance with the ISO 6974 series, ~~ISO 6141, ISO 6142-1, ISO 6143 or ISO 6145-7.~~

#### 7.4.6 Cathode exhaust gas component

For cathode exhaust gas composition, the oxygen concentration shall be measured using a gas chromatograph or an oxygen concentration meter after cooling the gas. When it is necessary to measure an extremely low water concentration, a dew point meter, water content meter or gas chromatograph shall be used while controlling the gas temperature to prevent condensation of water vapour.

#### 7.4.7 Cathode exhaust gas temperature

The cathode exhaust gas temperature shall be measured near the cathode gas exhaust port of the cell/stack assembly unit by selecting a thermocouple ~~or sheathed thermocouple and an extension leadwire of the type and class that are appropriate in accordance with IEC 60584-1, IEC 60584-3 or IEC 61515~~ of a type and class in accordance with IEC 60584-1 or sheathed thermocouple of a type and class in accordance with IEC 61515, and an extension lead wire of the type and class IEC 60584-3. If it is difficult to measure the gas temperature during the cell performance test, the cathode exhaust gas temperature shall be measured during the preparation of the performance test under the same conditions as those of the performance test.

NOTE There can be significant differences between the temperature of the tube wall and the temperature of the bulk gas.

#### 7.4.8 Cathode exhaust gas pressure

The cathode exhaust gas pressure shall be measured downstream of the cathode gas exhaust port of the cell/stack assembly unit by using a pressure sensor, manometer, Bourdon tube or similar device. The measuring instrument should be located in such a manner that the uncertainty is minimized in consideration of any pressure loss within the piping, piping temperature and other factors.

#### 7.5 Output voltage

A voltage meter shall be connected to the voltage measuring points, as described in Clause 5. The voltage thus measured shall be deemed to be the voltage of the cell/stack. The connecting cable shall be durable enough for the test conditions.

#### 7.6 Output current

A galvanostat or electric load connected to the current lead points, as described in Clause 5, or a current sensor, or both, such as a shunt resistor located within the current circuit, shall be used to measure the current by sending its output to a measuring or recording instrument. The connecting cable shall be selected for appropriate materials and geometry in consideration of the test conditions and possible voltage drop within the cable.

#### 7.7 Cell/stack assembly unit temperature

~~A thermocouple, or sheathed thermocouple, and an extension leadwire of the type and class that are appropriate shall be selected in accordance with IEC 60584-1, IEC 60584-3 or IEC 61515.~~

A thermocouple of a type and class in accordance with IEC 60584-1 or sheathed thermocouple of a type and class in accordance with IEC 61515, and an extension lead wire of the type and class in accordance with IEC 60584-3 shall be selected. They shall be placed at the temperature measuring point as described in Clause 5 and connected with a recorder or similar device for measurement. When there is more than one temperature measuring point, the unit temperature and its distribution shall be obtained by the calculation method recommended by the manufacturer.

#### 7.8 Mechanical load

A mechanical load applied as recommended by the manufacturer shall be measured.

## 7.9 Total impedance

The total impedance of the cell/stack assembly unit shall be measured by either the alternating current (AC) impedance method or the current interruption method. An appropriate measuring line shall be used in order to ensure high-quality data over the entire frequency range investigated.

## 7.10 Ambient conditions

In defining the ambient conditions, ambient temperature, pressure and relative humidity shall be measured. The sampling interval shall be the value specified in ISO 8756 or less.

# 8 Test preparation

## 8.1 General

The type of cell/stack assembly unit to be tested, the number of samples, test parameters, and test conditions shall be determined.

Each measuring instrument shall be checked for its last calibration, the uncertainty under the calibration conditions, or estimated from the class of the instrument, and its dependency on the environmental conditions in order to estimate the uncertainty of the instrument. The method and cycle of calibration and replacement shall be designed to ensure that there is no increase in measurement uncertainty.

The components of the anode and cathode gases and their main impurities shall be verified. As described in Clause 7, a preliminary test shall be performed for gas composition and temperature in order to ensure that the gas compositions are established within the anticipated uncertainty and that the supply gas temperature does not affect the unit temperature. Further, the test procedure, test conditions, and judging criteria for stable state, among others, shall be determined based on the preliminary test results and other factors.

## 8.2 Standard test conditions and test range

The standard test conditions and the typical test range that are recommended by the manufacturer shall be reviewed for the following parameters in order to determine the test conditions and range:

- a) cell/stack assembly unit temperature;
- b) allowable cell/stack assembly unit temperature distribution (if multiple measuring points);
- c) anode gas flow rate;
- d) anode gas composition;
- e) anode gas pressure;
- f) cathode gas flow rate;
- g) cathode gas composition;
- h) cathode gas pressure;
- i) effective fuel utilization;
- j) effective oxygen utilization;
- k) current or current density;
- l) minimum cell/stack assembly unit voltage;
- m) minimum cell/stack assembly unit current (under a constant effective fuel utilization, see Annex E for more information);

- n) maximum cell/stack assembly unit current (under a constant effective fuel utilization). Damage due to excessive degradation is possible beyond this value;
- o) mechanical load.

### 8.3 Components and impurities of anode gas and cathode gas

If gases are used to prepare the anode gas, purity level or components and major impurities of each gas shall be verified by the composition tables published by the respective ~~gas~~ fuel suppliers or through analysis. When the anode gas is produced from liquid fuel, its density, its carbon, hydrogen, and oxygen content, and content of impurities, such as sulfur, shall be verified by the composition table published by the ~~gas~~ fuel supplier or through analysis in accordance with ISO 12185.

The purity or components and major impurities of the cathode gas shall be verified by the composition table published by the ~~gas~~ fuel supplier or through analysis. If a compressor is used, the compressed air shall be free of oil and particles in accordance with ISO 8573-1.

The result of each verification or analysis shall be described in the test report.

### 8.4 Basis of the test procedure

The start-up conditions, such as heating rate and ambient conditions during the heating ramp, the condition of the anode (i.e. the extent of reduction of the nickel oxide to nickel), and the shutdown conditions, such as cooling rate and ambient conditions during the cooling ramp, shall be based on those recommended by the manufacturer or the results of preliminary tests.

### 8.5 Confirmation of aging conditions of unit

The aging conditions of the cell/stack assembly unit shall be determined based on the aging conditions recommended by the manufacturer, as well as the preliminary tests to be conducted, to ensure that the output drift at the time of measurement is insignificant.

### 8.6 Confirmation of criteria of stable state

The tolerance level of variation shall be determined for the output current or output voltage of the cell/stack assembly unit, and the ~~judgement~~ assessment criteria of stable state shall be determined through preliminary testing and others.

The judgement criteria of stable state shall be described in the test report.

### 8.7 Data acquisition method

Preliminary tests shall be conducted while taking into consideration the variation of each test parameter and the sampling rate of each measuring instrument, amongst other things, to determine the sampling interval and the number of samplings and measurements. The sampling interval (e.g. 1 s) shall be short enough to observe the variation of the measured parameter with sufficient time resolution. The number of samplings and repetitions for a single measurement shall be decided so that the total measurement period becomes sufficiently longer than the dominant variation cycle of each test parameter.

## 9 Test procedure

### 9.1 Set-up

The test set-up procedure shall be as follows:

- a) Check each control or measurement subsystem for possible leakage. There are many methods for leak-checking, such as pressure hold and helium leak detectors. The choice of method will depend on the equipment in use. The proper operation of the test equipment should be verified by comparing its performance with the parameters specified in 7.2.
- b) Prepare a cell/stack assembly unit consisting of cell(s), gas passage, interconnectors, current collectors, insulation and other components in accordance with the assembly method and procedure recommended by the manufacturer. Before connecting the cell/stack to the test bench, measure the resistance between cathode and anode current lead points to determine if there is a short-circuit. Measure the resistance between cell voltage measuring points to determine if they are electrically insulated. Measure the cell-to-cell resistances. These values should not indicate a short-circuit, but, rather, should be similar to those specified by the manufacturer.
- c) Set up the cell/stack assembly unit in a temperature control subsystem and install the wiring for voltage measurement and current leads, the mechanical load, and thermocouples as well as the piping for the gas supply and exhaust. Connect the wires to their corresponding subsystems. Ensure proper insulation between thermocouples and the cell/stack. There should also be electrical insulation between the mechanical load and the cell/stack.
- d) Check gas pipe connections for leakage (see 9.1.a).
- e) If ~~needed~~ necessary, verify the wiring for insulation to earth. It is recommended to check the insulation before the output control subsystem or measurement subsystem is connected. In addition, proper wiring shall be verified at joint connections.
- f) When the above are all completed, the measurement subsystem is checked for its proper operation.

### 9.2 Initial conditioning

The cell/stack assembly unit shall be started up at the temperature increasing rate and ambient conditions as specified in 8.4 and operated until it reaches the stable state after going through anode reduction and conditioning.

### 9.3 Shutdown

The shutdown procedure shall be initiated at the specified temperature decreasing rate and ambient conditions as specified in 8.4. The temperature shall be decreased under such conditions as the user has determined, based on preliminary tests or as directed by the manufacturer, unless otherwise provided. During this time, the air flow to the air electrode is maintained and hydrogen diluted with nitrogen (or other inert gas) is flowing to the fuel electrode. The concentration of hydrogen in this gas mixture shall be below the lower explosive limit.

## 10 Performance test

### 10.1 Rated power test

#### 10.1.1 Objective

The objective of this test is to measure and verify the output of the cell/stack assembly unit under rated conditions.

### 10.1.2 Test method

All control parameters shall be set at rated conditions, and after the cell/stack assembly unit has reached the stable state, the voltage, current and other control parameters shall be measured repeatedly at a sampling interval until the number of samples and measurements are obtained as specified in 8.7. The average value of the measurements shall be the measured value. Optionally, include the standard deviation of the measurements.

### 10.1.3 Presentation of results

The measurement results shall be used to calculate the rated power output and recorded in the test report with voltage, current and other measurements of the test conditions.

## 10.2 Current-voltage characteristics test

### 10.2.1 Objective

The objective of this test is to determine the dependency of current-voltage ( $I/V$ ) characteristics on temperature, pressure, gas composition, gas flow rate or effective gas utilization.

### 10.2.2 Test method

#### 10.2.2.1 Test under constant flow rate

The control parameter on which the dependency is to be measured shall be set at its initial value while the anode gas and cathode gas flow rates as well as other control parameters shall be set at those of the test conditions. The unit shall be operated until the stable state is reached under open-circuit conditions, and current-voltage characteristics are measured by changing the current or voltage stepwise or sweeping it at a constant speed. After the measurement, the control parameter is set to the next value and the measurement shall be repeated within the measuring range specified in 8.2.

- a) When current is step-changed, the cell/stack assembly unit shall be operated until it reaches the stable state in each step (temperature and voltage), and at each step measurements are taken over the duration of time at the sampling intervals as specified in 8.7. The average value of the measurements after the stable state is reached shall be the measured value for that step. Optionally, include the standard deviation of the measurements.
- b) When a current sweep is used, the sweep ~~speed~~ rate shall be determined such that the maximum width of the voltage hysteresis does not exceed the voltage variation in the stable state.
- c) When voltage control is used, step a) or b) shall be taken with step voltage or voltage sweep, respectively.

NOTE The meaning of the maximum width of the voltage hysteresis is explained in Annex D.

#### 10.2.2.2 Test under constant effective fuel or oxygen utilization, or both, or constant stoichiometric ratio

The control parameter on which the dependency is to be measured shall be set at its initial value while effective fuel utilization or effective oxygen utilization, or both, and other control parameters shall be set at those of the test operating conditions given in 8.2. The unit shall be operated at the minimum current specified by the manufacturer until it reaches the stable state, and current-voltage characteristics shall be measured by changing the current or voltage stepwise. The unit shall be operated in each step until it reaches the stable state with measurements being taken over the duration of time at the sampling rate as specified in 8.7. The average value of the measurements after the stable state is reached shall be the measured value for that step. Optionally, include the standard deviation of the measurements. After the measurement, the control parameter is set to the next value and the measurement shall be repeated within the measuring range specified in 8.2.

An example of the record of  $I$ - $V$  characteristics test under constant effective fuel utilization is given in Annex E.

### 10.2.3 Presentation of results

The results shall be expressed in a two-dimensional plot with its horizontal axis representing current density or effective fuel utilization, or a combination of current density and fuel utilization, and its vertical axis representing cell/stack assembly unit voltage. This plot shall be included in the test report with the other test conditions. Alternatively to the stack voltage, the measured cell voltages or the average repeating unit voltage ~~may~~ can be plotted against current density. Optionally, include the standard deviation of the measurements.

## 10.3 Effective fuel utilization dependency test

### 10.3.1 Objective

The objective of this test is to study the dependency of the performance of the cell/stack assembly unit on effective fuel utilization and to confirm the maximum effective fuel utilization under various operating conditions. A method for the calculation of effective fuel utilization is described in Annex B.

### 10.3.2 Test method

#### 10.3.2.1 General

The maximum fuel utilization as well as the conditions used to obtain this value shall be obtained from the manufacturer or determined through consultation between the manufacturer and the evaluator.

The test shall be conducted either by decreasing the anode gas flow rate at constant current or by increasing the current at constant anode gas flow rate.

#### 10.3.2.2 Test at constant current

The following steps shall be carried out:

- a) Set the cell/stack assembly unit at the test conditions as specified in 8.2, operate it and verify that it has reached the stable state.
- b) Decrease the anode gas flow rate stepwise until the cell/stack assembly unit reaches the maximum effective fuel utilization as specified by the manufacturer. For each step, verify that the voltage has reached the stable state and record it.
- c) When the fuel utilization reaches the maximum effective value, return the anode gas flow rate stepwise to the original value ~~and record the voltage. Comparing this voltage to that before this step provides information about whether or not the maximum effective fuel utilization specified by the manufacturer as achievable, if only for a short period of time, can be reached by the cell/stack assembly unit. This is different to that seen in 10.4.~~ For each step, verify that the voltage has reached the stable state and record it. The step sizes will or will possibly not match the decreases from the previous step. Comparing this voltage to that before this step provides information about whether the maximum effective fuel utilization specified by the manufacturer which can be achieved by the cell/stack assembly unit.

#### 10.3.2.3 Test at constant anode gas flow rate

The following steps shall be carried out:

- a) Set the cell/stack assembly unit at the specified test conditions, operate it and verify that it has reached the stable state.

- b) Increase the current stepwise until the cell/stack assembly unit reaches the maximum effective fuel utilization. For each step, verify that the voltage has reached the stable state and record it.
- c) When the fuel utilization reaches the maximum effective value, return the current to the original value stepwise and record the voltage. Comparing this voltage to that before this step provides information about the maximum effective fuel utilization which can be achieved by the cell/stack assembly unit, if only for a short period of time. This is different to that seen in 10.4.

The criteria for suspending the test should be determined in advance either by preliminary testing or by consulting with the manufacturer in order to prevent any performance degradation or damage to the cell/stack assembly unit.

### 10.3.3 Presentation of results

The results shall be expressed in a two-dimensional plot with its horizontal axis representing effective fuel utilization, or a combination of anode gas flow rate and effective fuel utilization in the case of 10.3.2.2, and effective fuel utilization, or a combination of current density and effective fuel utilization in the case of 10.3.2.3, and its vertical axis representing the cell/stack assembly unit voltage. Optionally, include the standard deviation of the measurements. This plot shall be included in the test report with the other test conditions.

The stack assembly unit voltage can be replaced by the average repeating unit voltage. Alternatively, the measured cell voltages ~~may~~ can be plotted against current density.

## 10.4 Long term durability test

### 10.4.1 Objective

The objective of this test is to evaluate the performance degradation of the cell/stack assembly unit when it is exposed to certain test conditions over a long period of time and to examine the effect of temperature, current, gas composition, gas impurities and other factors on the durability of the cell/stack assembly unit.

### 10.4.2 Test method

#### 10.4.2.1 General

This test shall be conducted by maintaining the test conditions constant for the duration of the test, either measuring change in the cell/stack assembly unit voltage at constant current or measuring change in the unit voltage together with total resistance. The total resistance shall be measured ~~at a certain interval (100 h to 500 h) by the  $I-V$  curve~~ intervals below 10 % of the total duration of the durability study by the  $I-V$  characteristics in 10.2 or the impedance spectrum in 10.7.

NOTE Other operating modes such as constant voltage with constant gas utilization (constant efficiency operation with decreasing power) or constant power operation and constant gas utilization (decreases efficiency at constant power) or some mixed mode operation, can be used to measure the durability as specified by the test request.

#### 10.4.2.2 Voltage change in long term durability test

Set up all the controlling parameters at the specified test conditions and measure the voltage of the cell/stack assembly unit at regular intervals. The measured values shall be used to obtain the rate of voltage change for the entire test period or a specified duration.

#### 10.4.2.3 Total resistance change in long term durability test

This test ~~may~~ can be performed at the same time as the test carried out in accordance with 10.4.2.2.

The following method shall be used for the measurement of total resistance:

- a) Measure whole  $I$ - $V$  characteristics between 0 to maximum current as specified in 8.2, or measure partial  $I$ - $V$  characteristics in the vicinity of the holding current, both in a similar manner to that stated in 10.2.
- b) Derive the approximate tangent to the  $I$ - $V$ -~~curve~~ characteristics at holding current by connecting two points on the  $I$ - $V$ -~~curve~~ characteristics in the vicinity across the holding current and find the slope of the tangent. Report the slope as total resistance.  
~~Selection of the points in the vicinity of the holding current requires care. The points should be closer to the holding current when the curvature of the  $I$ - $V$  curve is large.~~ The points in the vicinity of the holding current are selected such that they are closer to the holding current when the curvature of the  $I$ - $V$  characteristics is large. The uncertainty in the voltage and current measurements should be kept as small as possible.
- c) After measuring the  $I$ - $V$ -~~curve~~ characteristics, restore the values to those of the original test conditions and measure voltage until the next measuring cycle.
- d) Repeat this measurement at a certain interval throughout the test duration.

The results shall be used to calculate the voltage variation rate and total resistance variation rate for the entire test period or specific time duration within the test period.

It is also possible to measure the total impedance described in 10.7 at the time of total resistance measurement.

#### 10.4.3 Presentation of results

In the case of 10.4.2.2, the results shall be expressed in a two-dimensional plot with the horizontal axis representing time and the vertical axis representing voltage and shall be included in the test report with the voltage change rate and test conditions. In the case of 10.4.2.3, the vertical axis shall represent voltage and total resistance, and the plot shall be included in the test report along with voltage change rate, total resistance variation rate and test conditions.

The stack assembly unit voltage ~~may be replaced with the average repeating unit voltage.~~ ~~Alternatively, the measured cell voltages may~~ can be plotted against time.

NOTE These results can also be represented as area-specific resistance (or impedance). Here, the area-specific resistance (impedance) equals the measured resistance (impedance) multiplied by the active electrode area.

### 10.5 Thermal cycling durability test

#### 10.5.1 Objective

The objective of this test is to evaluate the durability of the cell/stack assembly unit with thermal cycling. The thermal cycles shall be within the manufacturer's specifications.

#### 10.5.2 Test method

##### 10.5.2.1 General

For this test, the following test conditions shall be obtained from the manufacturer or determined through consultation between the manufacturer and the evaluator in advance.

The operating temperature shall be the temperature of the standard test conditions.

- a) number of thermal cycles;
- b) cooling rate;
- c) heating rate;
- d) maximum temperature;

- e) minimum temperature;
- f) operating conditions at operating temperature;
- g) period to maintain operating temperature;
- h) period to maintain minimum temperature;
- i) total test period;
- j) gas flow rate and composition at heating and cooling conditions, and minimum temperature, respectively.

### 10.5.2.2 Test procedure

In accordance with the above test conditions, either the method of measuring cell/stack assembly unit voltage variation at the operating temperature or measuring total resistance together with cell/stack assembly unit voltage shall be chosen. When measuring total resistance, follow the method of 10.4.2.3. The voltage shall be measured over the specified duration. After a certain operating ~~temperature~~ period, the temperature of the cell/stack assembly unit shall be decreased to the minimum temperature with the cell/stack assembly unit at open circuit under the specified conditions, and the minimum temperature shall be maintained for the specified time. The temperature of the cell/stack assembly unit shall then be raised under the specified conditions to the operating temperature and the measurement shall resume as before.

These measurements shall be repeated until the specified number of thermal cycles is reached. The results obtained shall be used to calculate the voltage variation rate and total resistance variation rate over the entire test period or specific duration within the test period.

### 10.5.3 Presentation of results

The results shall be expressed in a two-dimensional plot with the horizontal axis representing either (a) time or (b) number of cycles. The test conditions shall also be described in the test report.

In the case (a), unit voltage, ~~total resistance (optional)~~, and temperature shall be plotted on the vertical axis. Total resistance shall be plotted if applicable.

In the case (b), unit voltage ~~and total resistance (optional)~~ shall be plotted on the vertical axis. Total resistance shall be plotted if applicable.

## 10.6 Internal reforming performance test

### 10.6.1 Objective

The objective of this test is to evaluate the internal reforming (or cracking) performance of the cell/stack assembly unit under the open-circuit conditions or rated conditions against hydrocarbons (HC) such as methane or ammonia (NH<sub>3</sub>) contained in the anode gas.

### 10.6.2 Test method

#### 10.6.2.1 General

This test shall be performed in accordance with the manufacturer's recommendation or after consultation between the manufacturer and the evaluator regarding the cell/stack's ability to reform HCs (or crack NH<sub>3</sub>) internally; it is possible that some cell/stack assembly units ~~may~~ will not be able to do so.

NOTE Many complex reactions can occur with HC-containing anode gases, depending on composition and thermodynamic equilibria. These reactions can affect the temperature and pressure gradients in the cell and stack.

This test is applicable when anode gas and anode exhaust gas can be sampled without any mixing with cathode gas or cathode exhaust gas.

### 10.6.2.2 Test procedure

Anode gas containing HCs (or  $\text{NH}_3$ ) shall be supplied to the cell/stack assembly unit. After the unit reaches the stable state in the open circuit or rated conditions, the anode gas and anode exhaust gas shall be analysed for their compositions; these are used to calculate the HC (or  $\text{NH}_3$ ) conversion rate that indicates the internal reforming (or cracking) characteristic.

The HC (or  $\text{NH}_3$ ) conversion rate for a specific HC<sub>j</sub> ( $\text{NH}_3$ ),  $\xi_j$  (%) is calculated as follows:

$$\xi_j = 100 (q_{\text{HC},j,\text{in}} - q_{\text{HC},j,\text{out}}) / q_{\text{HC},j,\text{in}} \quad (2)$$

$$\xi_j = 100 (q_{\text{NH}_3,j,\text{in}} - q_{\text{NH}_3,j,\text{out}}) / q_{\text{NH}_3,j,\text{in}} \quad (3)$$

where  $q_{\text{HC},j,\text{in}}$  ( $q_{\text{NH}_3,j,\text{in}}$ ) and  $q_{\text{HC},j,\text{out}}$  ( $q_{\text{NH}_3,j,\text{out}}$ ) represent the flow rates of the specific HC<sub>j</sub> ( $\text{NH}_3$ ) at the anode inlet and outlet, respectively, which are calculated from the anode gas flow rate and its HC ( $\text{NH}_3$ ) concentration, and the anode exhaust gas flow rate and its HC ( $\text{NH}_3$ ) concentration, respectively.

### 10.6.3 Presentation of results

The compositions and flow rates of anode gas and anode exhaust gas as well as the HC (or  $\text{NH}_3$ ) conversion rate shall be described in the test report with the test conditions.

## 10.7 Resistance components identification test

### 10.7.1 Objective

The main objective of this test is to identify and evaluate ohmic and non-ohmic components of the total resistance of the cell/stack assembly unit.

### 10.7.2 Test method

#### 10.7.2.1 General

The separation of resistance components of the cell/stack assembly unit shall be evaluated by either the AC impedance method or the current interruption method.

#### 10.7.2.2 AC electrochemical impedance method

The following test conditions shall be determined in advance by conducting preliminary tests.

- Measuring range for frequencies:

The highest frequency should roughly identify point A, and the lowest frequency should roughly identify point C when plotted in the complex impedance diagram (see Figure 2).

- Number of measuring points:

Four to twenty points per one order of frequencies (to be distributed evenly as logarithms, if possible) ~~shall be~~ are required; they shall be numerous enough to identify clearly the geometry of impedance plots. If possible, avoid the fundamental and harmonics of the electrical grid frequency.

The test shall be conducted using the following procedure:

- a) establish the test conditions;
- b) verify that the stable state has been reached;
- c) superimpose AC sinusoidal waves on DC current or voltage and start measurements. Sweep the AC sinusoidal waves within the specified frequency range and measure the impedance at each frequency.

The amplitude of the AC signal for the measurement shall be enough to activate the cell but not overly polarize it. The amplitude per cell ~~may~~ can be obtained by dividing the total voltage amplitude with the number of cells in series. As an option, the validity of the impedance spectrum shall be verified by using appropriate validation relations such as the Kramers-Kronig (KK) relationships or Z-hit.

#### 10.7.2.3 Current interruption method

When employing this method, the current interruption characteristics and sampling rate to allow the evaluation of a target property of the cell/stack assembly unit shall be identified by preliminary testing. Measurements shall be taken after ensuring that the unit is in the stable state under the test conditions.

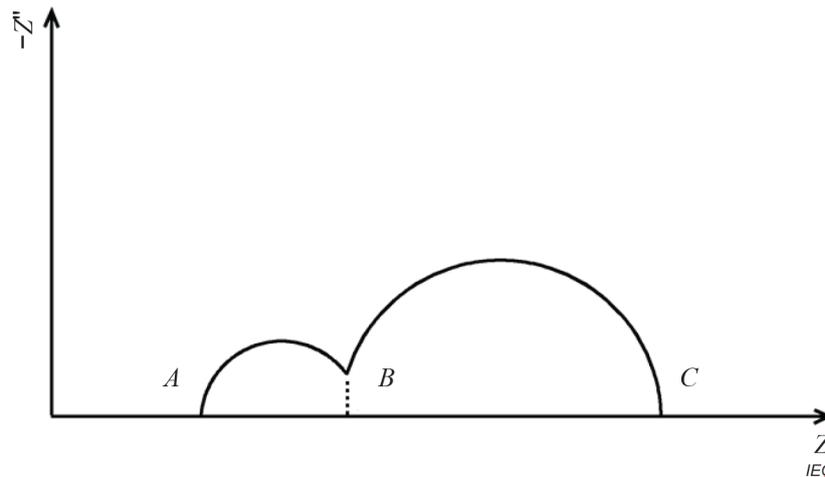
#### 10.7.3 Presentation of results

- a) AC impedance method

The test results shall be expressed as a complex impedance or Cole-Cole plot (indicate characteristic frequencies) or Bode plot (a plot of impedance components against the logarithm of measured frequency), or both. The impedance times unit area calculated by active area defined in 3.1.2 shall be plotted.

- b) Current interruption method

The current (density) and voltage response waveforms observed before and after current interruption shall be plotted against the time axis. The sampling rate shall be appropriate to identify the ohmic resistance component. The ohmic resistance component so obtained shall be reported with the test conditions.

**Key**

<i>A</i>	high-frequency end impedance
<i>C</i>	low-frequency end impedance
<i>A-B</i>	high-frequency arc impedance
<i>B-C</i>	low-frequency arc impedance
$Z'$	real part of impedance
$-Z''$	negative imaginary part of impedance

**Figure 2 – Typical diagram of complex impedance plot for SOFC**

NOTE 1 These results can also be represented as area-specific resistance (or impedance). Here, the area-specific resistance (impedance) equals the measured resistance (impedance) multiplied by the active electrode area.

The data collection interval shall be selected to be able to define the current switch-off point accurately within 1  $\mu$ s. High frequency interference that increases the error shall be cared in determining the ohmic resistance.

## 11 Test report

### 11.1 General

Test reports shall accurately, clearly and objectively present sufficient information to demonstrate that all the objectives of the tests have been attained. A suggested template for the test report is given in Annex F.

A controlled document policy should be used within an engineering change control process.

Test reports can indicate the test procedure document revision that is used for the test and can detail any agreed amendments to the released test procedure in an annexure.

The test report can be archived in electronic form.

### 11.2 Report items

The report shall present the following information, at a minimum:

- title of the report;
- authors of the report;
- date of the report;
- test report reference or identification number;

- e) location and (start) date and time of the test;
- f) test bench used;
- g) test unit data (see 11.3 for details);
- h) test conditions (see 11.4 for details);
- i) test data (see 11.5 for details).

### 11.3 Test unit data description

Test unit data shall include the following information, at a minimum:

- a) product name and brand name of the unit;
- b) active electrode area;
- c) number of cells (total, series, parallel);
- d) cell materials and thicknesses, if known, and cell identification number(s);
- e) stacking materials, if known;
- f) geometry of the unit;
- g) temperature measurement and load application positions.

### 11.4 Test conditions description

The test conditions description shall include the following information, at a minimum:

- a) name of person(s) and entity conducting the test;
- b) instruments and calibration record;
- c) test procedure;
- d) aging conditions;
- e) criteria of stable state;
- f) data acquisition method;
- g) gas purity and impurities;
- h) test bench layout.

### 11.5 Test data description

Test data shall include the following information:

- a) title of the test(s);
- b) test operating conditions;
- c) test result;
- d) ambient conditions;
- e) uncertainty evaluation (see 11.6 for details).

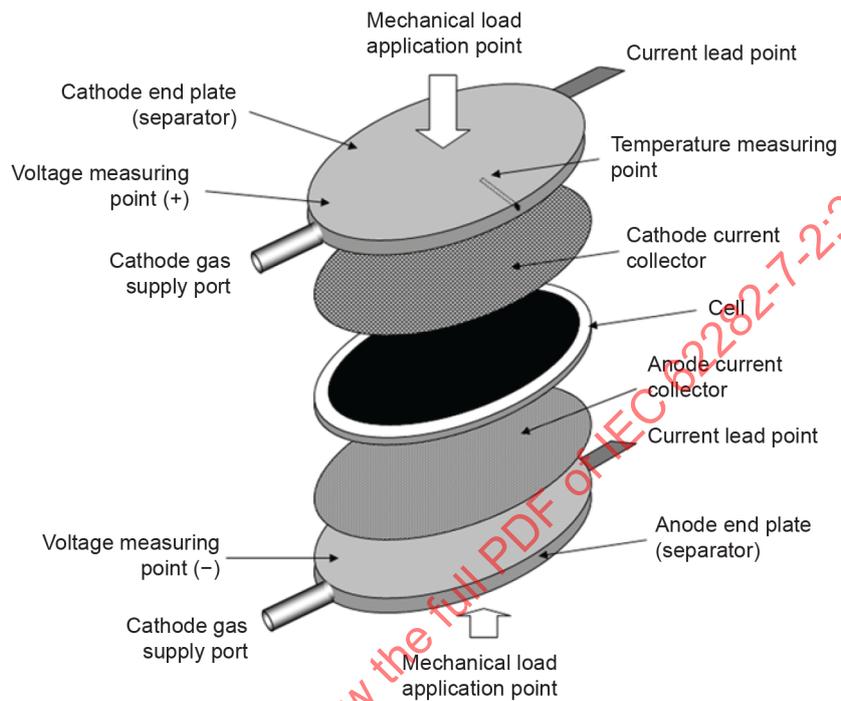
### 11.6 Uncertainty evaluation

Uncertainties of instruments shall be reported. If necessary, variation of measurements or measurement uncertainties, or both, calculated from the variation of measurements and uncertainties of instruments should be reported.

## Annex A (informative)

### Example of cell assembly unit

An example configuration and test boundary of a cell assembly unit described in this document is shown in Figure A.1.



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**Figure A.1 – Example of cell assembly unit**

In this schematic example, the cathode current collector and anode current collector also work as cathode gas flow channel and anode gas flow channel, respectively. The exhaust gas comes out from the respective circumference of each current collector.

**NOTE** It is possible to measure the voltage at the current collector to exclude the voltage drop due to contact resistances (plate-current collector).

## Annex B (informative)

### Calculation of effective fuel utilization

#### B.1 General

Annex B describes a method for the calculation of effective fuel utilization as defined in 3.1.10.

#### B.2 Calculation method

In a performance test, an anode gas is supplied at a rate of  $q_a$  (l/min (STP)). The flow rate of each fuel component in the anode gas is expressed as  $q_j$  (l/min (STP)) ( $j = \text{H}_2, \text{CO}, \text{CH}_4, \dots, \text{C}_p\text{H}_q\text{O}_r, \text{NH}_3$ ) where  $\text{C}_p\text{H}_q\text{O}_r$  is the chemical formula of a general hydrocarbon fuel. In the case where the anode gas composition is analysed,  $q_j$  shall be calculated from the molar fraction of each fuel component ( $x_j$  (mol/mol)) and  $q_a$  using Equation (B.1):

$$q_j = x_j \times q_a \tag{B.1}$$

In the ~~example that follows~~ cell/stack assembly unit, it is assumed that  $N$  cells are connected in series and that the fuel is uniformly distributed between the cells. A theoretical current defined in 3.1.9,  $I_{\text{theory}}$  (A), assuming that the supplied fuel gas is completely consumed in electrochemical reactions, shall be calculated from Equation (B.2):

$$I_{\text{theory}} = \frac{P_{\text{st}}}{R \times T_{\text{st}} \times 60 \times 1\,000} \times F \times \left[ \frac{\sum_j n_j \times q_j}{N} \right] \tag{B.2}$$

$$= \frac{101\,325}{8,314\,51 \times 273,15 \times 60 \times 1\,000} \times 96\,485 \times \left[ \frac{\sum_j n_j \times q_j}{N} \right] = 71,74 \times \left[ \frac{\sum_j n_j \times q_j}{N} \right]$$

where

$P_{\text{st}}$  is the standard pressure (101 325 Nm<sup>-2</sup>);

$T_{\text{st}}$  is the standard temperature (273,15 K);

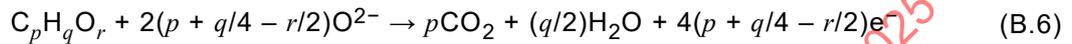
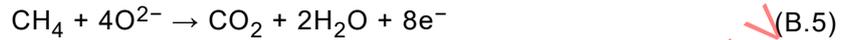
$R$  is the universal gas constant (8,314 4 Jmol<sup>-1</sup>K<sup>-1</sup>);

$F$  is Faraday's constant (96 485 C/mol);

$n_j$  is the number of electrons transferred when one molecule of fuel component  $j$  is electrochemically reacted;

$N$  is the number of cells in series.

The electron transferred,  $n_j$  for representative fuels, is determined by Equation (B.3) to Equation (B.6) and is summarized in Table B.1. For the general  $C_pH_qO_r$  component,  $n_j$  is equal to  $4(p + q/4 - r/2)$ :



The measured current output of each cell or that from the total cell/stack assembly unit is expressed as  $I_{\text{measured}}$ . Therefore, effective fuel utilization, or  $U_f$  (%) can be calculated from Equation (B.8):

$$U_f = \frac{I_{\text{measured}}}{I_{\text{theory}}} \times 100 \% \quad (B.8)$$

**Table B.1. –  $n_j$  for representative fuels**

Fuel	$n_j$
H <sub>2</sub>	2
CO	2
CH <sub>4</sub>	8
C <sub>p</sub> H <sub>q</sub> O <sub>r</sub>	4(p + q/4 - r/2)
NH <sub>3</sub>	3

### B.3 Calculation examples

#### B.3.1 Calculation from anode gas composition and flow rate

Normalized anode gas mole fraction,  $x_i$ , is assumed to be as indicated in Table B.2 as a result of anode gas composition analysis. It is presumed that anode gas flow rate,  $q_a$ , is 0,500 l/min (STP). The flow rates,  $q_j$ , of H<sub>2</sub>, CO and CH<sub>4</sub>, which are fuel components in the anode gas, are calculated using Equation (B.1) as 0,281, 0,047, 0,003 l/min (STP), respectively. Then,  $\sum_j n_j q_j$  is calculated by summing up each  $n_j \times q_j$ , leading to  $\sum_j n_j q_j = 0,562 + 0,094 + 0,024 = 0,680$  l/min (STP). Hence, if  $N = 10$  cells,  $I_{\text{theory}} = 71,74 \times 0,680/10 = 4,88$  A is obtained using Equation (B.2). If it is assumed that the actual output current of the stack is 3,90 A,  $I_{\text{measured}}$  is equal to 3,90 A. Therefore, effective fuel utilization can be calculated using Equation (B.8) as  $U_f = \frac{3,90}{4,88} \times 100 = 80 \%$ .

**Table B.2 – Anode gas composition, flow rate of each fuel component  $q_j$ , and  $n_j q_j$**

Component	$x_i$ mol %	$q_j$ l/min (STP)	$n_j q_j$ l/min (STP)
H <sub>2</sub>	56,1	$56,1/100 \times 0,500 = 0,281$	$2 \times 0,281 = 0,562$
H <sub>2</sub> O	27,1		
CO	9,3	$9,3/100 \times 0,500 = 0,047$	$2 \times 0,047 = 0,094$
CO <sub>2</sub>	7,1		
CH <sub>4</sub>	0,5	$0,5/100 \times 0,500 = 0,003$	$8 \times 0,003 = 0,024$

#### B.3.2 Calculation from supplied H<sub>2</sub> and H<sub>2</sub>O flow rate

It is assumed that H<sub>2</sub> and H<sub>2</sub>O are supplied to the anode by controlling each flow rate. It is also assumed that in a performance test of a 40-cell-stack in which the number of series connection is 10 with 4 parallel connections, H<sub>2</sub> flow rate,  $q_j$ , and output current are to be equal to 3,00 l/min (STP) and 32,3 A, respectively. Using Equation (B.2),  $I_{\text{theory}}$  is calculated as  $71,74 \times (2 \times 3,00) / 10 = 43,0$  A. Therefore, with  $I_{\text{measured}} = 32,3$  A, effective fuel utilization can be calculated as  $U_f = \frac{32,3}{43,0} \times 100 = 75,1 \%$  from Equation (B.8).

NOTE The number of parallel connections in the stack does not make any difference to the calculation.

## Annex C (informative)

### Calculation of effective oxygen utilization

#### C.1 General

Annex C describes a method for the calculation of effective oxygen utilization as defined in 3.1.11.

#### C.2 Calculation method

In a performance test, a cathode gas is supplied at a rate of  $q_c$  (l/min (STP)). Oxygen flow rate ( $q_{O_2}$  (l/min (STP))) shall be calculated from the oxygen molar fraction in the cathode gas, or  $x_{O_2}$  (mol/mol), using Equation (C.1):

$$q_{O_2} = x_{O_2} \times q_c \quad (C.1)$$

The theoretical current defined in 3.1.9,  $I_{\text{theory}}$  (A), assuming that the cathode gas is uniformly distributed among  $N$  cells connected in series in the stack and that the cathode gas is completely consumed in electrochemical reactions, shall be calculated from Equation (C.2):

$$\begin{aligned} I_{\text{theory}} &= \frac{P_{\text{st}}}{R \times T_{\text{st}} \times 60 \times 1\,000} \times F \times \frac{n_{O_2} \times q_{O_2}}{N} \\ &= \frac{101\,325}{8,314\,51 \times 273,15 \times 60 \times 1\,000} \times 96\,485 \times \frac{n_{O_2} \times q_{O_2}}{N} = 287,0 \times \frac{q_{O_2}}{N} \end{aligned} \quad (C.2)$$

where

$P_{\text{st}}$  is the standard pressure (101 325 Nm<sup>-2</sup>);

$T_{\text{st}}$  is the standard temperature (273,15 K);

$R$  is the universal gas constant (8,314 4 Jmol<sup>-1</sup>K<sup>-1</sup>);

$F$  is Faraday's constant (96 485 C/mol);

$n_{O_2}$  is the number of electrons transferred when one molecule of oxygen is electrochemically reduced, leading to  $n_{O_2} = 4$  as shown in Equation (C.3);

$N$  is the number of cells connected in series:



The measured current output of each cell or that from the total cell/stack assembly unit is expressed as  $I_{\text{measured}}$ . Therefore, effective oxygen utilization, or  $U_{O_2}$  (%) shall be calculated from Equation (C.4):

$$U_{O_2} = \frac{I_{\text{measured}}}{I_{\text{theory}}} \times 100 \% \quad (C.4)$$

### C.3 Calculation example

In a performance test, it is assumed that the cathode gas flow rate,  $f_c$ , is 1,50 l/min (STP), and that there are  $N = 10$  cells connected in series in the stack with the gas composition as indicated in Table C.1.  $f_{O_2} = 0,314$  l/min (STP) and  $I_{\text{theory}} = 9,01$  A can be obtained using Equation (C.1) and Equation (C.2), respectively. When the actual output current at the stack performance test is 2,70 A,  $I_{\text{measured}}$  is equal to 2,70 A. Therefore, effective oxygen utilization can be calculated as  $U_{O_2} = 2,70/9,01 \times 100 = 30$  % using Equation (C.4).

**Table C.1 – Cathode gas composition,  $q_{O_2}$ , and  $I_{\text{theory}}$**

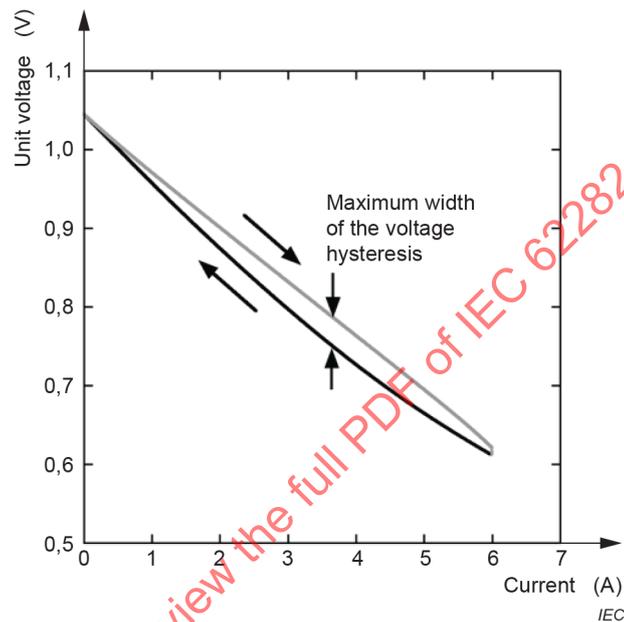
Component	$x_i$ mol %	$q_{O_2}$ l/min (STP)	$I_{\text{theory}}$ A
O <sub>2</sub>	20,95	$20,95/100 \times 1,50 = 0,314$	$287,0 \times 0,314/10 = 9,01$
N <sub>2</sub>	79,05		

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## Annex D (informative)

### Maximum width of the voltage hysteresis in I-V characteristics test

When  $I$ - $V$  characteristics are taken with the current sweep method described in 10.2.2.1 b), measured voltages ~~may~~ can be different with a different sweep rate due to hysteresis, as shown in the example in Figure D.1. The appropriate sweep rate shall be determined in such a way that the maximum width of the voltage hysteresis is smaller than the allowable maximum variation of voltage that is defined in 7.2.



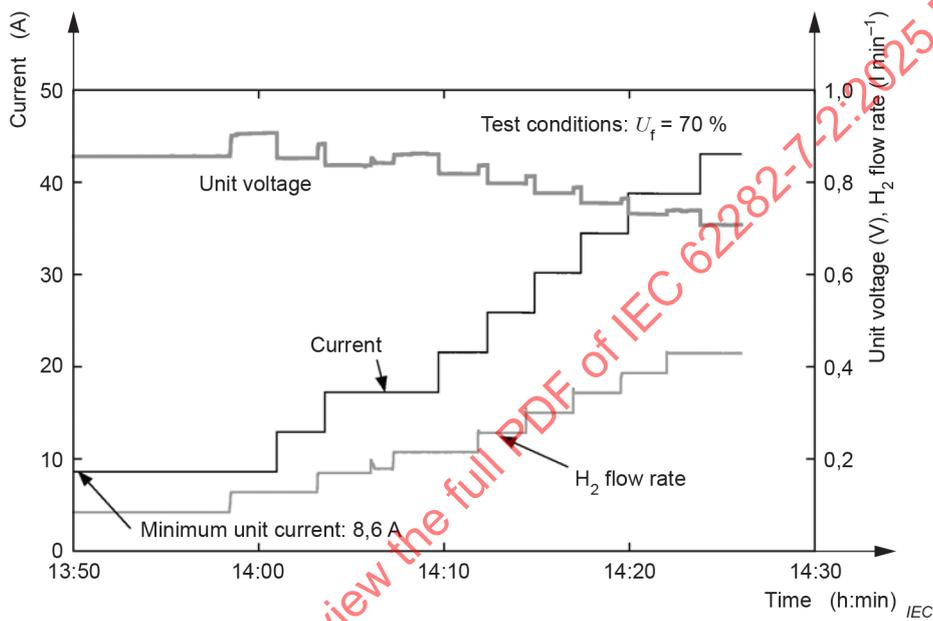
NOTE In this Figure D.1, the right pointing arrow relates to increasing current that is from OCV to maximum current and the left pointing arrow relates to decreasing current that is from maximum current to OCV.

**Figure D.1 – Voltage hysteresis at a given sweep rate in  $I$ - $V$  characteristics test**

**Annex E**  
(informative)

**Current-voltage characteristics test  
under constant effective fuel utilization**

In order to keep the effective fuel or oxygen utilization constant, or both in the measurement range, fuel or oxygen flow rates, or both are changed as the current changes. Dependency of unit voltage on current accordingly differs from those observed under a constant gas flow rate. An example of such changes among the unit voltage, current and hydrogen flow rate is shown in Figure E.1.



**Figure E.1 – Example of the record in current-voltage characteristics test under constant effective fuel utilization at increasing steps in current**

When  $I-V$  characteristics are taken under a constant effective fuel utilization, the unit shall initially be kept at the minimum cell/stack assembly unit current as defined in 8.2 m).

At low current, voltage becomes unstable due to low gas flow rate. Therefore, the minimum cell/stack assembly unit current should be determined so as to avoid such unstable voltage.

## Annex F (informative)

### Test report (template)

#### F.1 Overview

Examples of a report for general information, test unit data description and test conditions as well as a test report for each test specified in the body text are given below. Instructions to the author are given in *italics* and should not be included in the test report. The method for determining "instrument uncertainty" shown in Clause F.5, Clause F.6, Clause F.7, Clause F.8 and Clause F.9 is given in Annex G.

#### F.2 General information

Test report title	
Author(s) of report	
Date of report	
Test report reference or identification number	
Location of test	
Start date and time of test	
Test bench	

#### F.3 Test unit data description

Product name and brand name of the unit	
Active electrode area	
Number of cells (total, series, parallel)	
Unit identification number	
Geometry of the unit	
Materials and thickness of electrolyte and electrodes, interconnect	<i>If known</i>

*If the following are not available from the manufacturer, they shall be reported.*

Configuration of assembly unit and assembling method	
Materials and geometry of the peripheral components	
Flow patterns and directions of anode and cathode gases	
Temperature measurement positions	
Mechanical load (unit) and its application positions	
Voltage measurement positions	
Current lead positions	
Minimum cell voltage (unit)	

### F.4 Test conditions

Name of person(s) and entity conducting the test	
Instruments and calibration record	
Test procedure	
Aging conditions	
Criterion or criteria of stable state	
Data acquisition method	
Gas purity and impurities	
Mechanical load	

### F.5 Rated power test

Operating conditions

Input	Value (unit)	Instrument uncertainty (unit)	Variation of measurement (Standard deviation (combined)) (unit)
$q_a$			
$q_c$			
$p_a$			
$p_c$			
$T_{op}$			

Test results

Output	Value (unit)	Instrument uncertainty (unit)	Variation of measurement (Standard deviation (combined)) (unit)
OCV			
$V$			
$I$			
$P$			

The data average method shall be described.

### F.6 Current-voltage characteristics test

Operating conditions

Input	Value (unit)
$q_a$ (or its range in the case of 10.2.2.2)	
$q_c$ (or its range in the case of 10.2.2.2)	
$p_a$	
$p_c$	
$T_{op}$	
$U_f$ (in the case of 10.2.2.2)	
$U_{02}$ (in the case of 10.2.2.2)	

Test results

Output	Value (unit)	Instrument uncertainty (unit)	Variation of measurement (Standard deviation) (unit)
OCV			

The  $I$ - $V$ -curve characteristics shall be presented (see 10.2.3).

## F.7 Effective fuel utilization dependency test

Initial operating conditions

Input	Value (unit)
$q_a$	
$q_c$	
$p_a$	
$p_c$	
$T_{op}$	
$I$	

Preliminary information

Expected maximum fuel utilization	/ %
-----------------------------------	-----

Test results

Output	Value (unit)	Instrument uncertainty (unit)	Variation of measurement (Standard deviation) (unit)
OCV			

In the case of 10.3.2.2

Step number	$f_a$ (unit)	$V$ (unit)	$U_f$ / %
0			
1			
2			
3			
$m$			

The  $U_f(f_a)$ - $V$  curve that can replace the above table (see 10.3.3) shall be presented.

In the case of 10.3.2.3

Step number	$I$ (unit)	$V$ (unit)	$U_f I$ %
0			
1			
2			
3			
$m$			

The  $U_f(I) - V$  curve that can replace the above table (see 10.3.3) shall be presented.

### F.8 Long-term durability test

Operating conditions

Input	Value (unit)
$q_a$	
$U_f$	
$q_c$	
$U_{O_2}$	
$p_a$	
$p_c$	
$T_{op}$	
$I$	

Test results

Output	Value (unit)	Instrument uncertainty (unit)	Variation of measurement (Standard deviation) (unit)
OCV (start)			
OCV (end)			
$i$ (end)			

The other test results should be presented by means of a two-dimensional plot (see 10.4.3).

## F.9 Thermal cycling durability test

### Test conditions

Cooling rate	
Heating rate	
Minimum temperature	
Period to maintain operating temperature	
Period to maintain minimum temperature	
Gas composition during heating	
Gas flow rate during heating	
Gas composition during cooling	
Gas flow rate during cooling	
Gas composition at minimum temperature	
Gas flow rate at minimum temperature	

### Operating conditions

Input	Value (unit)
$q_a$	
$q_c$	
$p_a$	
$p_c$	
$T_{op}$	
$I$	

### Test results

Output	Value (unit)	Instrument uncertainty (unit)	Variation of measurement (standard deviation) (unit)
OCV (start)			
OCV (end)			

The other results should be presented as specified in 10.5.3.

### F.10 Internal reforming performance test

Operating conditions

Input	Value (unit)
$q_a$	
$q_c$	
$U_f$	
$p_a$	
$p_c$	
$T_{op}$	
$I$	

Test results

Output	Value (unit)
OCV	
Composition of anode exhaust gas	
Flow rate of anode exhaust gas	
HC conversion rate (Optional)	/ %

### F.11 Resistance components identification test

Operating conditions

Input	Value (unit)
$q_a$	
$q_c$	
$p_a$	
$p_c$	
$T_{op}$	
$I$ or $V$	
Frequency range	
Operating mode (galvanostatic or potentiostatic)	
Amplitude	

Test results

Output	Value (unit)
Total resistance	
Ohmic resistance	

Depending upon the method, corresponding figures shall be attached (see 10.7.3).

## Annex G (informative)

### Method for determining instrument expanded uncertainty

Instrument expanded uncertainty,  $U_1$ , can be obtained by the calibrations using traceable standard instruments. See ISO/IEC Guide 98-3 for further information.

It ~~may~~ can also be obtained from the error limit ( $\pm a$ ) of instrument as shown in Formula (G.1) assuming uniform distribution of the probability within the error limit range:

~~$$U_1 = 2u_1 = 2 \frac{a}{\sqrt{3}} \quad (\text{G.1})$$~~

$$U_1 = 2u_c = 2 \frac{a}{\sqrt{3}} \quad (\text{G.1})$$

where  ~~$u_1$~~   $u_c$  is the standard instrument uncertainty.

Some of the measurements are made using several instruments (e.g. for current measurement, the combination of current sensor and digital voltage recorder ~~may~~ can be used). The standard uncertainty for such a case can be obtained as shown in Formula (G.2) assuming no correlation between the instruments:

~~$$u_1^2 = \sum u_{1,i}^2 \quad (\text{G.2})$$~~

where  ~~$u_{1,i}$~~  represents standard instrument uncertainty for the  $i^{\text{th}}$  instrument.

$$u_c^2 = \sum_j \left( \frac{\partial f}{\partial x_j} \right)^2 u_{1,j}^2 \quad (\text{G.2})$$

where

$f$  (e.g. power as the product of current and voltage) is the functional relation of the instruments (current sensor and digital voltage recorder) to determine the derived quantity (power);

$x_j$  is the parameter (current or voltage) measured by the  $j^{\text{th}}$  instrument;

$u_{1,j}$  represents standard instrument uncertainty of the  $j^{\text{th}}$  instrument.

Therefore,

$$\cancel{U_1 = 2u_1 = 2\left(\sum u_{1,i}\right)^{\frac{1}{2}}} \quad (\text{G.3})$$

$$U_1 = 2u_c = 2\left(\sum_j \left(\frac{\partial f}{\partial x_j}\right)^2 u_{1,j}^2\right)^{\frac{1}{2}} \quad (\text{G.3})$$

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# INTERNATIONAL STANDARD

## NORME INTERNATIONALE

**Fuel cell technologies –  
Part 7-2: Test methods – Single cell and stack performance tests for solid oxide  
fuel cells (SOFCs)**

**Technologies des piles à combustible –  
Partie 7-2: Méthodes d'essai – Essais de performance de cellule élémentaire et  
de pile pour les piles à combustible à oxyde solide (SOFC)**

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## INTERNATIONAL ELECTROTECHNICAL COMMISSION

## FUEL CELL TECHNOLOGIES –

**Part 7-2: Test methods – Single cell and stack performance tests  
for solid oxide fuel cells (SOFCs)**

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This second edition cancels and replaces the first edition published in 2021. This edition constitutes a technical revision.

This edition includes the following significant technical changes with respect to the previous edition:

- a) Table 1 has been revised to specify the units missing for some terms;
- b) bibliographical entries (ISO/TR 15916, SOCTESQA test modules and ISO/IEC Guide 98-6:2021) have been added to provide further information.

The text of this International Standard is based on the following documents:

Draft	Report on voting
105/1093/FDIS	105/1099/RVD

Full information on the voting for its approval can be found in the report on voting indicated in the above table.

The language used for the development of this International Standard is English.

This document was drafted in accordance with ISO/IEC Directives, Part 2, and developed in accordance with ISO/IEC Directives, Part 1 and ISO/IEC Directives, IEC Supplement, available at [www.iec.ch/members\\_experts/refdocs](http://www.iec.ch/members_experts/refdocs). The main document types developed by IEC are described in greater detail at [www.iec.ch/publications](http://www.iec.ch/publications).

A list of all parts in the IEC 62282 series, published under the general title *Fuel cell technologies*, can be found on the IEC website.

The committee has decided that the contents of this document will remain unchanged until the stability date indicated on the IEC website under [webstore.iec.ch](http://webstore.iec.ch) in the data related to the specific document. At this date, the document will be

- reconfirmed,
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## INTRODUCTION

Solid oxide fuel cells (SOFCs) have a broad range of geometry and size. As such, in general, peripherals like current collectors and gas manifolds are unique to each cell or stack and are often incorporated into a cell or stack to form one integrated unit. In addition, they tend to have a significant effect on the power generation characteristics of the cell or stack. This document therefore introduces as its subject "cell/stack assembly units", which are defined as those units containing not only a cell or stack but also peripherals.

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## FUEL CELL TECHNOLOGIES –

### Part 7-2: Test methods – Single cell and stack performance tests for solid oxide fuel cells (SOFCs)

#### 1 Scope

This part of IEC 62282 applies to SOFC cell/stack assembly units, testing systems, instruments and measuring methods, and specifies test methods to test the performance of SOFC cells and stacks.

This document is not applicable to small button cells that are designed for SOFC material testing and provide no practical means of fuel utilization measurement.

This document is used based on the recommendation of the entity that provides the cell performance specification or for acquiring data on a cell or stack in order to estimate the performance of a system based on it. Users of this document can selectively execute test items suitable for their purposes from those described in this document.

Users can substitute selected test methods of this document with equivalent test methods of IEC 62282-8-101 for solid oxide cell (SOC) operation for energy storage purposes, operated in reverse or reversible mode.

#### 2 Normative references

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

IEC 60050-485, *International Electrotechnical Vocabulary (IEV) – Part 485: Fuel cell technologies*, available at <https://www.electropedia.org>

IEC 60584-1, *Thermocouples – Part 1: EMF specifications and tolerances*

IEC 60584-3, *Thermocouples – Part 3: Extension and compensating cables – Tolerances and identification system*

IEC 61515, *Mineral insulated metal-sheathed thermocouple cables and thermocouples*

ISO 5168, *Measurement of fluid flow – Procedures for the evaluation of uncertainties*

ISO 6974 (all parts), *Natural gas – Determination of composition with defined uncertainty by gas chromatography*

ISO 7066-2, *Assessment of uncertainty in the calibration and use of flow measurement devices – Part 2: Non-linear calibration relationships*

ISO 8573-1, *Compressed air – Part 1: Contaminants and purity classes*

ISO 8756, *Air quality – Handling of temperature, pressure and humidity data*

ISO 12185, *Crude petroleum, petroleum products and related products – Determination of density – Laboratory density meter with an oscillating U-tube sensor*

### 3 Terms, definitions and symbols

#### 3.1 Terms and definitions

For the purposes of this document, the terms and definitions given in IEC 60050-485 and the following apply.

ISO and IEC maintain terminology databases for use in standardization at the following addresses:

- IEC Electropedia: available at <https://www.electropedia.org/>
- ISO Online browsing platform: available at <https://www.iso.org/obp>

##### 3.1.1

##### **cell/stack assembly unit**

unit including a single cell or stack, as well as gas supply parts, current collector parts, and any other peripherals used in power generation tests

##### 3.1.2

##### **active electrode area**

geometric electrode area upon which an electrochemical reaction occurs

Note 1 to entry: Usually the active electrode area is the smaller of the anode and cathode areas.

##### 3.1.3

##### **current density**

current divided by the active electrode area

##### 3.1.4

##### **average repeating unit voltage**

cell/stack assembly unit voltage divided by the number of the cells in a series connection in the unit

##### 3.1.5

##### **anode gas**

gas that is supplied to the inlet of the anode of a single cell/stack assembly unit

Note 1 to entry: Such a gas belongs to one of the following categories:

- a) pure hydrogen or mixture that contains hydrogen as a principal component with water vapour or nitrogen;
- b) reformed gas of raw fuel of SOFC such as methane or kerosene premixed with water vapour or air as oxidant;
- c) simulated gas of reformat that contains hydrogen, water vapour, carbon monoxide, carbon dioxide, methane, nitrogen, etc., as main components;
- d) methane, alcohols and other raw fuels directly supplied in pure form or mixed with water vapour or air, or both.
- e) condensable gas operating in gas phase such as anhydrous ammonia ( $\text{NH}_3$ ) as raw input fuel or in cracked form.

##### 3.1.6

##### **cathode gas**

gas that is supplied to the inlet of the cathode of a single cell/stack assembly unit

Note 1 to entry: Oxygen and nitrogen are its main components.

##### 3.1.7

##### **current collector**

conductive material in a cell/stack assembly unit that collects electrons from the anode side or conducts electrons to the cathode side

**3.1.8****stable state**

condition of a cell/stack assembly unit at which the unit is stable enough for any controlling parameter and the output voltage or output current of the unit to remain within its tolerance range of variation

**3.1.9****theoretical current**

current when the supplied anode gas or cathode gas is completely consumed in electrochemical reactions divided by the number of cells in a series connection

**3.1.10****effective fuel utilization**

ratio of the actual output current of the cell/stack assembly unit to the theoretical current that is calculated for the supplied fuel

Note 1 to entry: The effective utilization is the utilization of reactants in the electrochemical reaction at the anode due to the actual current. This can be less than the actual or total utilization if there are gas inlet and cross leaks.

Note 2 to entry: Causes of less-than-optimal currents include losses due to electronic conduction within the cell/stack assembly, gas leaks.

Note 3 to entry: A calculation method of effective fuel utilization is given in Annex B.

**3.1.11****effective oxygen utilization**

ratio of the actual output current of the cell/stack assembly unit to the theoretical current that is calculated for the supplied oxygen

Note 1 to entry: The effective utilization is the utilization of reactants in the electrochemical reaction at the cathode due to the actual current. This can be less than the actual or total utilization if there are gas inlet and cross leaks.

Note 2 to entry: A calculation method of effective oxygen utilization is given in Annex C.

**3.1.12****maximum effective fuel utilization**

highest effective fuel utilization that the cell/stack assembly unit can operate at, without causing unacceptable degradation

Note 1 to entry: The acceptable degradation rate is usually obtained from the developer.

**3.1.13****minimum cell/stack assembly unit voltage**

lowest cell/stack assembly unit voltage specified by the manufacturer

**3.1.14****open circuit voltage****OCV**

voltage across the terminals of a cell/stack assembly unit with cathode and anode gases present and in the absence of external current flow

Note 1 to entry: Also known as "no-load voltage".

**3.1.15****total impedance**

frequency-dependent losses due to ohmic, activation, diffusion, concentration effects, stray (parasitic) capacitance and inductances

**3.1.16****total resistance**

real part of the low-frequency limit of total impedance

**3.1.17****stoichiometric ratio**

ratio between the number of moles of reactant gas flowing per unit time to that used by the electrochemical reaction

Note 1 to entry: The terms, "stoichiometric ratio" and "reactant gas utilization," are related. The reciprocal of the fraction of the gas utilized is the stoichiometric ratio.

**3.2 Symbols**

Table 1 lists the symbols and units that are used in this document.

**Table 1 – Symbols**

Symbol	Term	Unit
$a$	Error limit specified from specification of instrument	<sup>a</sup>
$I$	Current	A
$J$	Current density	A/cm <sup>2</sup>
$A$	Active electrode area	cm <sup>2</sup>
$Z$	Total impedance	Ω cm <sup>2</sup>
$n$	Number of transferred electrons	
$N$	Number of cells in a series connection in the cell/stack assembly unit	
$p_a$	Absolute pressure of anode gas	kPa
$p_c$	Absolute pressure of cathode gas	kPa
$P$	Output power	W
$P_d$	Output power density	W/cm <sup>2</sup>
$q_a$	Flow rate of anode gas	l/min (STP <sup>b</sup> )
$q_c$	Flow rate of cathode gas	l/min (STP)
$q_j$	Flow rate of fuel component $j$ in anode gas	l/min (STP)
$t$	Time	s, min, h
$T_{op}$	Cell/stack assembly unit operating temperature	°C or K
$u_c$	Combined standard uncertainty for instruments	<sup>a</sup>
$u_{1,i}$	Standard uncertainty for instrument $i$	<sup>a</sup>
$U_f$	Effective fuel utilization	%
$U_{O_2}$	Effective oxygen utilization	%
$U_l$	Instrument expanded uncertainty	<sup>a</sup>
$V$	Voltage	V
$x_i$	Molar fraction of component $i$ or mole percent of component $i$	mol/mol or mol % <sup>c</sup>
$c_i$	Concentration of component $i$	mol/m <sup>3</sup>
$\xi_j$	Hydrocarbon conversion rate for hydrocarbon component $j$	%
<sup>a</sup> Denotes where the unit varies depending on the specification. <sup>b</sup> Abbreviation for standard temperature and pressure <sup>c</sup> Mole percent expressed as one hundred times mole fraction.		

## 4 General safety conditions

An operating fuel cell uses oxidizing and combustible gases. Typically, these gases are stored in high-pressure containers. In some cases, the fuel can be a toxic condensable gas (such as ammonia). The fuel cell itself can be operated at pressures greater than atmospheric pressure. Leaks or outlet flows from cell/stack assembly unit can contain toxic elements (e.g. when using ammonia as a fuel). Those who carry out cell/stack assembly unit testing shall be trained and experienced in the operation of test systems and specifically in safety procedures involving electrical equipment and reactive, compressed gases, and toxic compounds if applicable (e.g. when using ammonia as a fuel).

Materials which are compatible with the use and storage of the reactant gases shall be used during testing.

In summary, safely operating a test station requires appropriate technical training and experience as well as safety facilities and equipment, all of which are outside the scope of this document.

## 5 Cell/stack assembly unit

A cell/stack assembly unit includes a cell or stack, gas supply, current leads, and such other peripherals as required for power generation. It shall be provided with single or multiple measuring points for temperature and voltage, and one set of current lead points, all to be specified by the manufacturer.

As shown in Annex A, the boundary of a cell assembly unit goes through the anode gas supply port, cathode gas supply port, temperature, pressure measuring point, current lead points, voltage measuring points and mechanical load application points.

Some cell/stack assembly units can have no exhaust port for the anode gas or cathode gas because of the configuration of the cells. In such cases, the gas flow field pattern and its material shall be determined by the method recommended by the manufacturer. The load application method shall be also based on the recommendation of the manufacturer. The maximum operating temperature recommended by the manufacturer shall not be exceeded.

If the components of a cell/stack assembly unit other than a cell/stack are not specified by the manufacturer, the following shall be described in the test report, as a minimum:

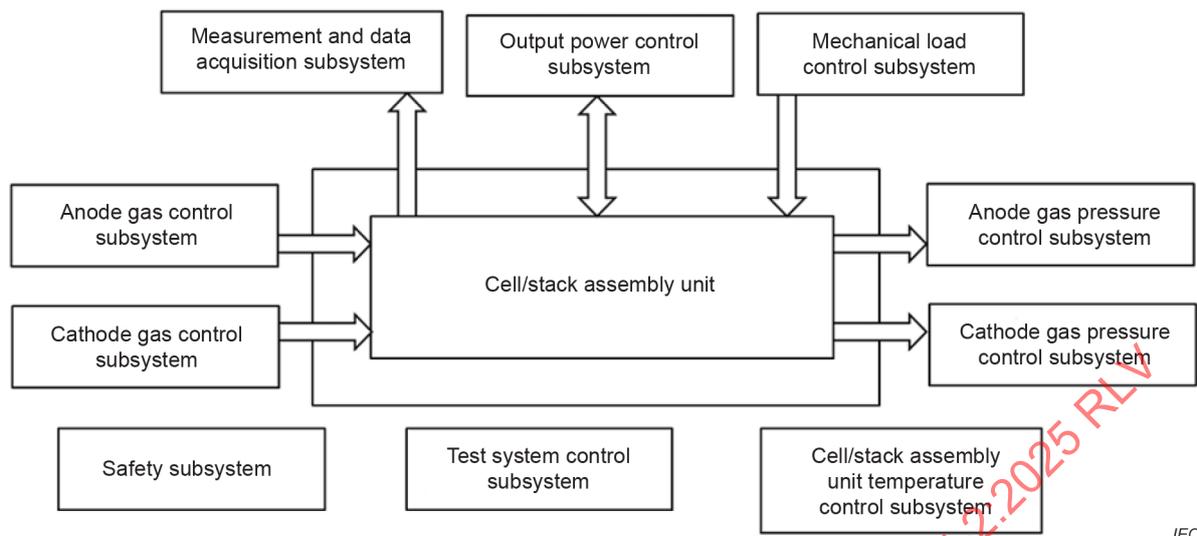
- a) materials and geometry of the peripheral components to be used for testing;
- b) flow patterns and directions of anode and cathode gases;
- c) locations of temperature measurement, mechanical load application, voltage measurement and current leads;
- d) magnitude of the mechanical load;
- e) configuration of assembly unit and its assembling method.

## 6 Testing system

### 6.1 Subsystems in testing system

#### 6.1.1 General

As shown in Figure 1, a testing system consists of an anode gas control subsystem, cathode gas control subsystem, cell/stack assembly unit temperature control subsystem, output power control subsystem, measurement and data acquisition subsystem and safety subsystem. It can also include a mechanical load control subsystem, anode gas and cathode gas pressure control subsystem or a test system control subsystem that controls the whole testing system, or both, if necessary.



**Figure 1 – Testing system**

### 6.1.2 Anode gas control subsystem

The anode gas control subsystem controls the flow rate, composition and temperature of the anode gas supplied to the cell/stack assembly unit. If the gas composition is to be maintained throughout the piping, then the materials, temperature, inner diameter and length of the piping shall be selected such as to ensure that any changes the gas composition can have within the piping are insignificant. Where necessary, the piping shall be heated or thermally insulated, or both in order to prevent condensation of water vapour.

Care should be taken to avoid other phenomena, such as carbon deposits, and the evaporation and transport of undesired materials in the gas streams, such as chromium species.

### 6.1.3 Cathode gas control subsystem

The cathode gas control subsystem controls the flow rate, composition and temperature of the cathode gas supplied to the cell/stack assembly unit.

### 6.1.4 Cell/stack assembly unit temperature control subsystem

The cell/stack assembly unit temperature control subsystem controls, at least, the electric furnace or the unit temperature. It maintains the operating temperature. The electric furnace shall be selected to maintain the temperature distribution within the specified tolerance level. Efforts should be made to minimize the electrical noise that the electric furnace generates while providing heat. It is assumed that all the test systems will use an electrical furnace for simplicity and safety reasons.

### 6.1.5 Output power control subsystem

The output power control subsystem controls the output current or output voltage of the cell/stack assembly unit.

### 6.1.6 Measurement and data acquisition subsystem

The measurement and data acquisition subsystem acquires and records the cell/stack assembly unit temperature, current, voltage, anode gas flow rate, cathode gas flow rate, and optionally, environmental conditions (ambient temperature, relative humidity, and atmospheric pressure) in accordance with the specified method. If necessary, it also acquires and records the mechanical load applied to the cell; the temperature, composition and pressure of the cathode gas and the anode gas; the flow rate, composition, temperature and pressure of anode and cathode exhaust gases; and cell/stack assembly unit impedance data, etc., in accordance with the specified method.

### 6.1.7 Safety subsystem

The safety subsystem functions as a detector and alarm system for malfunctioning of the test system based on predefined parameters and criteria. If it detects a serious fault, then it shall automatically establish a safe state in the test system. The anode should be purged with an inert gas, such as nitrogen, which can also contain hydrogen at concentrations below the lower flammability limit.

### 6.1.8 Mechanical load control subsystem

The optional mechanical load control subsystem regulates the mechanical load that is applied to increase the contact among components in the cell/stack assembly unit. The subsystem should be strong enough to apply the required mechanical load under the test conditions and to maintain the load for long term operation.

### 6.1.9 Gas pressure control subsystem for anode and cathode

The optional gas pressure control subsystem for anode and cathode gases regulates the pressure of these gases by the use of a back pressure control valve, etc.

### 6.1.10 Test system control subsystem

The test system control subsystem provides the integrated control for each control subsystem and data acquisition subsystem.

## 6.2 Maximum variation in control items of testing system

The tolerable variation of each control item in the testing system shall fall within the following ranges:

in the case of current control:	current: $\pm 1$ % relative to rated value point;
in the case of voltage control:	voltage: $\pm 1$ % relative to set point;
temperature:	$\pm 1,0$ % relative to set point;

NOTE 1 Temperature variation at the set point of less than  $\pm 5$  K will increase reproducibility.

anode and cathode gas flow rates:	$\pm 1$ % relative to rated;
anode gas composition:	$\pm 2,0$ mol % for $H_2$ , $N_2$ ;
	$\pm 2,0$ mol % for $CO$ , $CO_2$ , $CH_4$ , $NH_3$ ;
	$\pm 5,0$ mol % for $H_2O$ (water vapour concentration);

in case of bubbler or sparger humidification: dew point temperature:  $\pm 1$  °C;

NOTE 2 At water vapour concentrations greater than 10 mol %, a bubbler system (sparger) can cause higher uncertainty

cathode gas composition:  $\pm 1,0$  mol % of the target O<sub>2</sub> concentration;

where pressures of anode and cathode gases are to be controlled, pressures of anode and cathode gases:  $\pm 1$  % of rated condition, when pressure of rated condition is equal to or larger than 0,3 MPa; and 3 kPa, when pressure of rated condition is smaller than 0,3 MPa.

## 7 Instruments and measurement methods

### 7.1 General

Measuring instruments shall meet the requirement of 7.2. As a minimum, the flow rate and composition of the anode and cathode gases as well as the temperature, voltage, and current of the cell/stack assembly unit shall be measured. Additional measurements shall be taken based on the test parameters or test conditions, or both. It is possible that some of the following items specified in 7.3 or 7.4 will not be measurable in the case of a cell/stack assembly unit having no anode or cathode gas exhaust port.

### 7.2 Instrument uncertainty

The expanded uncertainty of each measuring instrument (coverage factor  $k = 2$ ) at the time of calibration or that estimated from the class of instrument shall meet the following requirements:

NOTE Coverage factor is defined in ISO/IEC Guide 98-3.

current:	$\pm 1$ % relative to rated;
voltage:	$\pm 0,5$ % relative to the open circuit voltage (OCV);
temperature:	$\pm 1,0$ % of reading;
flow rates of anode and cathode gases:	$\pm 2$ % of rated;
pressures of anode and cathode gases:	$\pm 1$ % of reading; average
anode gas composition:	$\pm 2$ mol % for H <sub>2</sub> , H <sub>2</sub> O, and N <sub>2</sub> ; $\pm 1$ mol % for CO, CO <sub>2</sub> , and CH <sub>4</sub> , NH <sub>3</sub> ;
cathode gas composition:	$\pm 0,3$ mol % for O <sub>2</sub> (balance N <sub>2</sub> ).

A method for determining instrument expanded uncertainty is given in Annex G.

### 7.3 Anode gas

#### 7.3.1 Anode gas flow rate

The anode gas flow rate shall be measured using mass flow meters, volumetric flow meters or turbine-type flow meters. The flow meter shall be selected by taking into consideration the species in the supplied gas, the range of flow rates, and the allowable uncertainty of the flow meter. When measurements are made on a volumetric basis, they shall be converted to mass flow rate by measuring the gas temperature and pressure or gas density in the vicinity of the flow meters. Measurement uncertainty for dry gases should be evaluated in accordance with ISO 5168 or ISO 7066-2.

#### 7.3.2 Anode gas composition

The anode gas composition should be measured when the performance of the cell/stack assembly unit is measured. If this is not possible, however, the anode gas composition shall be measured during the preparation of the performance test under the same conditions as those of the cell performance test. See 10.6.2.2.

When anode gas is supplied in one of the following conditions a) to d) below, and if the gas supply line has no reactors, such as a reformer, and is confirmed to insignificantly change the gas composition, composition can be calculated based on the composition table published by the fuel supplier and values obtained from each flow meter, in accordance with ISO 6145-7:

- a) a single-composition gas such as hydrogen is supplied;
- b) a mixed gas of known composition is supplied;
- c) anode gas is supplied by mixing component gases in a controlled manner using multiple flow meters;
- d) gases under b) and c) above are supplied in combination.

The anode gas shall be sampled near the anode gas supply port of the cell/stack assembly unit and analysed using an infrared spectrometer, mass spectrometer, gas chromatograph or similar device. The gas sample shall be transported from its origin to the point of analysis in a manner which minimizes changes in composition. Thus, the material, temperature, diameter and the length of the tubing shall be carefully chosen to minimize the compositional change in the sampling tubing. When necessary, it shall be heated to avoid the condensation of the water vapour.

If water vapour is likely to affect measurement, remove water from the gas sample or dilute the gas sample with argon or a similar inert gas.

The result of such analysis for gas component  $i$ , expressed as  $c_i$  (mol/m<sup>3</sup>) shall be normalized to obtain a normalized concentration,  $x_i$  (mol/mol), using the following equation:

$$x_i = \frac{c_i}{\sum_i c_i} \quad (1)$$

where  $\sum_i c_i$  represents the sum of concentrations of all component gases in the analysis.

The gas analyser shall be calibrated using a standard gas of known mass ratio.

The measurement uncertainty shall be evaluated in accordance with the ISO 6974 series.

### 7.3.3 Anode gas temperature

The gas temperature shall be measured near the anode gas supply port of the cell/stack assembly unit by using a thermocouple of a type and class in accordance with IEC 60584-1 or sheathed thermocouple of a type and class in accordance with IEC 61515 and an extension lead wire of a type and class in accordance with IEC 60584-3. When there is a reactor such as a reformer, the gas temperature at the exit of the reactor should also be measured.

NOTE There can be significant differences between the temperature of the tube wall and the temperature of the bulk gas.

If it is difficult to measure the gas temperature during the cell performance test, the anode gas temperature shall be measured during the preparation of the performance test under the same conditions as those of the performance test.

#### 7.3.4 Anode gas pressure

The anode gas pressure shall be measured upstream of the anode gas supply port of the cell/stack assembly unit by using a calibrated pressure sensor, manometer, Bourdon tube or similar instrument. The measuring instrument shall be located in such a manner that the uncertainty is minimized in consideration of any pressure loss within the piping, piping temperature and other factors. Condensation of water vapour during measurement shall be prevented. One way can be to measure the pressure by injecting a very small amount of dry nitrogen gas or similar into the pipe, close to the measuring instrument.

#### 7.3.5 Anode exhaust gas flow rate

The anode exhaust gas flow rate shall be measured using mass flow meters, volumetric flow meters or turbine-type flow meters after implementing a means to prevent water condensation from affecting the stability of anode gas flow or after removing water from the gas flow. When measurements are made on a volumetric basis, they shall be converted to mass flow rate by measuring the gas temperature and pressure or gas density in the vicinity of the flow meters. Alternatively, the anode exhaust gas flow rate can be calculated from the component concentrations of the anode exhaust gas, tracer concentration and tracer flow rate by precisely adding a minute amount of a gas that is not contained in the anode exhaust gas as the tracer. The gas analyser shall be calibrated using a standard gas of known mass ratio. Measurement uncertainty shall be evaluated in accordance with the ISO 6974 series.

The exhaust gas shall be handled with caution for reasons of safety and the environment, since it can still contain hydrogen, carbon monoxide and hydrocarbons.

#### 7.3.6 Anode exhaust gas component

The anode exhaust gas shall be sampled near the anode gas exhaust port of the cell/stack assembly unit. See 10.6.2.2. The sample shall be analysed using an infrared spectrophotometer, mass spectrometer, gas chromatograph or similar device. If water vapour is likely to affect the measurement, remove water from the gas sample or dilute the sample with argon gas or similar. When measuring, the materials, temperature, inner diameter and length of piping shall be selected such as to ensure that any changes the gas composition can have within the piping are insignificant. In particular, the piping shall be heated where necessary to prevent water vapour from condensing in the piping. The gas analyser shall be calibrated using a standard gas of known mass ratio.

#### 7.3.7 Anode exhaust gas temperature

The gas temperature shall be measured near the anode gas exhaust port of the cell/stack assembly unit by selecting a thermocouple of a type and class in accordance with IEC 60584-1 or sheathed thermocouple of a type and class in accordance with IEC 61515, and an extension lead wire of the type and class IEC 60584-3. If it is difficult to measure the gas temperature during the cell performance test, the anode exhaust gas temperature shall be measured during the preparation of the performance test under the same conditions as those of the performance test.

NOTE There can be significant differences between the temperature of the tube wall and the temperature of the bulk gas.

#### 7.3.8 Anode exhaust gas pressure

The anode exhaust gas pressure shall be measured downstream of the anode gas exhaust port of the cell/stack assembly unit by using a pressure sensor, manometer, Bourdon tube or similar device. The measuring instrument should be located in such a manner that the uncertainty is minimized in consideration of any pressure loss within the piping, piping (gas) temperature and other factors. Condensation of water vapour during measurement shall be prevented. One way can be to measure the pressure by injecting a very small amount of dry nitrogen gas or similar into the pipe close to the measuring instrument.

## 7.4 Cathode gas

### 7.4.1 Cathode gas flow rate

The cathode gas flow rate shall be measured by using mass flow meters, volumetric flow meters or turbine-type flow meters. When measurements are made on a volumetric basis, they shall be converted to mass flow rate by measuring the gas temperature and pressure or gas density in the vicinity of the flow meters. The flow meter shall be selected in consideration of the expected range of flow rates and the allowable uncertainty of the flow meter. Usually, uncertainty shall be evaluated in accordance with ISO 5168, and if there is non-linearity, evaluated in accordance with ISO 7066-2.

### 7.4.2 Cathode gas component

For cathode gas composition, the oxygen concentration shall be measured using a gas chromatograph or an oxygen concentration meter. The cathode gas should consist of clean (oil-free), compressed air or bottled gas. If a bottled gas mixture is used, the values described on its composition certificate published by the fuel supplier can be used. The uncertainty of the instrument shall be evaluated in accordance with the ISO 6974 series.

When it is necessary to measure humidity, a dew point meter, water content meter or gas chromatograph shall be used while controlling the gas temperature to prevent condensation of water vapour.

### 7.4.3 Cathode gas temperature

The gas temperature shall be measured near the cathode gas supply port of the cell/stack assembly unit by selecting a thermocouple of a type and class in accordance with IEC 60584-1 or sheathed thermocouple of a type and class in accordance with IEC 61515, and an extension lead wire of the type and class IEC 60584-3.

NOTE There can be significant differences between the temperature of the tube wall and the temperature of the bulk gas.

If it is difficult to measure the gas temperature during the cell performance test, the cathode gas temperature shall be measured during the preparation of the performance test under the same conditions as those of the performance test.

### 7.4.4 Cathode gas pressure

The cathode gas pressure shall be measured upstream of the cathode gas supply port of the cell/stack assembly unit by using a pressure sensor, manometer, Bourdon tube, or similar device. The measuring instrument should be located in such a manner that the uncertainty is minimized in consideration of any pressure loss within the piping, piping temperature, and other factors.

### 7.4.5 Cathode exhaust gas flow rate

The cathode exhaust gas flow rate shall be measured using a mass flow meter, volumetric flow meter or turbine-type flow meter after cooling the gas. When measurements are made on a volumetric basis, they shall be converted to mass flow rate by measuring the gas temperature and pressure or gas density in the vicinity of the flow meter. The flow meter shall be selected in consideration of the expected range of flow rates and the allowable uncertainty of the instrument. The uncertainty of the instrument shall be evaluated in accordance with the ISO 6974 series.

#### **7.4.6 Cathode exhaust gas component**

For cathode exhaust gas composition, the oxygen concentration shall be measured using a gas chromatograph or an oxygen concentration meter after cooling the gas. When it is necessary to measure an extremely low water concentration, a dew point meter, water content meter or gas chromatograph shall be used while controlling the gas temperature to prevent condensation of water vapour.

#### **7.4.7 Cathode exhaust gas temperature**

The cathode exhaust gas temperature shall be measured near the cathode gas exhaust port of the cell/stack assembly unit by selecting a thermocouple of a type and class in accordance with IEC 60584-1 or sheathed thermocouple of a type and class in accordance with IEC 61515, and an extension lead wire of the type and class IEC 60584-3. If it is difficult to measure the gas temperature during the cell performance test, the cathode exhaust gas temperature shall be measured during the preparation of the performance test under the same conditions as those of the performance test.

NOTE There can be significant differences between the temperature of the tube wall and the temperature of the bulk gas.

#### **7.4.8 Cathode exhaust gas pressure**

The cathode exhaust gas pressure shall be measured downstream of the cathode gas exhaust port of the cell/stack assembly unit by using a pressure sensor, manometer, Bourdon tube or similar device. The measuring instrument should be located in such a manner that the uncertainty is minimized in consideration of any pressure loss within the piping, piping temperature and other factors.

#### **7.5 Output voltage**

A voltage meter shall be connected to the voltage measuring points, as described in Clause 5. The voltage thus measured shall be deemed to be the voltage of the cell/stack. The connecting cable shall be durable enough for the test conditions.

#### **7.6 Output current**

A galvanostat or electric load connected to the current lead points, as described in Clause 5, or a current sensor, or both, such as a shunt resistor located within the current circuit, shall be used to measure the current by sending its output to a measuring or recording instrument. The connecting cable shall be selected for appropriate materials and geometry in consideration of the test conditions and possible voltage drop within the cable.

#### **7.7 Cell/stack assembly unit temperature**

A thermocouple of a type and class in accordance with IEC 60584-1 or sheathed thermocouple of a type and class in accordance with IEC 61515, and an extension lead wire of the type and class in accordance with IEC 60584-3 shall be selected. They shall be placed at the temperature measuring point as described in Clause 5 and connected with a recorder or similar device for measurement. When there is more than one temperature measuring point, the unit temperature and its distribution shall be obtained by the calculation method recommended by the manufacturer.

#### **7.8 Mechanical load**

A mechanical load applied as recommended by the manufacturer shall be measured.

## 7.9 Total impedance

The total impedance of the cell/stack assembly unit shall be measured by either the alternating current (AC) impedance method or the current interruption method. An appropriate measuring line shall be used in order to ensure high-quality data over the entire frequency range investigated.

## 7.10 Ambient conditions

In defining the ambient conditions, ambient temperature, pressure and relative humidity shall be measured. The sampling interval shall be the value specified in ISO 8756 or less.

# 8 Test preparation

## 8.1 General

The type of cell/stack assembly unit to be tested, the number of samples, test parameters, and test conditions shall be determined.

Each measuring instrument shall be checked for its last calibration, the uncertainty under the calibration conditions, or estimated from the class of the instrument, and its dependency on the environmental conditions in order to estimate the uncertainty of the instrument. The method and cycle of calibration and replacement shall be designed to ensure that there is no increase in measurement uncertainty.

The components of the anode and cathode gases and their main impurities shall be verified. As described in Clause 7, a preliminary test shall be performed for gas composition and temperature in order to ensure that the gas compositions are established within the anticipated uncertainty and that the supply gas temperature does not affect the unit temperature. Further, the test procedure, test conditions, and judging criteria for stable state, among others, shall be determined based on the preliminary test results and other factors.

## 8.2 Standard test conditions and test range

The standard test conditions and the typical test range that are recommended by the manufacturer shall be reviewed for the following parameters in order to determine the test conditions and range:

- a) cell/stack assembly unit temperature;
- b) allowable cell/stack assembly unit temperature distribution (if multiple measuring points);
- c) anode gas flow rate;
- d) anode gas composition;
- e) anode gas pressure;
- f) cathode gas flow rate;
- g) cathode gas composition;
- h) cathode gas pressure;
- i) effective fuel utilization;
- j) effective oxygen utilization;
- k) current or current density;
- l) minimum cell/stack assembly unit voltage;
- m) minimum cell/stack assembly unit current (under a constant effective fuel utilization, see Annex E for more information);

- n) maximum cell/stack assembly unit current (under a constant effective fuel utilization). Damage due to excessive degradation is possible beyond this value;
- o) mechanical load.

### **8.3 Components and impurities of anode gas and cathode gas**

If gases are used to prepare the anode gas, purity level or components and major impurities of each gas shall be verified by the composition tables published by the respective fuel suppliers or through analysis. When the anode gas is produced from liquid fuel, its density, its carbon, hydrogen, and oxygen content, and content of impurities, such as sulfur, shall be verified by the composition table published by the fuel supplier or through analysis in accordance with ISO 12185.

The purity or components and major impurities of the cathode gas shall be verified by the composition table published by the fuel supplier or through analysis. If a compressor is used, the compressed air shall be free of oil and particles in accordance with ISO 8573-1.

The result of each verification or analysis shall be described in the test report.

### **8.4 Basis of the test procedure**

The start-up conditions, such as heating rate and ambient conditions during the heating ramp, the condition of the anode (i.e. the extent of reduction of the nickel oxide to nickel), and the shutdown conditions, such as cooling rate and ambient conditions during the cooling ramp, shall be based on those recommended by the manufacturer or the results of preliminary tests.

### **8.5 Confirmation of aging conditions of unit**

The aging conditions of the cell/stack assembly unit shall be determined based on the aging conditions recommended by the manufacturer, as well as the preliminary tests to be conducted, to ensure that the output drift at the time of measurement is insignificant.

### **8.6 Confirmation of criteria of stable state**

The tolerance level of variation shall be determined for the output current or output voltage of the cell/stack assembly unit, and the assessment criteria of stable state shall be determined through preliminary testing and others.

The judgement criteria of stable state shall be described in the test report.

### **8.7 Data acquisition method**

Preliminary tests shall be conducted while taking into consideration the variation of each test parameter and the sampling rate of each measuring instrument, amongst other things, to determine the sampling interval and the number of samplings and measurements. The sampling interval (e.g. 1 s) shall be short enough to observe the variation of the measured parameter with sufficient time resolution. The number of samplings and repetitions for a single measurement shall be decided so that the total measurement period becomes sufficiently longer than the dominant variation cycle of each test parameter.

## 9 Test procedure

### 9.1 Set-up

The test set-up procedure shall be as follows:

- a) Check each control or measurement subsystem for possible leakage. There are many methods for leak-checking, such as pressure hold and helium leak detectors. The choice of method will depend on the equipment in use. The proper operation of the test equipment should be verified by comparing its performance with the parameters specified in 7.2.
- b) Prepare a cell/stack assembly unit consisting of cell(s), gas passage, interconnectors, current collectors, insulation and other components in accordance with the assembly method and procedure recommended by the manufacturer. Before connecting the cell/stack to the test bench, measure the resistance between cathode and anode current lead points to determine if there is a short-circuit. Measure the resistance between cell voltage measuring points to determine if they are electrically insulated. Measure the cell-to-cell resistances. These values should not indicate a short-circuit, but, rather, should be similar to those specified by the manufacturer.
- c) Set up the cell/stack assembly unit in a temperature control subsystem and install the wiring for voltage measurement and current leads, the mechanical load, and thermocouples as well as the piping for the gas supply and exhaust. Connect the wires to their corresponding subsystems. Ensure proper insulation between thermocouples and the cell/stack. There should also be electrical insulation between the mechanical load and the cell/stack.
- d) Check gas pipe connections for leakage (see 9.1.a).
- e) If necessary, verify the wiring for insulation to earth. It is recommended to check the insulation before the output control subsystem or measurement subsystem is connected. In addition, proper wiring shall be verified at joint connections.
- f) When the above are all completed, the measurement subsystem is checked for its proper operation.

### 9.2 Initial conditioning

The cell/stack assembly unit shall be started up at the temperature increasing rate and ambient conditions as specified in 8.4 and operated until it reaches the stable state after going through anode reduction and conditioning.

### 9.3 Shutdown

The shutdown procedure shall be initiated at the specified temperature decreasing rate and ambient conditions as specified in 8.4. The temperature shall be decreased under such conditions as the user has determined, based on preliminary tests or as directed by the manufacturer, unless otherwise provided. During this time, the air flow to the air electrode is maintained and hydrogen diluted with nitrogen (or other inert gas) is flowing to the fuel electrode. The concentration of hydrogen in this gas mixture shall be below the lower explosive limit.

## 10 Performance test

### 10.1 Rated power test

#### 10.1.1 Objective

The objective of this test is to measure and verify the output of the cell/stack assembly unit under rated conditions.

### 10.1.2 Test method

All control parameters shall be set at rated conditions, and after the cell/stack assembly unit has reached the stable state, the voltage, current and other control parameters shall be measured repeatedly at a sampling interval until the number of samples and measurements are obtained as specified in 8.7. The average value of the measurements shall be the measured value. Optionally, include the standard deviation of the measurements.

### 10.1.3 Presentation of results

The measurement results shall be used to calculate the rated power output and recorded in the test report with voltage, current and other measurements of the test conditions.

## 10.2 Current-voltage characteristics test

### 10.2.1 Objective

The objective of this test is to determine the dependency of current-voltage ( $I-V$ ) characteristics on temperature, pressure, gas composition, gas flow rate or effective gas utilization.

### 10.2.2 Test method

#### 10.2.2.1 Test under constant flow rate

The control parameter on which the dependency is to be measured shall be set at its initial value while the anode gas and cathode gas flow rates as well as other control parameters shall be set at those of the test conditions. The unit shall be operated until the stable state is reached under open-circuit conditions, and current-voltage characteristics are measured by changing the current or voltage stepwise or sweeping it at a constant speed. After the measurement, the control parameter is set to the next value and the measurement shall be repeated within the measuring range specified in 8.2.

- a) When current is step-changed, the cell/stack assembly unit shall be operated until it reaches the stable state in each step (temperature and voltage), and at each step measurements are taken over the duration of time at the sampling intervals as specified in 8.7. The average value of the measurements after the stable state is reached shall be the measured value for that step. Optionally, include the standard deviation of the measurements.
- b) When a current sweep is used, the sweep rate shall be determined such that the maximum width of the voltage hysteresis does not exceed the voltage variation in the stable state.
- c) When voltage control is used, step a) or b) shall be taken with step voltage or voltage sweep, respectively.

NOTE The meaning of the maximum width of the voltage hysteresis is explained in Annex D.

#### 10.2.2.2 Test under constant effective fuel or oxygen utilization, or both, or constant stoichiometric ratio

The control parameter on which the dependency is to be measured shall be set at its initial value while effective fuel utilization or effective oxygen utilization, or both, and other control parameters shall be set at those of the test operating conditions given in 8.2. The unit shall be operated at the minimum current specified by the manufacturer until it reaches the stable state, and current-voltage characteristics shall be measured by changing the current or voltage stepwise. The unit shall be operated in each step until it reaches the stable state with measurements being taken over the duration of time at the sampling rate as specified in 8.7. The average value of the measurements after the stable state is reached shall be the measured value for that step. Optionally, include the standard deviation of the measurements. After the measurement, the control parameter is set to the next value and the measurement shall be repeated within the measuring range specified in 8.2.

An example of the record of  $I-V$  characteristics test under constant effective fuel utilization is given in Annex E.

### 10.2.3 Presentation of results

The results shall be expressed in a two-dimensional plot with its horizontal axis representing current density or effective fuel utilization, or a combination of current density and fuel utilization, and its vertical axis representing cell/stack assembly unit voltage. This plot shall be included in the test report with the other test conditions. Alternatively to the stack voltage, the measured cell voltages or the average repeating unit voltage can be plotted against current density. Optionally, include the standard deviation of the measurements.

## 10.3 Effective fuel utilization dependency test

### 10.3.1 Objective

The objective of this test is to study the dependency of the performance of the cell/stack assembly unit on effective fuel utilization and to confirm the maximum effective fuel utilization under various operating conditions. A method for the calculation of effective fuel utilization is described in Annex B.

### 10.3.2 Test method

#### 10.3.2.1 General

The maximum fuel utilization as well as the conditions used to obtain this value shall be obtained from the manufacturer or determined through consultation between the manufacturer and the evaluator.

The test shall be conducted either by decreasing the anode gas flow rate at constant current or by increasing the current at constant anode gas flow rate.

#### 10.3.2.2 Test at constant current

The following steps shall be carried out:

- a) Set the cell/stack assembly unit at the test conditions as specified in 8.2, operate it and verify that it has reached the stable state.
- b) Decrease the anode gas flow rate stepwise until the cell/stack assembly unit reaches the maximum effective fuel utilization as specified by the manufacturer. For each step, verify that the voltage has reached the stable state and record it.
- c) When the fuel utilization reaches the maximum effective value, return the anode gas flow rate stepwise to the original value. For each step, verify that the voltage has reached the stable state and record it. The step sizes will or will possibly not match the decreases from the previous step. Comparing this voltage to that before this step provides information about whether the maximum effective fuel utilization specified by the manufacturer which can be achieved by the cell/stack assembly unit.

#### 10.3.2.3 Test at constant anode gas flow rate

The following steps shall be carried out:

- a) Set the cell/stack assembly unit at the specified test conditions, operate it and verify that it has reached the stable state.
- b) Increase the current stepwise until the cell/stack assembly unit reaches the maximum effective fuel utilization. For each step, verify that the voltage has reached the stable state and record it.
- c) When the fuel utilization reaches the maximum effective value, return the current to the original value stepwise and record the voltage. Comparing this voltage to that before this step provides information about the maximum effective fuel utilization which can be achieved by the cell/stack assembly unit, if only for a short period of time. This is different to that seen in 10.4.

The criteria for suspending the test should be determined in advance either by preliminary testing or by consulting with the manufacturer in order to prevent any performance degradation or damage to the cell/stack assembly unit.

### 10.3.3 Presentation of results

The results shall be expressed in a two-dimensional plot with its horizontal axis representing effective fuel utilization, or a combination of anode gas flow rate and effective fuel utilization in the case of 10.3.2.2, and effective fuel utilization, or a combination of current density and effective fuel utilization in the case of 10.3.2.3, and its vertical axis representing the cell/stack assembly unit voltage. Optionally, include the standard deviation of the measurements. This plot shall be included in the test report with the other test conditions.

The stack assembly unit voltage can be replaced by the average repeating unit voltage. Alternatively, the measured cell voltages can be plotted against current density.

## 10.4 Long term durability test

### 10.4.1 Objective

The objective of this test is to evaluate the performance degradation of the cell/stack assembly unit when it is exposed to certain test conditions over a long period of time and to examine the effect of temperature, current, gas composition, gas impurities and other factors on the durability of the cell/stack assembly unit.

### 10.4.2 Test method

#### 10.4.2.1 General

This test shall be conducted by maintaining the test conditions constant for the duration of the test, either measuring change in the cell/stack assembly unit voltage at constant current or measuring change in the unit voltage together with total resistance. The total resistance shall be measured at intervals below 10 % of the total duration of the durability study by the  $I$ - $V$  characteristics in 10.2 or the impedance spectrum in 10.7.

NOTE Other operating modes such as constant voltage with constant gas utilization (constant efficiency operation with decreasing power) or constant power operation and constant gas utilization (decreases efficiency at constant power) or some mixed mode operation, can be used to measure the durability as specified by the test request.

#### 10.4.2.2 Voltage change in long term durability test

Set up all the controlling parameters at the specified test conditions and measure the voltage of the cell/stack assembly unit at regular intervals. The measured values shall be used to obtain the rate of voltage change for the entire test period or a specified duration.

#### 10.4.2.3 Total resistance change in long term durability test

This test can be performed at the same time as the test carried out in accordance with 10.4.2.2.

The following method shall be used for the measurement of total resistance:

- a) Measure whole  $I$ - $V$  characteristics between 0 to maximum current as specified in 8.2, or measure partial  $I$ - $V$  characteristics in the vicinity of the holding current, both in a similar manner to that stated in 10.2.
- b) Derive the approximate tangent to the  $I$ - $V$  characteristics at holding current by connecting two points on the  $I$ - $V$  characteristics in the vicinity across the holding current and find the slope of the tangent. Report the slope as total resistance.

The points in the vicinity of the holding current are selected such that they are closer to the holding current when the curvature of the  $I$ - $V$  characteristics is large. The uncertainty in the voltage and current measurements should be kept as small as possible.

- c) After measuring the  $I$ - $V$  characteristics, restore the values to those of the original test conditions and measure voltage until the next measuring cycle.
- d) Repeat this measurement at a certain interval throughout the test duration.

The results shall be used to calculate the voltage variation rate and total resistance variation rate for the entire test period or specific time duration within the test period.

It is also possible to measure the total impedance described in 10.7 at the time of total resistance measurement.

#### 10.4.3 Presentation of results

In the case of 10.4.2.2, the results shall be expressed in a two-dimensional plot with the horizontal axis representing time and the vertical axis representing voltage and shall be included in the test report with the voltage change rate and test conditions. In the case of 10.4.2.3, the vertical axis shall represent voltage and total resistance, and the plot shall be included in the test report along with voltage change rate, total resistance variation rate and test conditions.

The stack assembly unit voltage can be plotted against time.

NOTE These results can also be represented as area-specific resistance (or impedance). Here, the area-specific resistance (impedance) equals the measured resistance (impedance) multiplied by the active electrode area.

### 10.5 Thermal cycling durability test

#### 10.5.1 Objective

The objective of this test is to evaluate the durability of the cell/stack assembly unit with thermal cycling. The thermal cycles shall be within the manufacturer's specifications.

#### 10.5.2 Test method

##### 10.5.2.1 General

For this test, the following test conditions shall be obtained from the manufacturer or determined through consultation between the manufacturer and the evaluator in advance.

The operating temperature shall be the temperature of the standard test conditions.

- a) number of thermal cycles;
- b) cooling rate;
- c) heating rate;
- d) maximum temperature;
- e) minimum temperature;
- f) operating conditions at operating temperature;
- g) period to maintain operating temperature;
- h) period to maintain minimum temperature;
- i) total test period;
- j) gas flow rate and composition at heating and cooling conditions, and minimum temperature, respectively.

### 10.5.2.2 Test procedure

In accordance with the above test conditions, either the method of measuring cell/stack assembly unit voltage variation at the operating temperature or measuring total resistance together with cell/stack assembly unit voltage shall be chosen. When measuring total resistance, follow the method of 10.4.2.3. The voltage shall be measured over the specified duration. After a certain operating period, the temperature of the cell/stack assembly unit shall be decreased to the minimum temperature with the cell/stack assembly unit at open circuit under the specified conditions, and the minimum temperature shall be maintained for the specified time. The temperature of the cell/stack assembly unit shall then be raised under the specified conditions to the operating temperature and the measurement shall resume as before.

These measurements shall be repeated until the specified number of thermal cycles is reached. The results obtained shall be used to calculate the voltage variation rate and total resistance variation rate over the entire test period or specific duration within the test period.

### 10.5.3 Presentation of results

The results shall be expressed in a two-dimensional plot with the horizontal axis representing either (a) time or (b) number of cycles. The test conditions shall also be described in the test report.

In the case (a), unit voltage, and temperature shall be plotted on the vertical axis. Total resistance shall be plotted if applicable.

In the case (b), unit voltage shall be plotted on the vertical axis. Total resistance shall be plotted if applicable.

## 10.6 Internal reforming performance test

### 10.6.1 Objective

The objective of this test is to evaluate the internal reforming (or cracking) performance of the cell/stack assembly unit under the open-circuit conditions or rated conditions against hydrocarbons (HC) such as methane or ammonia (NH<sub>3</sub>) contained in the anode gas.

### 10.6.2 Test method

#### 10.6.2.1 General

This test shall be performed in accordance with the manufacturer's recommendation or after consultation between the manufacturer and the evaluator regarding the cell/stack's ability to reform HCs (or crack NH<sub>3</sub>) internally; it is possible that some cell/stack assembly units will not be able to do so.

NOTE Many complex reactions can occur with HC-containing anode gases, depending on composition and thermodynamic equilibria. These reactions can affect the temperature and pressure gradients in the cell and stack.

This test is applicable when anode gas and anode exhaust gas can be sampled without any mixing with cathode gas or cathode exhaust gas.

#### 10.6.2.2 Test procedure

Anode gas containing HCs (or NH<sub>3</sub>) shall be supplied to the cell/stack assembly unit. After the unit reaches the stable state in the open circuit or rated conditions, the anode gas and anode exhaust gas shall be analysed for their compositions; these are used to calculate the HC (or NH<sub>3</sub>) conversion rate that indicates the internal reforming (or cracking) characteristic.

The HC (or NH<sub>3</sub>) conversion rate for a specific HC<sub>j</sub> (NH<sub>3j</sub>),  $\xi_j$  (%) is calculated as follows:

$$\xi_j = 100 (q_{\text{HC},j,\text{in}} - q_{\text{HC},j,\text{out}}) / q_{\text{HC},j,\text{in}} \quad (2)$$

$$\xi_j = 100 (q_{\text{NH}_3,j,\text{in}} - q_{\text{NH}_3,j,\text{out}}) / q_{\text{NH}_3,j,\text{in}} \quad (3)$$

where  $q_{\text{HC},j,\text{in}}$  ( $q_{\text{NH}_3,j,\text{in}}$ ) and  $q_{\text{HC},j,\text{out}}$  ( $q_{\text{NH}_3,j,\text{out}}$ ) represent the flow rates of the specific HC<sub>j</sub> (NH<sub>3j</sub>) at the anode inlet and outlet, respectively, which are calculated from the anode gas flow rate and its HC (NH<sub>3</sub>) concentration, and the anode exhaust gas flow rate and its HC (NH<sub>3</sub>) concentration, respectively.

### 10.6.3 Presentation of results

The compositions and flow rates of anode gas and anode exhaust gas as well as the HC (or NH<sub>3</sub>) conversion rate shall be described in the test report with the test conditions.

## 10.7 Resistance components identification test

### 10.7.1 Objective

The main objective of this test is to identify and evaluate ohmic and non-ohmic components of the total resistance of the cell/stack assembly unit.

### 10.7.2 Test method

#### 10.7.2.1 General

The separation of resistance components of the cell/stack assembly unit shall be evaluated by either the AC impedance method or the current interruption method.

#### 10.7.2.2 AC electrochemical impedance method

The following test conditions shall be determined in advance by conducting preliminary tests.

- Measuring range for frequencies:

The highest frequency should roughly identify point A, and the lowest frequency should roughly identify point C when plotted in the complex impedance diagram (see Figure 2).

- Number of measuring points:

Four to twenty points per one order of frequencies (to be distributed evenly as logarithms, if possible) are required; they shall be numerous enough to identify clearly the geometry of impedance plots. If possible, avoid the fundamental and harmonics of the electrical grid frequency.

The test shall be conducted using the following procedure:

- a) establish the test conditions;
- b) verify that the stable state has been reached;
- c) superimpose AC sinusoidal waves on DC current or voltage and start measurements. Sweep the AC sinusoidal waves within the specified frequency range and measure the impedance at each frequency.

The amplitude of the AC signal for the measurement shall be enough to activate the cell but not overly polarize it. The amplitude per cell can be obtained by dividing the total voltage amplitude with the number of cells in series. As an option, the validity of the impedance spectrum shall be verified by using appropriate validation relations such as the Kramers-Kronig (KK) relationships or Z-hit.

### 10.7.2.3 Current interruption method

When employing this method, the current interruption characteristics and sampling rate to allow the evaluation of a target property of the cell/stack assembly unit shall be identified by preliminary testing. Measurements shall be taken after ensuring that the unit is in the stable state under the test conditions.

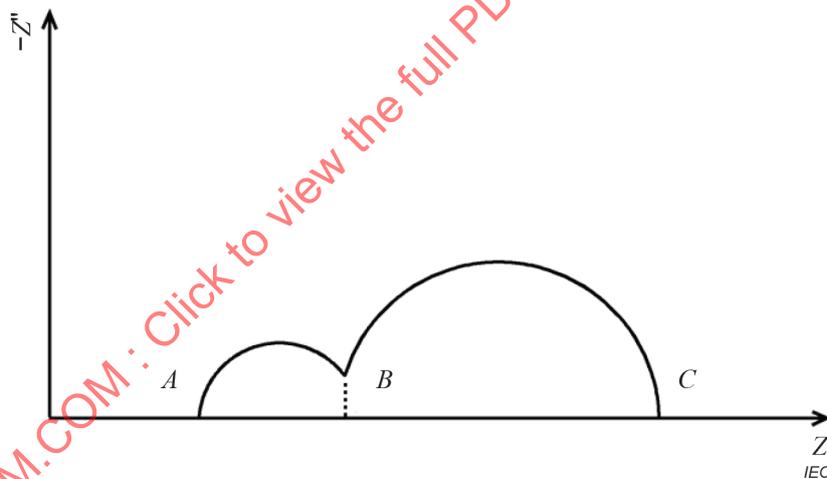
### 10.7.3 Presentation of results

#### a) AC impedance method

The test results shall be expressed as a complex impedance or Cole-Cole plot (indicate characteristic frequencies) or Bode plot (a plot of impedance components against the logarithm of measured frequency), or both. The impedance times unit area calculated by active area defined in 3.1.2 shall be plotted.

#### b) Current interruption method

The current (density) and voltage response waveforms observed before and after current interruption shall be plotted against the time axis. The sampling rate shall be appropriate to identify the ohmic resistance component. The ohmic resistance component so obtained shall be reported with the test conditions.



#### Key

- A* high-frequency end impedance
- C* low-frequency end impedance
- A-B* high-frequency arc impedance
- B-C* low-frequency arc impedance
- Z'* real part of impedance
- Z''* negative imaginary part of impedance

**Figure 2 – Typical diagram of complex impedance plot for SOFC**

NOTE 1 These results can also be represented as area-specific resistance (or impedance). Here, the area-specific resistance (impedance) equals the measured resistance (impedance) multiplied by the active electrode area.

The data collection interval shall be selected to be able to define the current switch-off point accurately within 1  $\mu$ s. High frequency interference that increases the error shall be cared in determining the ohmic resistance.

## 11 Test report

### 11.1 General

Test reports shall accurately, clearly and objectively present sufficient information to demonstrate that all the objectives of the tests have been attained. A suggested template for the test report is given in Annex F.

A controlled document policy should be used within an engineering change control process.

Test reports can indicate the test procedure document revision that is used for the test and can detail any agreed amendments to the released test procedure in an annexure.

The test report can be archived in electronic form.

### 11.2 Report items

The report shall present the following information, at a minimum:

- a) title of the report;
- b) authors of the report;
- c) date of the report;
- d) test report reference or identification number;
- e) location and (start) date and time of the test;
- f) test bench used;
- g) test unit data (see 11.3 for details);
- h) test conditions (see 11.4 for details);
- i) test data (see 11.5 for details).

### 11.3 Test unit data description

Test unit data shall include the following information, at a minimum:

- a) product name and brand name of the unit;
- b) active electrode area;
- c) number of cells (total, series, parallel);
- d) cell materials and thicknesses, if known, and cell identification number(s);
- e) stacking materials, if known;
- f) geometry of the unit;
- g) temperature measurement and load application positions.

### 11.4 Test conditions description

The test conditions description shall include the following information, at a minimum:

- a) name of person(s) and entity conducting the test;
- b) instruments and calibration record;
- c) test procedure;
- d) aging conditions;
- e) criteria of stable state;
- f) data acquisition method;
- g) gas purity and impurities;

h) test bench layout.

### 11.5 Test data description

Test data shall include the following information:

- a) title of the test(s);
- b) test operating conditions;
- c) test result;
- d) ambient conditions;
- e) uncertainty evaluation (see 11.6 for details).

### 11.6 Uncertainty evaluation

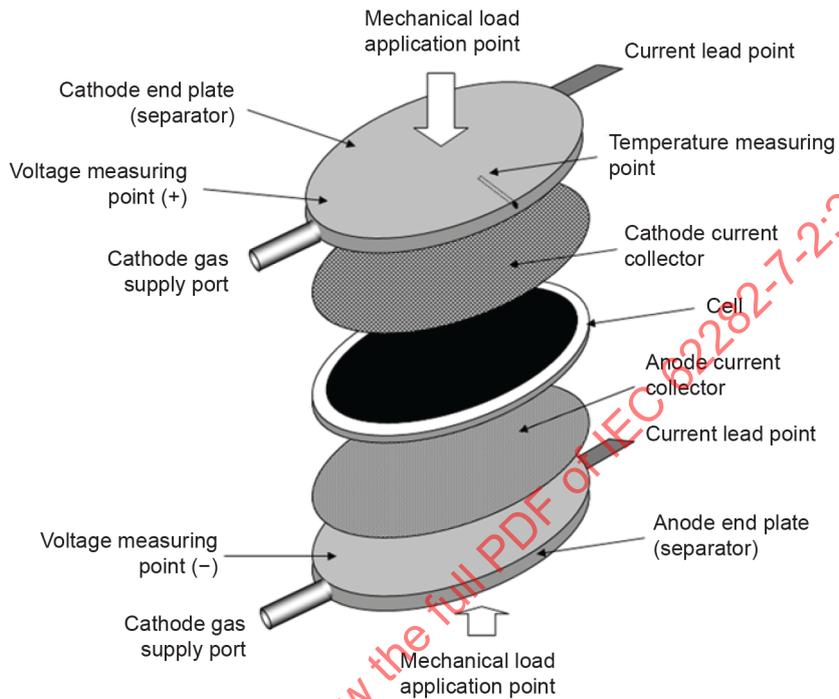
Uncertainties of instruments shall be reported. If necessary, variation of measurements or measurement uncertainties, or both, calculated from the variation of measurements and uncertainties of instruments should be reported.

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## Annex A (informative)

### Example of cell assembly unit

An example configuration and test boundary of a cell assembly unit described in this document is shown in Figure A.1.



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**Figure A.1 – Example of cell assembly unit**

In this schematic example, the cathode current collector and anode current collector also work as cathode gas flow channel and anode gas flow channel, respectively. The exhaust gas comes out from the respective circumference of each current collector.

NOTE It is possible to measure the voltage at the current collector to exclude the voltage drop due to contact resistances (plate-current collector).

## Annex B (informative)

### Calculation of effective fuel utilization

#### B.1 General

Annex B describes a method for the calculation of effective fuel utilization as defined in 3.1.10.

#### B.2 Calculation method

In a performance test, an anode gas is supplied at a rate of  $q_a$  (l/min (STP)). The flow rate of each fuel component in the anode gas is expressed as  $q_j$  (l/min (STP)) ( $j = \text{H}_2, \text{CO}, \text{CH}_4, \dots, \text{C}_p\text{H}_q\text{O}_r, \text{NH}_3$ ) where  $\text{C}_p\text{H}_q\text{O}_r$  is the chemical formula of a general hydrocarbon fuel. In the case where the anode gas composition is analysed,  $q_j$  shall be calculated from the molar fraction of each fuel component ( $x_j$  (mol/mol)) and  $q_a$  using Equation (B.1):

$$q_j = x_j \times q_a \quad (\text{B.1})$$

In the cell/stack assembly unit, it is assumed that  $N$  cells are connected in series and that the fuel is uniformly distributed between the cells. A theoretical current defined in 3.1.9,  $I_{\text{theory}}$  (A), assuming that the supplied fuel gas is completely consumed in electrochemical reactions, shall be calculated from Equation (B.2):

$$I_{\text{theory}} = \frac{P_{\text{st}}}{R \times T_{\text{st}} \times 60 \times 1\,000} \times F \times \left[ \frac{\sum_j n_j \times q_j}{N} \right] \quad (\text{B.2})$$

$$= \frac{101\,325}{8,314\,51 \times 273,15 \times 60 \times 1\,000} \times 96\,485 \times \left[ \frac{\sum_j n_j \times q_j}{N} \right] = 71,74 \times \left[ \frac{\sum_j n_j \times q_j}{N} \right]$$

where

$P_{\text{st}}$  is the standard pressure (101 325 Nm<sup>-2</sup>);

$T_{\text{st}}$  is the standard temperature (273,15 K);

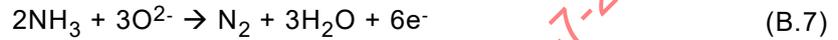
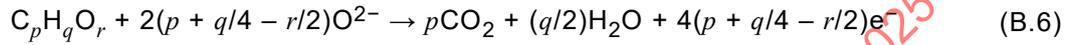
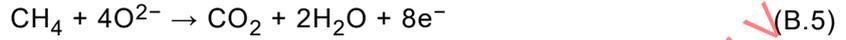
$R$  is the universal gas constant (8,314 4 Jmol<sup>-1</sup>K<sup>-1</sup>);

$F$  is Faraday's constant (96 485 C/mol);

$n_j$  is the number of electrons transferred when one molecule of fuel component  $j$  is electrochemically reacted;

$N$  is the number of cells in series.

The electron transferred,  $n_j$  for representative fuels, is determined by Equation (B.3) to Equation (B.6) and is summarized in Table B.1. For the general  $C_pH_qO_r$  component,  $n_j$  is equal to  $4(p + q/4 - r/2)$ :



The measured current output of each cell or that from the total cell/stack assembly unit is expressed as  $I_{\text{measured}}$ . Therefore, effective fuel utilization, or  $U_f$  (%) can be calculated from Equation (B.8):

$$U_f = \frac{I_{\text{measured}}}{I_{\text{theory}}} \times 100 \% \quad (B.8)$$

**Table B.1. –  $n_j$  for representative fuels**

Fuel	$n_j$
H <sub>2</sub>	2
CO	2
CH <sub>4</sub>	8
C <sub>p</sub> H <sub>q</sub> O <sub>r</sub>	4(p + q/4 - r/2)
NH <sub>3</sub>	3

### B.3 Calculation examples

#### B.3.1 Calculation from anode gas composition and flow rate

Normalized anode gas mole fraction,  $x_i$ , is assumed to be as indicated in Table B.2 as a result of anode gas composition analysis. It is presumed that anode gas flow rate,  $q_a$ , is 0,500 l/min (STP). The flow rates,  $q_j$ , of H<sub>2</sub>, CO and CH<sub>4</sub>, which are fuel components in the anode gas, are calculated using Equation (B.1) as 0,281, 0,047, 0,003 l/min (STP), respectively. Then,  $\sum_j n_j q_j$

is calculated by summing up each  $n_j \times q_j$ , leading to  $\sum_j n_j q_j = 0,562 + 0,094 + 0,024$

= 0,680 l/min (STP). Hence, if  $N = 10$  cells,  $I_{\text{theory}} = 71,74 \times 0,680/10 = 4,88$  A is obtained using Equation (B.2). If it is assumed that the actual output current of the stack is 3,90 A,  $I_{\text{measured}}$  is equal to 3,90 A. Therefore, effective fuel utilization can be calculated using Equation (B.8) as

$$U_f = \frac{3,90}{4,88} \times 100 = 80 \text{ \%}.$$

**Table B.2 – Anode gas composition, flow rate of each fuel component  $q_j$ , and  $n_j q_j$**

Component	$x_i$ mol %	$q_j$ l/min (STP)	$n_j q_j$ l/min (STP)
H <sub>2</sub>	56,1	$56,1/100 \times 0,500 = 0,281$	$2 \times 0,281 = 0,562$
H <sub>2</sub> O	27,1		
CO	9,3	$9,3/100 \times 0,500 = 0,047$	$2 \times 0,047 = 0,094$
CO <sub>2</sub>	7,1		
CH <sub>4</sub>	0,5	$0,5/100 \times 0,500 = 0,003$	$8 \times 0,003 = 0,024$

#### B.3.2 Calculation from supplied H<sub>2</sub> and H<sub>2</sub>O flow rate

It is assumed that H<sub>2</sub> and H<sub>2</sub>O are supplied to the anode by controlling each flow rate. It is also assumed that in a performance test of a 40-cell-stack in which the number of series connection is 10 with 4 parallel connections, H<sub>2</sub> flow rate,  $q_j$ , and output current are to be equal to 3,00 l/min (STP) and 32,3 A, respectively. Using Equation (B.2),  $I_{\text{theory}}$  is calculated as  $71,74 \times (2 \times 3,00) / 10 = 43,0$  A. Therefore, with  $I_{\text{measured}} = 32,3$  A, effective fuel utilization can be calculated as  $U_f = \frac{32,3}{43,0} \times 100 = 75,1 \text{ \%}$  from Equation (B.8).

NOTE The number of parallel connections in the stack does not make any difference to the calculation.

## Annex C (informative)

### Calculation of effective oxygen utilization

#### C.1 General

Annex C describes a method for the calculation of effective oxygen utilization as defined in 3.1.11.

#### C.2 Calculation method

In a performance test, a cathode gas is supplied at a rate of  $q_c$  (l/min (STP)). Oxygen flow rate ( $q_{O_2}$  (l/min (STP))) shall be calculated from the oxygen molar fraction in the cathode gas, or  $x_{O_2}$  (mol/mol), using Equation (C.1):

$$q_{O_2} = x_{O_2} \times q_c \quad (C.1)$$

The theoretical current defined in 3.1.9,  $I_{\text{theory}}$  (A), assuming that the cathode gas is uniformly distributed among  $N$  cells connected in series in the stack and that the cathode gas is completely consumed in electrochemical reactions, shall be calculated from Equation (C.2):

$$\begin{aligned} I_{\text{theory}} &= \frac{P_{\text{st}}}{R \times T_{\text{st}} \times 60 \times 1\,000} \times F \times \frac{n_{O_2} \times q_{O_2}}{N} \\ &= \frac{101\,325}{8,314\,51 \times 273,15 \times 60 \times 1\,000} \times 96\,485 \times \frac{n_{O_2} \times q_{O_2}}{N} = 287,0 \times \frac{q_{O_2}}{N} \end{aligned} \quad (C.2)$$

where

$P_{\text{st}}$  is the standard pressure (101 325 Nm<sup>-2</sup>);

$T_{\text{st}}$  is the standard temperature (273,15 K);

$R$  is the universal gas constant (8,314 4 Jmol<sup>-1</sup>K<sup>-1</sup>);

$F$  is Faraday's constant (96 485 C/mol);

$n_{O_2}$  is the number of electrons transferred when one molecule of oxygen is electrochemically reduced, leading to  $n_{O_2} = 4$  as shown in Equation (C.3);

$N$  is the number of cells connected in series:



The measured current output of each cell or that from the total cell/stack assembly unit is expressed as  $I_{\text{measured}}$ . Therefore, effective oxygen utilization, or  $U_{O_2}$  (%) shall be calculated from Equation (C.4):

$$U_{O_2} = \frac{I_{\text{measured}}}{I_{\text{theory}}} \times 100 \% \quad (C.4)$$

### C.3 Calculation example

In a performance test, it is assumed that the cathode gas flow rate,  $f_c$ , is 1,50 l/min (STP), and that there are  $N = 10$  cells connected in series in the stack with the gas composition as indicated in Table C.1.  $f_{O_2} = 0,314$  l/min (STP) and  $I_{\text{theory}} = 9,01$  A can be obtained using Equation (C.1) and Equation (C.2), respectively. When the actual output current at the stack performance test is 2,70 A,  $I_{\text{measured}}$  is equal to 2,70 A. Therefore, effective oxygen utilization can be calculated as  $U_{O_2} = 2,70/9,01 \times 100 = 30$  % using Equation (C.4).

**Table C.1 – Cathode gas composition,  $q_{O_2}$ , and  $I_{\text{theory}}$**

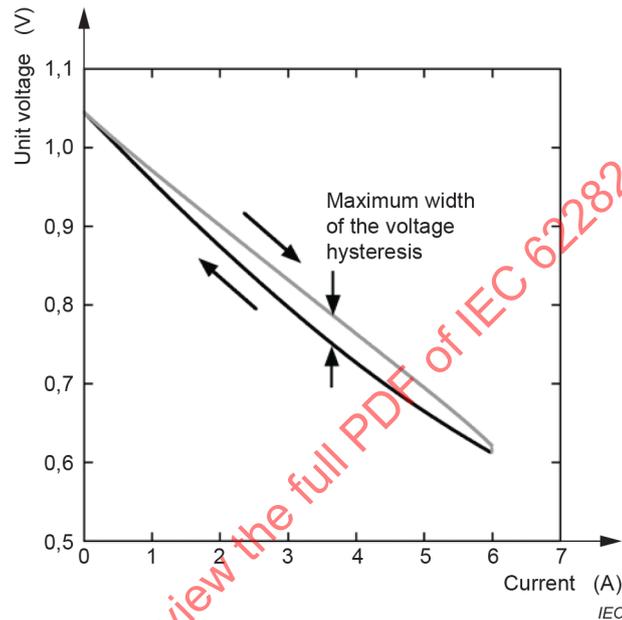
Component	$x_i$ mol %	$q_{O_2}$ l/min (STP)	$I_{\text{theory}}$ A
O <sub>2</sub>	20,95	$20,95/100 \times 1,50 = 0,314$	$287,0 \times 0,314/10 = 9,01$
N <sub>2</sub>	79,05		

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### Annex D (informative)

#### Maximum width of the voltage hysteresis in $I$ - $V$ characteristics test

When  $I$ - $V$  characteristics are taken with the current sweep method described in 10.2.2.1 b), measured voltages can be different with a different sweep rate due to hysteresis, as shown in the example in Figure D.1. The appropriate sweep rate shall be determined in such a way that the maximum width of the voltage hysteresis is smaller than the allowable maximum variation of voltage that is defined in 7.2.



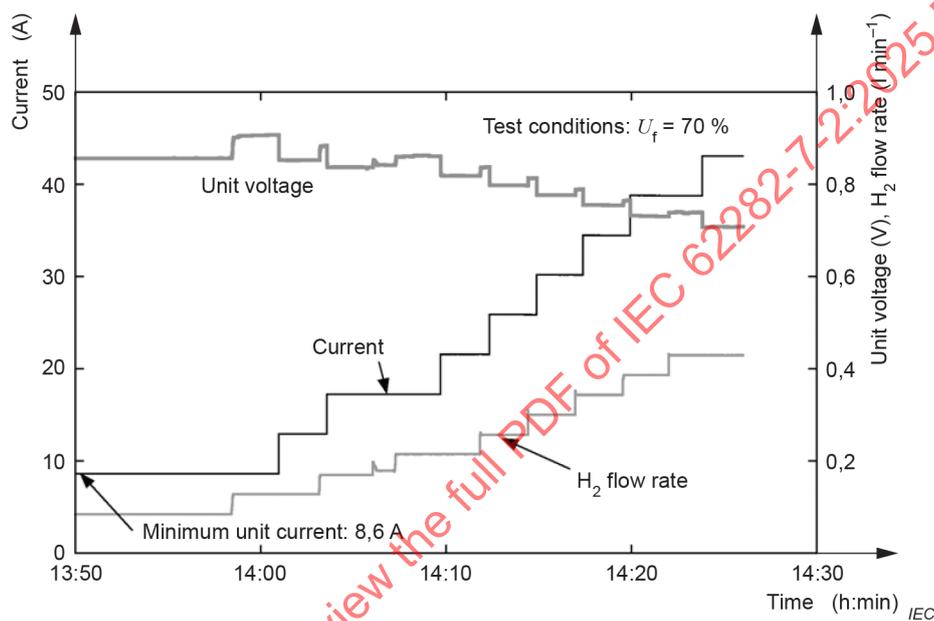
NOTE In this Figure D.1, the right pointing arrow relates to increasing current that is from OCV to maximum current and the left pointing arrow relates to decreasing current that is from maximum current to OCV.

Figure D.1 – Voltage hysteresis at a given sweep rate in  $I$ - $V$  characteristics test

## Annex E (informative)

### Current-voltage characteristics test under constant effective fuel utilization

In order to keep the effective fuel or oxygen utilization constant, or both in the measurement range, fuel or oxygen flow rates, or both are changed as the current changes. Dependency of unit voltage on current accordingly differs from those observed under a constant gas flow rate. An example of such changes among the unit voltage, current and hydrogen flow rate is shown in Figure E.1.



**Figure E.1 – Example of the record in current-voltage characteristics test under constant effective fuel utilization at increasing steps in current**

When  $I$ - $V$  characteristics are taken under a constant effective fuel utilization, the unit shall initially be kept at the minimum cell/stack assembly unit current as defined in 8.2 m).

At low current, voltage becomes unstable due to low gas flow rate. Therefore, the minimum cell/stack assembly unit current should be determined so as to avoid such unstable voltage.

**Annex F**  
(informative)

**Test report (template)**

**F.1 Overview**

Examples of a report for general information, test unit data description and test conditions as well as a test report for each test specified in the body text are given below. Instructions to the author are given in *italics* and should not be included in the test report. The method for determining "instrument uncertainty" shown in Clause F.5, Clause F.6, Clause F.7, Clause F.8 and Clause F.9 is given in Annex G.

**F.2 General information**

Test report title	
Author(s) of report	
Date of report	
Test report reference or identification number	
Location of test	
Start date and time of test	
Test bench	

**F.3 Test unit data description**

Product name and brand name of the unit	
Active electrode area	
Number of cells (total, series, parallel)	
Unit identification number	
Geometry of the unit	
Materials and thickness of electrolyte and electrodes, interconnect	<i>If known</i>

*If the following are not available from the manufacturer, they shall be reported.*

Configuration of assembly unit and assembling method	
Materials and geometry of the peripheral components	
Flow patterns and directions of anode and cathode gases	
Temperature measurement positions	
Mechanical load (unit) and its application positions	
Voltage measurement positions	
Current lead positions	
Minimum cell voltage (unit)	

#### F.4 Test conditions

Name of person(s) and entity conducting the test	
Instruments and calibration record	
Test procedure	
Aging conditions	
Criterion or criteria of stable state	
Data acquisition method	
Gas purity and impurities	
Mechanical load	

#### F.5 Rated power test

Operating conditions

Input	Value (unit)	Instrument uncertainty (unit)	Variation of measurement (Standard deviation (combined)) (unit)
$q_a$			
$q_c$			
$p_a$			
$p_c$			
$T_{op}$			

Test results

Output	Value (unit)	Instrument uncertainty (unit)	Variation of measurement (Standard deviation (combined)) (unit)
OCV			
$V$			
$I$			
$P$			

The data average method shall be described.

#### F.6 Current-voltage characteristics test

Operating conditions

Input	Value (unit)
$q_a$ (or its range in the case of 10.2.2.2)	
$q_c$ (or its range in the case of 10.2.2.2)	
$p_a$	
$p_c$	
$T_{op}$	
$U_f$ (in the case of 10.2.2.2)	
$U_{02}$ (in the case of 10.2.2.2)	

Test results

Output	Value (unit)	Instrument uncertainty (unit)	Variation of measurement (Standard deviation) (unit)
OCV			

The  $I$ - $V$  characteristics shall be presented (see 10.2.3).

### F.7 Effective fuel utilization dependency test

Initial operating conditions

Input	Value (unit)
$q_a$	
$q_c$	
$p_a$	
$p_c$	
$T_{op}$	
$I$	

Preliminary information

Expected maximum fuel utilization	/ %
-----------------------------------	-----

Test results

Output	Value (unit)	Instrument uncertainty (unit)	Variation of measurement (Standard deviation) (unit)
OCV			

In the case of 10.3.2.2

Step number	$f_a$ (unit)	$V$ (unit)	$U_f$ / %
0			
1			
2			
3			
$m$			

The  $U_f(f_a)$ - $V$  curve that can replace the above table (see 10.3.3) shall be presented.

In the case of 10.3.2.3

Step number	$I$ (unit)	$V$ (unit)	$U_f I$ %
0			
1			
2			
3			
$m$			

The  $U_f(I) - V$  curve that can replace the above table (see 10.3.3) shall be presented.

## F.8 Long-term durability test

Operating conditions

Input	Value (unit)
$q_a$	
$U_f$	
$q_c$	
$U_{O_2}$	
$p_a$	
$p_c$	
$T_{op}$	
$I$	

Test results

Output	Value (unit)	Instrument uncertainty (unit)	Variation of measurement (Standard deviation) (unit)
OCV (start)			
OCV (end)			
$i$ (end)			

The other test results should be presented by means of a two-dimensional plot (see 10.4.3).

### F.9 Thermal cycling durability test

#### Test conditions

Cooling rate	
Heating rate	
Minimum temperature	
Period to maintain operating temperature	
Period to maintain minimum temperature	
Gas composition during heating	
Gas flow rate during heating	
Gas composition during cooling	
Gas flow rate during cooling	
Gas composition at minimum temperature	
Gas flow rate at minimum temperature	

#### Operating conditions

Input	Value (unit)
$q_a$	
$q_c$	
$p_a$	
$p_c$	
$T_{op}$	
$I$	

#### Test results

Output	Value (unit)	Instrument uncertainty (unit)	Variation of measurement (standard deviation) (unit)
OCV (start)			
OCV (end)			

The other results should be presented as specified in 10.5.3.

## F.10 Internal reforming performance test

Operating conditions

Input	Value (unit)
$q_a$	
$q_c$	
$U_f$	
$p_a$	
$p_c$	
$T_{op}$	
$I$	

Test results

Output	Value (unit)
OCV	
Composition of anode exhaust gas	
Flow rate of anode exhaust gas	
HC conversion rate (Optional)	/ %

## F.11 Resistance components identification test

Operating conditions

Input	Value (unit)
$q_a$	
$q_c$	
$p_a$	
$p_c$	
$T_{op}$	
$I$ or $V$	
Frequency range	
Operating mode (galvanostatic or potentiostatic)	
Amplitude	

Test results

Output	Value (unit)
Total resistance	
Ohmic resistance	

Depending upon the method, corresponding figures shall be attached (see 10.7.3).

## Annex G (informative)

### Method for determining instrument expanded uncertainty

Instrument expanded uncertainty,  $U_1$ , can be obtained by the calibrations using traceable standard instruments. See ISO/IEC Guide 98-3 for further information.

It can also be obtained from the error limit ( $\pm a$ ) of instrument as shown in Formula (G.1) assuming uniform distribution of the probability within the error limit range:

$$U_1 = 2u_c = 2 \frac{a}{\sqrt{3}} \quad (\text{G.1})$$

where  $u_c$  is the standard instrument uncertainty.

Some of the measurements are made using several instruments (e.g. for current measurement, the combination of current sensor and digital voltage recorder can be used). The standard uncertainty for such a case can be obtained as shown in Formula (G.2) assuming no correlation between the instruments:

$$u_c^2 = \sum_j \left( \frac{\partial f}{\partial x_j} \right)^2 u_{i,j}^2 \quad (\text{G.2})$$

where

$f$  (e.g. power as the product of current and voltage) is the functional relation of the instruments (current sensor and digital voltage recorder) to determine the derived quantity (power);

$x_j$  is the parameter (current or voltage) measured by the  $j^{\text{th}}$  instrument;

$u_{i,j}$  represents standard instrument uncertainty of the  $j^{\text{th}}$  instrument.

Therefore,

$$U_1 = 2u_c = 2 \left( \sum_j \left( \frac{\partial f}{\partial x_j} \right)^2 u_{i,j}^2 \right)^{1/2} \quad (\text{G.3})$$

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## COMMISSION ÉLECTROTECHNIQUE INTERNATIONALE

## TECHNOLOGIES DES PILES À COMBUSTIBLE –

**Partie 7-2: Méthodes d'essai – Essais de performance de cellule élémentaire et de pile pour les piles à combustible à oxyde solide (SOFC)**

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Cette deuxième édition annule et remplace la première édition parue en 2021. Cette édition constitue une révision technique.

Cette édition inclut les modifications techniques majeures suivantes par rapport à l'édition précédente:

- a) le Tableau 1 a été révisé afin de spécifier les unités manquantes pour certaines caractéristiques;
- b) des ressources ont été ajoutées dans la Bibliographie (l'ISO/TR 15916, les modules d'essai définis par le projet SOCTESQA et le Guide ISO/IEC 98-6:2021) afin de fournir de plus amples informations.

Le texte de cette Norme internationale est issu des documents suivants:

Projet	Rapport de vote
105/1093/FDIS	105/1099/RVD

Le rapport de vote indiqué dans le tableau ci-dessus donne toute information sur le vote ayant abouti à son approbation.

La langue employée pour l'élaboration de cette Norme internationale est l'anglais.

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## INTRODUCTION

Il existe une grande variété de formes et de tailles de piles à combustible à oxyde solide (SOFC). De ce fait, en général, les périphériques tels que les collecteurs de courant et les collecteurs de gaz sont uniques à chaque cellule ou à chaque pile et sont souvent incorporés à une cellule ou à une pile afin de constituer une entité intégrée. De plus, ils ont tendance à avoir un impact significatif sur les caractéristiques de production de puissance de la cellule ou de la pile. Le présent document présente par conséquent, comme sujet d'étude, des "entités d'assemblage de cellules/piles", définies comme des entités qui contiennent non seulement une cellule ou une pile, mais également des périphériques.

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## TECHNOLOGIES DES PILES À COMBUSTIBLE –

### Partie 7-2: Méthodes d'essai – Essais de performance de cellule élémentaire et de pile pour les piles à combustible à oxyde solide (SOFC)

#### 1 Domaine d'application

La présente partie de l'IEC 62282 s'applique aux entités d'assemblage de cellules/piles SOFC, aux systèmes d'essai, aux appareils et aux méthodes de mesure, et spécifie des méthodes d'essai afin de vérifier les performances des cellules et des piles SOFC.

Le présent document ne s'applique pas aux petites piles boutons qui sont conçues pour les essais de matériaux SOFC et ne prévoient aucun moyen pratique de mesure de l'utilisation de combustible.

Le présent document est utilisé en réponse à la recommandation de l'entité qui fournit la spécification des performances des cellules ou pour l'acquisition de données sur une cellule ou une pile, afin d'estimer les performances d'un système qui serait basé sur celle-ci. Parmi les éléments d'essai décrits dans le présent document, les utilisateurs du présent document peuvent choisir d'exécuter ceux qui sont pertinents pour les applications visées.

Les utilisateurs peuvent remplacer les méthodes d'essai choisies dans le présent document par les méthodes d'essai équivalentes de l'IEC 62282-8-101, concernant l'utilisation de piles à oxyde solide (SOC, Solid Oxide Cell) à des fins de stockage de l'énergie, en mode inversé ou réversible.

#### 2 Références normatives

Les documents suivants sont cités dans le texte de sorte qu'ils constituent, pour tout ou partie de leur contenu, des exigences du présent document. Pour les références datées, seule l'édition citée s'applique. Pour les références non datées, la dernière édition du document de référence s'applique (y compris les éventuels amendements).

IEC 60050-485, *Vocabulaire Électrotechnique International (IEV) – Partie 485: Technologies des piles à combustible*, disponible à l'adresse <https://www.electropedia.org>

IEC 60584-1, *Couples thermoélectriques – Partie 1: Spécifications et tolérances en matière de FEM*

IEC 60584-3, *Couples thermoélectriques – Partie 3: Câbles d'extension et de compensation – Tolérances et système d'identification*

IEC 61515, *Câbles et couples thermoélectriques à isolation minérale dits "chemisés"*

ISO 5168, *Mesure de débit des fluides – Procédures pour le calcul de l'incertitude*

ISO 6974 (toutes les parties), *Gaz naturel – Détermination de la composition et de l'incertitude associée par chromatographie en phase gazeuse*

ISO 7066-2, *Évaluation de l'incertitude dans l'étalonnage et l'utilisation des appareils de mesure du débit – Partie 2: Relations d'étalonnage non linéaires*

ISO 8573-1, *Air comprimé – Partie 1: Polluants et classes de pureté*

ISO 8756, *Qualité de l'air – Traitement des données de température, de pression et d'humidité*

ISO 12185, *Pétroles bruts, produits pétroliers et produits connexes – Détermination de la masse volumique – Appareil de masse volumique de laboratoire à capteur à tube en U oscillant*

### 3 Termes, définitions et symboles

#### 3.1 Termes et définitions

Pour les besoins du présent document, les termes et les définitions de l'IEC 60050-485 ainsi que les suivants s'appliquent.

L'ISO et l'IEC tiennent à jour des bases de données terminologiques destinées à être utilisées en normalisation, consultables aux adresses suivantes:

- IEC Electropedia: disponible à l'adresse <https://www.electropedia.org/>
- ISO Online browsing platform: disponible à l'adresse <https://www.iso.org/obp>

##### 3.1.1

#### **entité d'assemblage de cellules/piles**

entité comprenant une cellule simple ou une pile, ainsi que des éléments d'alimentation en gaz, des éléments de collecteur de courant et d'autres périphériques éventuels utilisés pour les essais de production de puissance

##### 3.1.2

#### **surface d'électrode active**

surface d'électrode géométrique sur laquelle se produit une réaction électrochimique

Note 1 à l'article: Habituellement, la surface d'électrode active correspond à la plus petite des surfaces entre celle de l'anode et celle de la cathode.

##### 3.1.3

#### **densité de courant**

courant divisé par la surface d'électrode active

##### 3.1.4

#### **tension moyenne de l'entité répétée**

tension d'entité d'assemblage de cellules/piles divisée par le nombre de cellules dans une connexion série de l'entité

##### 3.1.5

#### **gaz d'anode**

gaz fourni à l'admission de l'anode d'une entité d'assemblage de cellules simples/piles

Note 1 à l'article: Ce type de gaz appartient à l'une des catégories suivantes:

- a) hydrogène pur ou mélange qui contient de l'hydrogène comme composant principal avec de la vapeur d'eau ou de l'azote;
- b) gaz reformé de combustible brut de SOFC tel que du méthane ou du kérosène prémélangé avec de la vapeur d'eau ou de l'air comme oxydant;
- c) gaz simulé de gaz reformé qui contient de l'hydrogène, de la vapeur d'eau, du monoxyde de carbone, du dioxyde de carbone, du méthane, de l'azote, etc., comme composants principaux;
- d) méthane, alcools et autres combustibles bruts fournis directement sous forme pure ou mélangés avec de la vapeur d'eau ou de l'air, ou les deux.
- e) gaz condensable exploité dans sa phase gazeuse, tel que l'ammoniac ( $\text{NH}_3$ ), sous forme de combustible d'entrée brut ou sous sa forme craquée.

**3.1.6****gaz de cathode**

gaz fourni à l'admission de la cathode d'une entité d'assemblage de cellules simples/piles

Note 1 à l'article: L'oxygène et l'azote sont ses principaux composants.

**3.1.7****collecteur de courant**

matériau conducteur dans une entité d'assemblage de cellules/piles qui collecte les électrons du côté anode ou les conduit vers le côté cathode

**3.1.8****état stable**

état d'une entité d'assemblage de cellules/piles auquel l'entité est suffisamment stable pour que tout paramètre de régulation et la tension ou le courant de sortie de l'entité restent dans les limites de tolérance de leur plage de variation

**3.1.9****courant théorique**

courant existant lorsque le gaz d'anode ou de cathode fourni est entièrement consommé dans des réactions électrochimiques, divisé par le nombre de cellules dans une connexion série

**3.1.10****utilisation de combustible efficace**

rapport du courant de sortie réel de l'entité d'assemblage de cellules/piles sur le courant théorique qui a été calculé pour le combustible fourni

Note 1 à l'article: L'utilisation efficace est l'utilisation de réactants dans la réaction électrochimique à l'anode due au courant réel. Cette utilisation peut être moindre que l'utilisation réelle ou totale en cas d'entrée de gaz et de fuites croisées.

Note 2 à l'article: Les causes de génération de courants inférieurs aux courants optimaux incluent les pertes dues à une conduction électronique au sein de l'assemblage de cellules/piles, et les fuites de gaz.

Note 3 à l'article: Une méthode de calcul de l'utilisation de combustible efficace est donnée à l'Annexe B.

**3.1.11****utilisation d'oxygène efficace**

rapport du courant de sortie réel de l'entité d'assemblage de cellules/piles sur le courant théorique qui a été calculé pour l'oxygène fourni

Note 1 à l'article: L'utilisation efficace est l'utilisation de réactants dans la réaction électrochimique à la cathode due au courant réel. Cette utilisation peut être moindre que l'utilisation réelle ou totale en cas d'entrée de gaz et de fuites croisées.

Note 2 à l'article: Une méthode de calcul de l'utilisation d'oxygène efficace est donnée à l'Annexe C.

**3.1.12****utilisation de combustible efficace maximale**

utilisation de combustible efficace la plus élevée avec laquelle l'entité d'assemblage de cellules/piles peut fonctionner sans provoquer de dégradation inacceptable

Note 1 à l'article: Le taux de dégradation acceptable est généralement obtenu auprès du développeur.

**3.1.13****tension minimale d'entité d'assemblage de cellules/piles**

tension d'entité d'assemblage de cellules/piles la plus basse spécifiée par le fabricant

**3.1.14**  
**tension en circuit ouvert**  
**OCV**

tension aux bornes d'une entité d'assemblage de cellules/piles en présence de gaz de cathode et d'anode et en l'absence de courant extérieur

Note 1 à l'article: Également appelée "tension à vide".

Note 2 à l'article: L'abréviation "OCV" est dérivée du terme anglais développé correspondant "Open Circuit Voltage".

**3.1.15**  
**impédance totale**

pertes dépendant de la fréquence dues aux effets ohmiques, d'activation, de diffusion et de concentration, à la capacité parasite et aux inductances

**3.1.16**  
**résistance totale**

partie réelle de la limite de basse fréquence de l'impédance totale

**3.1.17**  
**rapport stœchiométrique**

rapport entre le nombre de moles de gaz réactant circulant par unité de temps, et celui utilisé par la réaction électrochimique

Note 1 à l'article: Les termes "rapport stœchiométrique" et "utilisation de gaz réactant" sont liés. Le rapport stœchiométrique correspond à l'inverse de la fraction du gaz utilisé.

**3.2 Symboles**

Le Tableau 1 présente les symboles et les unités utilisés dans le présent document.

**Tableau 1 – Symboles**

Symbole	Terme	Unité
$a$	Limite d'erreur définie par la spécification de l'appareil	a
$I$	Courant	A
$J$	Densité de courant	A/cm <sup>2</sup>
$A$	Surface d'électrode active	cm <sup>2</sup>
$Z$	Impédance totale	Ω cm <sup>2</sup>
$n$	Nombre d'électrons transférés	
$N$	Nombre de cellules constituant une connexion série dans l'entité d'assemblage de cellules/piles	
$p_a$	Pression absolue du gaz d'anode	kPa
$p_c$	Pression absolue du gaz de cathode	kPa
$P$	Puissance de sortie	W
$P_d$	Densité de puissance de sortie	W/cm <sup>2</sup>
$q_a$	Débit de gaz d'anode	l/min (STP <sup>b</sup> )
$q_c$	Débit de gaz de cathode	l/min (STP)
$q_j$	Débit de composant de combustible $j$ dans le gaz d'anode	l/min (STP)
$t$	Temps	s, min, h
$T_{op}$	Température de fonctionnement de l'entité d'assemblage de cellules/piles	°C ou K
$u_c$	Incertitude-type composée pour les appareils	a

Symbole	Terme	Unité
$u_{1,i}$	Incertitude-type pour l'appareil $i$	a
$U_f$	Utilisation de combustible efficace	%
$U_{O_2}$	Utilisation d'oxygène efficace	%
$U_1$	Incertitude élargie des appareils	a
$V$	Tension	V
$x_i$	Fraction molaire du composant $i$ ou pourcentage molaire du composant $i$	mol/mol ou mol % <sup>c</sup>
$c_i$	Concentration de composant $i$	mol/m <sup>3</sup>
$\xi_j$	Taux de conversion d'hydrocarbures pour le composant d'hydrocarbure $j$	%

a Indique le point de variation de l'unité selon la spécification.  
b Abréviation correspondant aux valeurs normales de température et de pression.  
c Pour cent molaire exprimé en tant que cent fois la fraction molaire.

#### 4 Conditions générales de sécurité

Une cellule élémentaire à combustible en fonctionnement utilise des gaz oxydants et combustibles. Ces gaz sont généralement stockés dans des conteneurs haute pression. Dans certains cas, le combustible peut être un gaz condensable toxique (tel que l'ammoniac). La cellule élémentaire à combustible à proprement parler peut être utilisée à des pressions supérieures à la pression atmosphérique. Les fuites ou les écoulements en sortie de l'entité d'assemblage de cellules/piles peuvent contenir des éléments toxiques (par exemple lorsque de l'ammoniac est utilisé comme combustible). Les personnes en charge des essais sur l'entité d'assemblage de cellules/piles doivent avoir été formées et avoir acquis de l'expérience dans l'utilisation de systèmes d'essai, et plus particulièrement concernant les procédures de sécurité qui impliquent du matériel électrique et des gaz comprimés réactifs, ainsi que des composés toxiques, le cas échéant (par exemple lorsque de l'ammoniac est utilisé comme combustible).

Des matériaux compatibles avec l'emploi et le stockage des gaz réactants doivent être utilisés pendant les essais.

En résumé, une utilisation en toute sécurité du poste d'essai exige une formation et une expérience techniques appropriées, ainsi que des installations et des équipements de sécurité, autant d'éléments qui ne relèvent pas du domaine d'application du présent document.

#### 5 Entité d'assemblage de cellules/piles

Une entité d'assemblage de cellules/piles comprend une cellule ou une pile, une alimentation en gaz, des conducteurs de courant, ainsi que les autres périphériques exigés pour la production de puissance. Elle doit comporter un ou plusieurs points de mesure de la température et de la tension, ainsi qu'un ensemble de points conducteurs de courant, autant d'éléments destinés à être spécifiés par le fabricant.

Comme cela est indiqué à l'Annexe A, la limite d'une entité d'assemblage de cellules couvre l'orifice d'alimentation en gaz d'anode, l'orifice d'alimentation en gaz de cathode, le point de mesure de la température et de la pression, les points conducteurs de courant, les points de mesure de la tension, ainsi que les points d'application de charge mécanique.

Certaines entités d'assemblage de cellules/piles peuvent ne pas comporter d'orifice d'échappement du gaz d'anode ou de cathode, du fait de la configuration des cellules. En pareil cas, le circuit d'écoulement du gaz et ses éléments constitutifs doivent être déterminés par la méthode recommandée par le fabricant. La méthode d'application de charge doit être également fondée sur la recommandation du fabricant. La température maximale de fonctionnement recommandée par le fabricant ne doit pas être dépassée.

Lorsque les composants d'une entité d'assemblage de cellules/piles autres qu'une cellule/pile ne sont pas spécifiés par le fabricant, les éléments suivants doivent être décrits dans le rapport d'essai, au minimum:

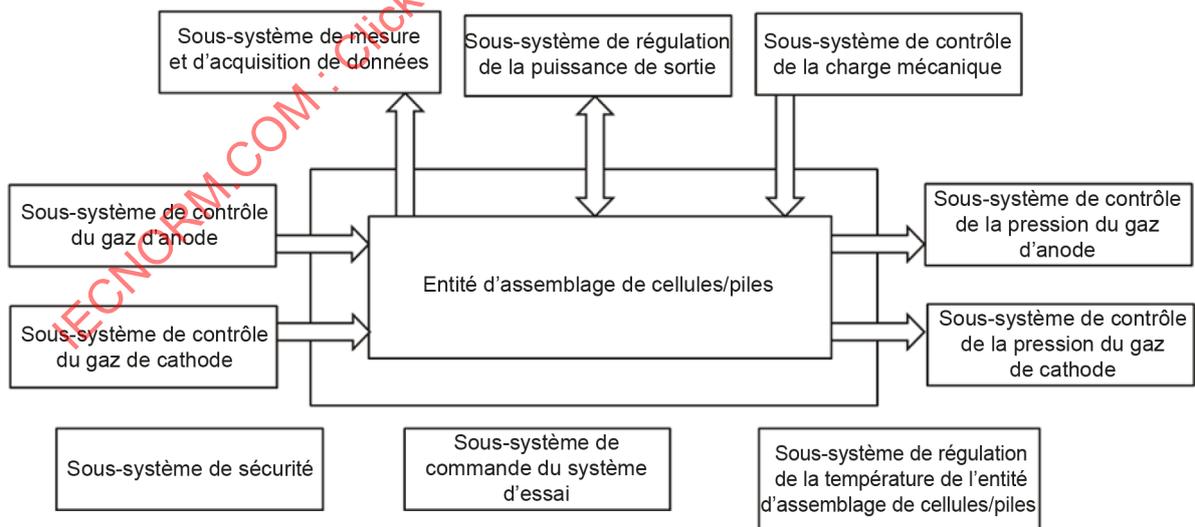
- a) matériaux et forme des composants périphériques à utiliser pour les essais;
- b) circuits et sens d'écoulement des gaz d'anode et de cathode;
- c) emplacements de mesure de température, d'application de charge mécanique, de mesure de la tension et des conducteurs de courant;
- d) importance de la charge mécanique;
- e) configuration de l'entité d'assemblage et de sa méthode d'assemblage.

## 6 Système d'essai

### 6.1 Sous-systèmes du système d'essai

#### 6.1.1 Généralités

Comme représenté à la Figure 1, un système d'essai comprend des sous-systèmes de contrôle du gaz d'anode et du gaz de cathode, un sous-système de régulation de la température de l'entité d'assemblage de cellules/piles, un sous-système de régulation de la puissance de sortie, un sous-système de mesure et d'acquisition de données, ainsi qu'un sous-système de sécurité. Il peut également inclure un sous-système de contrôle de la charge mécanique, ou bien un sous-système de contrôle de la pression des gaz d'anode et de cathode, ou un sous-système de commande du système d'essai qui commande l'ensemble du système d'essai, ou ces deux derniers sous-systèmes à la fois, si nécessaire.



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Figure 1 – Système d'essai

### 6.1.2 Sous-système de contrôle du gaz d'anode

Le sous-système de contrôle du gaz d'anode régule le débit, contrôle la composition et régule la température du gaz d'anode fourni à l'entité d'assemblage de cellules/piles. Lorsque la composition du gaz doit être maintenue sur toute la longueur de la tuyauterie, alors les matériaux, la température, le diamètre intérieur et la longueur de la tuyauterie doivent être choisis afin d'assurer que les variations éventuelles de la composition du gaz qui peuvent se produire dans la tuyauterie soient négligeables. Si nécessaire, la tuyauterie doit être chauffée ou isolée thermiquement, ou les deux à la fois, afin d'éviter toute condensation de vapeur d'eau.

Il convient de veiller à éviter l'occurrence d'autres phénomènes, tels que les dépôts de carbone, ainsi que l'évaporation et le transport d'éléments indésirables dans les flux gazeux, tels que les espèces chromées.

### 6.1.3 Sous-système de contrôle du gaz de cathode

Le sous-système de contrôle du gaz de cathode régule le débit, contrôle la composition et régule la température du gaz de cathode fourni à l'entité d'assemblage de cellules/piles.

### 6.1.4 Sous-système de régulation de la température de l'entité d'assemblage de cellules/piles

Le sous-système de régulation de la température de l'entité d'assemblage de cellules/piles régule, au moins, la température du four électrique ou de l'entité. Il maintient la température de fonctionnement. Le four électrique doit être choisi de manière à maintenir la répartition de la température dans les limites de niveau de tolérance indiquées. Il convient de s'efforcer de réduire le plus possible le bruit électrique émis par le four électrique lorsqu'il produit de la chaleur. Il est pris pour hypothèse, pour des raisons de simplicité et de sécurité, que tous les systèmes d'essai utilisent un four électrique.

### 6.1.5 Sous-système de régulation de la puissance de sortie

Le sous-système de régulation de la puissance de sortie règle le courant ou la tension de sortie de l'entité d'assemblage de cellules/piles.

### 6.1.6 Sous-système de mesure et d'acquisition de données

Le sous-système de mesure et d'acquisition de données acquiert et enregistre la température, le courant, la tension, le débit des gaz d'anode et de cathode, et de manière facultative, les conditions d'environnement (température ambiante, humidité relative et pression atmosphérique) de l'entité d'assemblage de cellules/piles, conformément à la méthode spécifiée. Si nécessaire, il acquiert et enregistre également la charge mécanique appliquée sur la cellule, la température, la composition et la pression des gaz de cathode et d'anode, le débit, la composition, la température et la pression des gaz d'anode et de cathode d'échappement, ainsi que les données d'impédance de l'entité d'assemblage de cellules/piles, etc. conformément à la méthode spécifiée.

### 6.1.7 Sous-système de sécurité

Le sous-système de sécurité fonctionne comme un détecteur et un système d'alarme de dysfonctionnement du système d'essai sur la base de paramètres et de critères prédéfinis. Lorsqu'il détecte une grave anomalie, il doit alors établir automatiquement un état sûr au sein du système d'essai. Il convient de purger l'anode avec un gaz inerte, tel que l'azote, qui peut également contenir de l'hydrogène à des concentrations inférieures à la limite inférieure d'inflammabilité.

### 6.1.8 Sous-système de contrôle de la charge mécanique

Le sous-système facultatif de contrôle de la charge mécanique régule la charge mécanique appliquée pour renforcer le contact entre les composants de l'entité d'assemblage de cellules/piles. Il convient que le sous-système soit suffisamment robuste pour appliquer la charge mécanique exigée dans les conditions d'essai et pour maintenir cette charge pendant un fonctionnement de longue durée.

### 6.1.9 Sous-système de contrôle de la pression des gaz d'anode et de cathode

Le sous-système facultatif de contrôle de la pression des gaz d'anode et de cathode régule la pression de ces gaz au moyen d'une vanne de contre-pression, etc.

### 6.1.10 Sous-système de commande du système d'essai

Le sous-système de commande du système d'essai assure la commande intégrée de chaque sous-système de contrôle/de régulation et du sous-système d'acquisition de données.

## 6.2 Variation maximale des éléments de commande du système d'essai

La variation tolérable de chaque élément de commande du système d'essai doit s'inscrire dans les plages suivantes:

dans le cas du réglage du courant: courant:  $\pm 1\%$  par rapport au point de valeur assigné;

dans le cas du réglage de la tension: tension:  $\pm 1\%$  par rapport à la valeur de consigne;

température:  $\pm 1,0\%$  par rapport à la valeur de consigne;

NOTE 1 Une variation de température par rapport à la valeur de consigne de moins de  $\pm 5$  K augmente la reproductibilité.

débit des gaz d'anode et de cathode:  $\pm 1\%$  par rapport à la valeur assignée;

composition du gaz d'anode:  $\pm 2,0$  mol % pour  $H_2$ ,  $N_2$ ;

$\pm 2,0$  mol % pour  $CO$ ,  $CO_2$ ,  $CH_4$ ,  $NH_3$ ;

$\pm 5,0$  mol % pour  $H_2O$  (concentration en vapeur d'eau);

dans le cas d'une humidification par ajoutage ou arrosage: température du point de rosée:  $\pm 1$  °C;

NOTE 2 À des concentrations en vapeur d'eau supérieures à 10 mol %, un ajuteur (arroseur) peut engendrer une plus grande incertitude.

composition du gaz de cathode:  $\pm 1,0$  mol % de la concentration cible en  $O_2$ ;

lorsque les pressions des gaz d'anode et de cathode doivent être contrôlées, pressions des gaz d'anode et de cathode:  $\pm 1\%$  de la condition assignée, lorsque la pression assignée est supérieure ou égale à 0,3 MPa; et 3 kPa, lorsque la pression assignée est inférieure à 0,3 MPa.

## 7 Appareils et méthodes de mesure

### 7.1 Généralités

Les appareils de mesure doivent satisfaire à l'exigence de 7.2. Au minimum, le débit et la composition des gaz d'anode et de cathode, ainsi que la température, la tension et le courant de l'entité d'assemblage de cellules/piles doivent être mesurés. Des mesurages supplémentaires doivent être effectués sur la base des paramètres ou des conditions d'essai, ou des deux à la fois. Il est possible que certains des éléments suivants indiqués en 7.3 ou 7.4 ne soient pas mesurables dans le cas d'une entité d'assemblage de cellules/piles qui ne comporte pas d'orifice d'échappement du gaz d'anode ou de cathode.

### 7.2 Incertitude liée aux appareils

L'incertitude élargie de chaque appareil de mesure (facteur d'élargissement  $k = 2$ ) au moment de l'étalonnage ou l'incertitude estimée à partir de la classe d'appareil doit satisfaire aux exigences suivantes:

NOTE Le facteur d'élargissement est défini dans le Guide ISO/IEC 98-3.

courant:	$\pm 1$ % par rapport à la valeur assignée;
tension:	$\pm 0,5$ % par rapport à la tension en circuit ouvert; (OCV);
température:	$\pm 1,0$ % de la valeur lue;
débites des gaz d'anode et de cathode:	$\pm 2$ % de la valeur assignée;
pressions des gaz d'anode et de cathode:	$\pm 1$ % de la valeur lue; moyenne;
composition du gaz d'anode:	$\pm 2$ mol % pour H <sub>2</sub> , H <sub>2</sub> O et N <sub>2</sub> ; $\pm 1$ mol % pour CO, CO <sub>2</sub> et CH <sub>4</sub> , NH <sub>3</sub> ;
composition du gaz de cathode:	$\pm 0,3$ mol % pour O <sub>2</sub> (équilibre N <sub>2</sub> ).

Une méthode de détermination de l'incertitude élargie des appareils est donnée à l'Annexe G.

### 7.3 Gaz d'anode

#### 7.3.1 Débit du gaz d'anode

Le débit du gaz d'anode doit être mesuré au moyen de débitmètres massiques, de débitmètres volumétriques ou de débitmètres à turbine. Le débitmètre doit être choisi en prenant en considération les espèces contenues dans le gaz fourni, la plage de débits et l'incertitude admissible du débitmètre. Lorsque les mesures sont volumétriques, elles doivent être converties en débit massique par la mesure de la température et de la pression du gaz ou de la densité de gaz au voisinage des débitmètres. Il convient d'évaluer l'incertitude de mesure pour les gaz secs conformément à l'ISO 5168 ou à l'ISO 7066-2.

#### 7.3.2 Composition du gaz d'anode

Il convient de mesurer la composition du gaz d'anode lors de la mesure des performances de l'entité d'assemblage de cellules/piles. Lorsque cette opération n'est toutefois pas possible, la composition du gaz d'anode doit être mesurée lors de la préparation à l'essai de performance dans les mêmes conditions que celles de l'essai de performance des cellules. Voir 10.6.2.2.

Lorsque le gaz d'anode est fourni dans l'une des conditions a) à d) suivantes, et si le tuyau d'alimentation en gaz ne comporte pas de réacteurs, tels qu'un reformeur, et s'il est confirmé qu'il ne modifie pas la composition du gaz de manière significative, la composition peut être calculée sur la base de la table de composition publiée par le fournisseur de combustible et des valeurs fournies par chaque débitmètre, conformément à l'ISO 6145-7:

- a) un gaz à composition unique tel que l'hydrogène est fourni;
- b) un gaz mixte de composition connue est fourni;
- c) le gaz d'anode est fourni par mélange des gaz constitutifs de manière régulée, au moyen de plusieurs débitmètres;
- d) les gaz décrits en b) et c) ci-dessus, sont fournis sous forme combinée.

Le gaz d'anode doit être échantillonné à proximité de l'orifice d'alimentation en gaz d'anode de l'entité d'assemblage de cellules/piles et analysé à l'aide d'un spectromètre infrarouge, d'un spectromètre de masse, d'un chromatographe en phase gazeuse ou d'un dispositif similaire. L'échantillon de gaz doit être acheminé de son point d'origine au point d'analyse selon une méthode qui réduit le plus possible les variations de composition. Ainsi, le matériau, la température, le diamètre et la longueur de la tubulure doivent être choisis avec soin afin de réduire le plus possible la variation de composition dans la tubulure d'échantillonnage. Si nécessaire, cette dernière doit être chauffée afin d'éviter la condensation de la vapeur d'eau.

Lorsque la vapeur d'eau est susceptible d'altérer la mesure, retirer l'eau de l'échantillon de gaz ou diluer celui-ci avec de l'argon ou un gaz inerte similaire.

Le résultat de ce type d'analyse pour le gaz constitutif  $i$ , exprimé sous la forme  $c_i$  (mol/m<sup>3</sup>), doit être normalisé afin d'obtenir une concentration normalisée,  $x_i$  (mol/mol), à l'aide de l'équation suivante:

$$x_i = \frac{c_i}{\sum_i c_i} \quad (1)$$

où  $\sum_i c_i$  représente la somme des concentrations de tous les gaz constitutifs soumis à l'analyse.

L'analyseur de gaz doit être étalonné à l'aide d'un gaz étalon de rapport massique connu.

L'incertitude de mesure doit être évaluée conformément à la série ISO 6974.

### 7.3.3 Température du gaz d'anode

La température du gaz doit être mesurée à proximité de l'orifice d'alimentation en gaz d'anode de l'entité d'assemblage de cellules/piles au moyen d'un couple thermoélectrique d'un type et d'une classe conformes à l'IEC 60584-1, ou au moyen d'un couple thermoélectrique gainé d'un type et d'une classe conformes à l'IEC 61515 et d'un conducteur d'extension d'un type et d'une classe conformes à l'IEC 60584-3. En présence d'un réacteur tel qu'un reformeur, il convient de mesurer également la température du gaz à la sortie dudit réacteur.

NOTE Il peut y avoir des différences importantes entre la température de la paroi de tube et la température du gaz en vrac.

Lorsqu'il est difficile de mesurer la température du gaz au cours de l'essai de performance des cellules, la température du gaz d'anode doit être mesurée lors de la préparation à l'essai de performance dans les mêmes conditions que celles dudit essai.

#### 7.3.4 Pression du gaz d'anode

La pression du gaz d'anode doit être mesurée en amont de l'orifice d'alimentation en gaz d'anode de l'entité d'assemblage de cellules/piles au moyen d'un capteur de pression, d'un manomètre, d'un tube de Bourdon ou d'un appareil similaire étalonné. L'appareil de mesure doit être situé de manière à réduire le plus possible l'incertitude liée à de quelconques pertes de pression dans la tuyauterie, à la température de la tuyauterie et à d'autres facteurs. Toute condensation de la vapeur d'eau pendant la mesure doit être évitée. Une méthode peut consister à mesurer la pression par injection d'une très faible quantité d'azote sec ou de gaz similaire dans la tuyauterie, à proximité de l'appareil de mesure.

#### 7.3.5 Débit du gaz d'anode d'échappement

Le débit du gaz d'anode d'échappement doit être mesuré au moyen de débitmètres massiques ou volumétriques, voire de débitmètres à turbine après mise en œuvre d'un moyen qui empêche la condensation de vapeur d'eau d'altérer la stabilité du débit de gaz d'anode, ou après retrait de l'eau du débit gazeux. Lorsque les mesures sont volumétriques, elles doivent être converties en débit massique par la mesure de la température et de la pression du gaz ou de la densité de gaz au voisinage des débitmètres. En variante, le débit du gaz d'anode d'échappement peut être calculé à partir des concentrations des constituants dudit gaz, de la concentration et du débit du gaz traceur, avec l'ajout précis d'une faible quantité de gaz, qui n'est pas contenu dans le gaz d'anode d'échappement, comme gaz traceur. L'analyseur de gaz doit être étalonné à l'aide d'un gaz étalon de rapport massique connu. L'incertitude de mesure doit être évaluée conformément à la série ISO 6974.

Le gaz d'échappement doit être manipulé avec précaution pour des raisons de sécurité et pour des raisons environnementales, étant donné qu'il peut toujours contenir de l'hydrogène, du monoxyde de carbone et des hydrocarbures.

#### 7.3.6 Constituant du gaz d'anode d'échappement

Le gaz d'anode d'échappement doit être échantillonné à proximité de l'orifice d'échappement du gaz d'anode de l'entité d'assemblage de cellules/piles. Voir 10.6.2.2. L'échantillon doit être analysé au moyen d'un spectrophotomètre infrarouge, d'un spectromètre massique, d'un chromatographe en phase gazeuse ou d'un dispositif similaire. Lorsque la vapeur d'eau est susceptible d'altérer la mesure, retirer l'eau de l'échantillon de gaz ou diluer celui-ci avec de l'argon ou un gaz similaire. Lors de la mesure, les matériaux, la température, le diamètre intérieur et la longueur de la tuyauterie doivent être choisis de façon à assurer que les variations éventuelles de la composition du gaz qui peuvent se produire dans la tuyauterie soient négligeables. En particulier, la tuyauterie doit être chauffée si nécessaire afin d'empêcher que la vapeur d'eau ne s'y condense. L'analyseur de gaz doit être étalonné à l'aide d'un gaz étalon de rapport massique connu.

#### 7.3.7 Température du gaz d'anode d'échappement

La température du gaz doit être mesurée à proximité de l'orifice d'échappement du gaz d'anode de l'entité d'assemblage de cellules/piles en choisissant un couple thermoélectrique d'un type et d'une classe conformes à l'IEC 60584-1, ou en choisissant un couple thermoélectrique gainé d'un type et d'une classe conformes à l'IEC 61515 et un conducteur d'extension d'un type et d'une classe conformes à l'IEC 60584-3. Lorsqu'il est difficile de mesurer la température du gaz au cours de l'essai de performance des cellules, la température du gaz d'anode d'échappement doit être mesurée lors de la préparation à l'essai de performance dans les mêmes conditions que celles dudit essai.

NOTE Il peut y avoir des différences importantes entre la température de la paroi de tube et la température du gaz en vrac.

### 7.3.8 Pression du gaz d'anode d'échappement

La pression du gaz d'anode d'échappement doit être mesurée en aval de l'orifice d'échappement du gaz d'anode de l'entité d'assemblage de cellules/piles au moyen d'un capteur de pression, d'un manomètre, d'un tube de Bourdon ou d'un dispositif similaire. Il convient que l'appareil de mesure soit situé de manière à réduire le plus possible l'incertitude liée à de quelconques pertes de pression dans la tuyauterie, à la température (du gaz) de la tuyauterie et à d'autres facteurs. Toute condensation de la vapeur d'eau pendant la mesure doit être évitée. Une méthode peut consister à mesurer la pression par injection d'une très faible quantité d'azote sec ou de gaz similaire dans la tuyauterie, à proximité de l'appareil de mesure.

## 7.4 Gaz de cathode

### 7.4.1 Débit du gaz de cathode

Le débit du gaz de cathode doit être mesuré au moyen de débitmètres massiques, de débitmètres volumétriques ou de débitmètres à turbine. Lorsque les mesures sont volumétriques, elles doivent être converties en débit massique par la mesure de la température et de la pression du gaz ou de la densité de gaz au voisinage des débitmètres. Le débitmètre doit être choisi en prenant en considération la plage de débits prévue et l'incertitude admissible du débitmètre. En général, l'incertitude doit être évaluée conformément à l'ISO 5168 et, en cas de non-linéarité, elle doit être évaluée conformément à l'ISO 7066-2.

### 7.4.2 Constituant du gaz de cathode

Pour la composition du gaz de cathode, la concentration en oxygène doit être mesurée au moyen d'un chromatographe en phase gazeuse ou d'un appareil de mesure de la concentration en oxygène. Il convient que le gaz de cathode consiste en de l'air comprimé ou en un gaz en bouteille propre (exempt d'huile). Lorsqu'un mélange de gaz en bouteille est utilisé, les valeurs décrites sur son certificat de composition publié par le fournisseur de combustible peuvent être utilisées. L'incertitude de l'appareil doit être évaluée conformément à la série ISO 6974.

Lorsqu'il est nécessaire de mesurer l'humidité, un appareil de mesure du point de rosée ou de la teneur en eau, ou un chromatographe en phase gazeuse doit être utilisé lors de la régulation de la température du gaz afin d'empêcher toute condensation de vapeur d'eau.

### 7.4.3 Température du gaz de cathode

La température du gaz doit être mesurée à proximité de l'orifice d'alimentation en gaz de cathode de l'entité d'assemblage de cellules/piles en choisissant un couple thermoélectrique d'un type et d'une classe conformes à l'IEC 60584-1, ou en choisissant un couple thermoélectrique gainé d'un type et d'une classe conformes à l'IEC 61515 et un conducteur d'extension d'un type et d'une classe conformes à l'IEC 60584-3.

NOTE Il peut y avoir des différences importantes entre la température de la paroi de tube et la température du gaz en vrac.

Lorsqu'il est difficile de mesurer la température du gaz au cours de l'essai de performance des cellules, la température du gaz de cathode doit être mesurée lors de la préparation à l'essai de performance dans les mêmes conditions que celles dudit essai.

### 7.4.4 Pression du gaz de cathode

La pression du gaz de cathode doit être mesurée en amont de l'orifice d'alimentation en gaz de cathode de l'entité d'assemblage de cellules/piles au moyen d'un capteur de pression, d'un manomètre, d'un tube de Bourdon ou d'un dispositif similaire. Il convient que l'appareil de mesure soit situé de manière à réduire le plus possible l'incertitude liée à de quelconques pertes de pression dans la tuyauterie, à la température de la tuyauterie et à d'autres facteurs.

#### 7.4.5 Débit du gaz de cathode d'échappement

Le débit du gaz de cathode d'échappement doit être mesuré au moyen d'un débitmètre massique, d'un débitmètre volumétrique ou d'un débitmètre à turbine après refroidissement du gaz. Lorsque les mesures sont volumétriques, elles doivent être converties en débit massique par la mesure de la température et de la pression du gaz ou de la densité de gaz au voisinage du débitmètre. Le débitmètre doit être choisi en prenant en considération la plage de débits prévue et l'incertitude admissible de l'appareil. L'incertitude de l'appareil doit être évaluée conformément à la série ISO 6974.

#### 7.4.6 Constituant du gaz de cathode d'échappement

Pour la composition du gaz de cathode d'échappement, la concentration en oxygène doit être mesurée au moyen d'un chromatographe en phase gazeuse ou d'un appareil de mesure de la concentration en oxygène après refroidissement du gaz. Lorsqu'il est nécessaire de mesurer une concentration en eau extrêmement faible, un appareil de mesure du point de rosée ou de la teneur en eau, ou un chromatographe en phase gazeuse doit être utilisé lors de la régulation de la température du gaz afin d'empêcher toute condensation de vapeur d'eau.

#### 7.4.7 Température du gaz de cathode d'échappement

La température du gaz de cathode d'échappement doit être mesurée à proximité de l'orifice d'échappement du gaz de cathode de l'entité d'assemblage de cellules/piles en choisissant un couple thermoélectrique d'un type et d'une classe conformes à l'IEC 60584-1, ou en choisissant un couple thermoélectrique gainé d'un type et d'une classe conformes à l'IEC 61515 et un conducteur d'extension d'un type et d'une classe conformes à l'IEC 60584-3. Lorsqu'il est difficile de mesurer la température du gaz au cours de l'essai de performance des cellules, la température du gaz de cathode d'échappement doit être mesurée lors de la préparation à l'essai de performance dans les mêmes conditions que celles dudit essai.

NOTE Il peut y avoir des différences importantes entre la température de la paroi de tube et la température du gaz en vrac.

#### 7.4.8 Pression du gaz de cathode d'échappement

La pression du gaz de cathode d'échappement doit être mesurée en aval de l'orifice d'échappement du gaz de cathode de l'entité d'assemblage de cellules/piles au moyen d'un capteur de pression, d'un manomètre, d'un tube de Bourdon ou d'un dispositif similaire. Il convient que l'appareil de mesure soit situé de manière à réduire le plus possible l'incertitude liée à de quelconques pertes de pression dans la tuyauterie à la température de la tuyauterie et à d'autres facteurs.

#### 7.5 Tension de sortie

Un voltmètre doit être raccordé aux points de mesure de la tension, comme décrit à l'Article 5. La tension ainsi mesurée doit être considérée comme étant la tension de la cellule/pile. Le câble de raccordement doit être suffisamment durable pour les conditions d'essai.

#### 7.6 Courant de sortie

Un galvanostat ou une charge électrique raccordés aux points conducteurs de courant, comme décrit à l'Article 5, ou un capteur de courant (ou les deux à la fois), tel qu'une résistance shunt située sur le circuit de courant, doivent être utilisés pour mesurer le courant en transmettant sa sortie à un appareil de mesure ou d'enregistrement. Le câble de raccordement choisi doit être un câble dont les matériaux et la forme sont adaptés aux conditions d'essai et à la chute de tension potentielle dans le câble.

## 7.7 Température de l'entité d'assemblage de cellules/piles

Un couple thermoélectrique d'un type et d'une classe conformes à l'IEC 60584-1, ou bien un couple thermoélectrique gainé d'un type et d'une classe conformes à l'IEC 61515 et un conducteur d'extension d'un type et d'une classe conformes à l'IEC 60584-3 doivent être choisis. Ils doivent être placés au point de mesure de la température, comme décrit à l'Article 5, et raccordés à un enregistreur ou un dispositif similaire pour la mesure. En présence d'au moins deux points de mesure de la température, la température de l'entité et sa répartition doivent être obtenues par la méthode de calcul recommandée par le fabricant.

## 7.8 Charge mécanique

Une charge mécanique appliquée selon les recommandations du fabricant doit être mesurée.

## 7.9 Impédance totale

L'impédance totale de l'entité d'assemblage de cellules/piles doit être mesurée au moyen soit de la méthode d'impédance en courant alternatif (c.a.), soit de la méthode d'interruption du courant. Une ligne de mesure appropriée doit être utilisée afin d'assurer l'existence de données de haute qualité sur toute la plage de fréquences analysées.

## 7.10 Conditions ambiantes

Pour définir les conditions ambiantes, la température ambiante, la pression et l'humidité relative doivent être mesurées. L'intervalle d'échantillonnage doit être la valeur spécifiée dans l'ISO 8756 ou une valeur inférieure.

# 8 Préparation aux essais

## 8.1 Généralités

Le type d'entité d'assemblage de cellules/piles à soumettre à essai, le nombre d'échantillons, les paramètres et les conditions d'essai doivent être déterminés.

Chaque appareil de mesure doit faire l'objet d'une vérification de son dernier étalonnage, ainsi que de l'incertitude dans les conditions d'étalonnage, ou faire l'objet d'une estimation à partir de la classe de l'appareil et de sa dépendance à l'égard des conditions d'environnement, afin d'évaluer son incertitude. La méthode et le cycle d'étalonnage et de remplacement doivent être conçus afin d'assurer que l'incertitude de mesure n'augmente pas.

Les constituants des gaz d'anode et de cathode et leurs principales impuretés doivent être vérifiés. Comme décrit à l'Article 7, un essai préliminaire doit être effectué pour déterminer la composition et la température des gaz afin d'assurer que les compositions de gaz soient établies dans les limites de l'incertitude prévue et que la température du gaz d'alimentation n'affecte pas la température de l'entité. Par ailleurs, la procédure et les conditions d'essai, ainsi que les critères d'évaluation applicables à un état stable, notamment, doivent être déterminés sur la base des résultats de l'essai préliminaire et d'autres facteurs.

## 8.2 Conditions d'essai normales et plage d'essai

Les conditions d'essai normales et la plage d'essai type qui sont recommandées par le fabricant doivent être examinées selon les paramètres suivants afin de déterminer les conditions et la plage d'essai:

- a) température de l'entité d'assemblage de cellules/piles;
- b) répartition admissible de la température de l'entité d'assemblage de cellules/piles (en présence de plusieurs points de mesure);
- c) débit du gaz d'anode;

- d) composition du gaz d'anode;
- e) pression du gaz d'anode;
- f) débit du gaz de cathode;
- g) composition du gaz de cathode;
- h) pression du gaz de cathode;
- i) utilisation de combustible efficace;
- j) utilisation d'oxygène efficace;
- k) courant ou densité de courant;
- l) tension minimale de l'entité d'assemblage de cellules/piles;
- m) courant minimal de l'entité d'assemblage de cellules/piles (avec une utilisation de combustible efficace constante, voir l'Annexe E pour de plus amples informations);
- n) courant maximal de l'entité d'assemblage de cellules/piles (avec une utilisation de combustible efficace constante). Un dommage dû à une dégradation excessive est possible au-delà de cette valeur;
- o) charge mécanique.

### 8.3 Composants et impuretés des gaz d'anode et de cathode

Lorsque des gaz sont utilisés pour préparer le gaz d'anode, le niveau de pureté ou les constituants et les principales impuretés de chaque gaz doivent être vérifiés par les tables de composition publiées par les fournisseurs de combustible respectifs ou par le biais d'une analyse. Lorsque le gaz d'anode est produit à partir d'un combustible liquide, sa densité, sa teneur en carbone, hydrogène et oxygène, ainsi que sa teneur en impuretés, telles que le soufre, doivent être vérifiées par la table de composition publiée par le fournisseur de combustible ou par le biais d'une analyse conformément à l'ISO 12185.

La pureté ou les constituants et les principales impuretés du gaz de cathode doivent être vérifiés par la table de composition publiée par le fournisseur de combustible ou par le biais d'une analyse. En cas d'utilisation d'un compresseur, l'air comprimé doit être exempt d'huile et de particules conformément à l'ISO 8573-1.

Le résultat de chaque vérification ou analyse doit être décrit dans le rapport d'essai.

### 8.4 Conditions fondamentales de la procédure d'essai

Les conditions de démarrage, telles que la vitesse d'échauffement et les conditions ambiantes au cours de la rampe d'échauffement, l'état de l'anode (c'est-à-dire le niveau de réduction de l'oxyde de nickel en nickel), et les conditions d'arrêt, telles que la vitesse de refroidissement et les conditions ambiantes au cours de la rampe de refroidissement, doivent être fondées sur celles recommandées par le fabricant ou sur les résultats des essais préliminaires.

### 8.5 Confirmation des conditions de vieillissement de l'entité

Les conditions de vieillissement de l'entité d'assemblage de cellules/piles doivent être déterminées sur la base des conditions de vieillissement recommandées par le fabricant, ainsi que sur les essais préliminaires à réaliser, afin d'assurer que la dérive de sortie au moment de la mesure soit négligeable.

### 8.6 Confirmation des critères d'état stable

Le niveau de tolérance de variation doit être déterminé pour le courant ou la tension de sortie de l'entité d'assemblage de cellules/piles, et les critères d'évaluation d'un état stable doivent être déterminés par le biais d'essais préliminaires et d'autres essais.

Les critères d'évaluation d'un état stable doivent être décrits dans le rapport d'essai.